

DUKE POWER COMPANY
PROCEDURE PREPARATION
PROCESS RECORD

(1) ID No: AP/O/A/5500/31
Change(s) 0 to
3 Incorporated

- (2) STATION: Catawba
- (3) PROCEDURE TITLE: Estimate of Failed Fuel Based on I-131 Concentration
- (4) PREPARED BY: Patricia Blessing DATE: 2/14/84
- (5) REVIEWED BY: Dennis Robinson DATE: 2/14/84
- Cross-Disciplinary Review By: _____ N/R: JMR
- (6) TEMPORARY APPROVAL (IF NECESSARY):
- By: _____ (SRO) Date: _____
- By: _____ Date: _____
- (7) APPROVED BY: JW. Gx Date: 2/14/84
- (8) MISCELLANEOUS:
- Reviewed/Approved By: _____ Date: _____
- Reviewed/Approved By: _____ Date: _____

FOR INFORMATION ONLY

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BASED ON I-131 CONCENTRATION

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A. PURPOSE

The purpose of this procedure is to provide a means of determining the quantity of failed fuel based on I-131 Concentration in the NC System.

B. SYMPTOMS

1. Any of the following EMF's in alarm:
 - a. 1EMF48 "Reactor Coolant Monitor"
 - b. 1EMF18 "Reactor Coolant Filter 1A"
 - c. 1EMF19 "Reactor Coolant Filter 1B"
2. Any condition in which the operator suspects failed fuel or want an estimate of the amount of failed fuel.

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ACTION/EXPECTED RESPONSE

RESPONSE NOT OBTAINED

C. IMMEDIATE ACTIONS

- _____ 1. Have Chemistry Obtain A Sample Of The NC System Coolant and determine temperature of sample and time sample was drawn.
- _____ 2. If, as a result of a LOCA, the containment sump level is not zero and ND pumps are taking suction from the sump (i.e. unit is in recirc mode) then request the following:
 - A) Have Chemistry obtain a sample from the ND System (i.e. sump) and determine the time the sample was drawn; and
 - B) Have Health Physics draw a containment air sample and determine the time the sample was drawn.

D. SUBSEQUENT ACTIONS

- _____ 1. Have Health Physics analyze the samples for isotopic content and concentration.
- _____ 2. Record sample concentrations and conditions on Enclosures 1 (NC Sample), Enclosure 2 (ND Sample), and Enclosure 3 (Containment Air Sample) as appropriate.
- _____ 3. If I-131 Concentration Is Greater Than 300 $\mu\text{Ci/ml}$, Refer To RP/1/A/5000/01 (EMERGENCY CLASS DETERMINATION).
- _____ 4. Calculate the density and inventory correction factors (X and Y) from Enclosure 4.

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ACTION/EXPECTED RESPONSE

RESPONSE NOT OBTAINED

- _____ 5. If the unit is in a normal operational mode and no suspected mechanically induced failure (e.g. loose part) has occurred, then go to 12.
- _____ 6. If safety injection has occurred then calculate the dilution factor Z from Enclosure 5. If this is not the case, then assume $Z = 1.0$.
- _____ 7. Compute the corrected concentrations of the analyzed isotopes by completing Enclosure 6.
- _____ 8. If cladding damage is suspected due to an abnormal transient or mechanically induced failure and:
 - A) the corrected concentrations of Xe and/or Kr exceed the limits specified in Enclosure 6; and
 - B) the corrected concentrations of I and Cs do not exceed the limits specified in Enclosure 6; and
 - C) incore thermocouples have not ever exceeded 700°F; and
 - D) RVLIS indicates no core uncover; and
 - E) no excess hydrogen is detected, then go to 14.
- _____ 9. If some fuel overheating is suspected as a result of observing several of the following conditions:
 - A) RVLIS indicates partial core uncover (<55% level) for a short period of time (<~ 30 minutes);
 - B) incore thermocouples have read >750°F;

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ACTION/EXPECTED RESPONSE

RESPONSE NOT OBTAINED

- C) Cs concentration exceeds the limit in Enclosure 6;
- D) Te and Ba concentrations do not exceed the limits in Enclosure 6; or
- E) excess hydrogen is detected, then go to 17.

____ 10. If some fuel melting is suspected as a results of observing several of the following conditions:

- A) RVLIS indicates a severe case of core uncover;
- B) incore thermocouples have read $>1100^{\circ}\text{F}$;
- C) Te and Ba concentrations exceed the limits in Enclosure 6; or
- D) excess hydrogen is detected, then go to 20.

____ 11. If none of the previous sets of conditions is met exactly (e.g. concentration vs. limit comparison is contradictory with other indications) then go to 14, 17 or 20; whichever is determined most appropriate.

____ 12. Calculate the amount of failed fuel present during normal operation using the following:

Percent Failed Fuel = Measured NC I-131 Concentration ($\mu\text{Ci}/\text{ml}$) (Enclosure 1) $\div 1.8 \mu\text{Ci}/\text{ml}\%$

_____ % = _____ $\div 1.8$

Number of Failed Pins = Percent Failed Fuel $\cdot 509.5 \text{ pins}/\%$

_____ pins = _____ $\cdot 509.5$

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ACTION/EXPECTED RESPONSE

RESPONSE NOT OBTAINED

NOTE: Iodine spiking may occur as a result of normal operational transients. This may lead to a 2 - 10 times increase in I-131 concentration while no increase in cladding damage had occurred.

_____ 13. Go To 21.

_____ 14. Calculate the amount of failed fuel due to clad damage using the following:

Percent = Corrected I-131 • Y ÷ 47.6 µCi/ml%
Failed Fuel Concentration (µCi/ml)
(Enclosure 6)

_____ % = _____ • _____ ÷ 47.6

_____ 15. If percent failed fuel exceeds 100%, then assume that some overtemperature damage has occurred and go to 17.

_____ 16. Go to 21.

_____ 17. Calculate the amount of failed fuel due to overtemperature conditions using the following:

Percent = Corrected I-131 • Y ÷ 1535 µCi/ml%
Failed Fuel Concentration (µCi/ml)
(Enclosure 6)

_____ % = _____ • _____ ÷ 1535

_____ 18. If any of the conditions listed in Step 10 are approached and the calculated overtemperature failed fuel percent is large or is >100% then suspect that some fuel melting has occurred and go to 20.

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ACTION/EXPECTED RESPONSE

RESPONSE NOT OBTAINED

____ 19. Go to 21.

____ 20. Calculate the maximum estimated
amount of failed fuel
due to melting by using the
following:

$$\begin{array}{lcl} \text{Percent} & = & \text{Corrected I-131} \cdot Y \div 2790 \mu\text{Ci/ml}\% \\ \text{Failed Fuel} & & \text{Concentration } (\mu\text{Ci/ml}) \\ & & \text{(Enclosure 6)} \end{array}$$

$$\text{_____ \%} = \text{_____} \cdot \text{_____} \div 2790$$

NOTE: Some of the I-131 is due to fuel clad damage and overtemperature conditions. The Reactor Engineer should examine other indicators of Fuel Melting (such as Te and Ba concentrations) in order to refine the estimate.

____ 21. Refer To Tech Spec 3.4.8
And Ensure I-131 Activity
Is Within Guidelines.

a. Refer to appropriate Tech
Spec Action Statement.

____ 22. Inform The Shift Supervisor
And Performance Section Duty
Engineer Of The Results Of
This Procedure. The Reactor
Engineer will review the
results in order to refine the
estimates made in this procedure.

-END-

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ENCLOSURE 1

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NC System Sample

Sample Time _____ Analysis Time _____

Sample Temperature _____ °F NC Loop Sampled _____

<u>Isotope</u>	<u>Measured Concentration (μCi/ml)</u>
----------------	--

I-131	_____
-------	-------

Kr-87	_____
-------	-------

Xe-133	_____
--------	-------

Cs-134	_____
--------	-------

Te-132	_____
--------	-------

Ba-140	_____
--------	-------

Hydrogen	_____
----------	-------

Additional:

_____	_____
-------	-------

_____	_____
-------	-------

_____	_____
-------	-------

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ENCLOSURE 2

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ND (Sump) Sample

Sample Time _____ Analysis Time _____

Sample Temperature _____ °F ND Loop Sampled _____

<u>Isotope</u>	<u>Measured Concentration (μCi/ml)</u>
----------------	--

I-131	_____
-------	-------

Kr-87	_____
-------	-------

Xe-133	_____
--------	-------

Cs-134	_____
--------	-------

Te-132	_____
--------	-------

Ba-140	_____
--------	-------

Additional:

_____	_____
-------	-------

_____	_____
-------	-------

_____	_____
-------	-------

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ENCLOSURE 3

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Containment Air Sample

Sample Time

Analysis Time

Temperature

_____ °F

Pressure

_____ psig

IsotopeMeasured Concentration ($\mu\text{Ci/ml}$)

I-131

Kr-87

Xe-133

Hydrogen

Additional:

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ENCLOSURE 4

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1. X Determination

NC SAMPLE TEMPERATURE °F

		80	90	100
NC System Temperature °F	100	.996	.998	1.00
	150	.983	.985	.987
	200	.966	.968	.970
	250	.945	.947	.949
	300	.921	.923	.924
	350	.894	.895	.897
	400	.862	.864	.865
	450	.827	.828	.830
	500	.787	.788	.790
	550	.739	.740	.741
	560	.728	.729	.731
	570	.717	.718	.719
	580	.706	.708	.708
	590	.693	.694	.695
	600	.680	.681	.683

NOTE: Assume NC Sample Temperature Is 90°F If
This Information Is Not Available.

X = _____

2. Y Determination

If power change is less than 10% within 22 days

$$Y = \frac{100}{\% \text{ F.P. at time of failure}} = \frac{100}{(\quad)} = \quad$$

Other times

$$Y = \frac{100}{OP(e^{-t\lambda}) + NP(1 - e^{-t\lambda})} = \quad$$

Where

Y = Correction Factor

OP = The % Full Power Before The Power Change

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NP = The % Full Power After The Power Change At Which Time The
Fuel Failure Has Occurred

λ = Decay Constant For I-131 --- 0.0864 day^{-1}

t = One Half The Time To Make The Power Change Plus The Time
After The Power Change Until Damage Is Suspected. (In Days)

i.e. 2 Hours For Power Change And 16 Hours To Damage.

$$t = \frac{2}{2} + 16 = 17 \text{ Hours} = .708 \text{ Days}$$

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Z (Dilution Factor) Determination

1. If the NC System has remained pressurized and no safety injection has occurred then $Z = 1.0$.
2. If safety injection has occurred but recirc from the containment sump is not occurring then perform the following:

- a) Determine the amount of injection flow:

Initial FWST level (%) prior to injection = _____ %

FWST level (%) at Sample Time = _____ %

$$\begin{array}{l} \% \text{ Level Change} = \text{Initial} - \text{Final Level} \\ \text{_____ \%} = \text{_____ \%} - \text{_____ \%} \end{array}$$
Gallons injected = $[(\% \text{ Level Change} \cdot 3.76) + 15.42] \cdot 10^3 \text{ gal.}$ (1) _____ gal. = $[(\text{_____} \cdot 3.76) + 15.42] \cdot 10^3 \text{ gal.}$

- b) Add any accumulator injection flow

If NC pressure has remained above 1240 psig:

(2) Accumulator Volume = 0 gal.

If NC pressure has dropped below 1240 psig but remained above 450 psig:

(2) Accumulator Volume = 13400 gal.

If NC pressure has not remained above 450 psig:

(2) Accumulator Volume = 43400 gal.

- c) Calculate Z:

 $Z = 1 + \text{Total injected flow} \div \text{NC Volume}$ $Z = 1 + [(1) + (2)] \div 92000 \text{ gal.}$ _____ = $1 + [\text{_____} + \text{_____}] \div 92000 \text{ gal.}$

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3. If safety injection has occurred and recirc is occurring from the containment sump then perform the following:

- a) Determine the volume of coolant in the containment sump:

Sump Level at Sample Time = _____ Ft.

If Sump Level is <10 Ft. then:

Sump Volume = [Level (Ft) • 50. + 40.] • 10³ gal.

(3) _____ gal. = [_____ Ft. • 50. + 40.] • 10³ gal.

If Sump Level is >10 Ft. refer to Curve 7.31 in the Unit One Data Book (OP/1/A/6700/01) or the appropriate curve in Unit Two Data Book (OP/2/A/6700/01) to obtain Sump Volume:

(3) _____ gal.

- b) Evaluate the ratio of concentrations:

Concentration ratio = 1.0 if ND Sample is not available.

Concentration Ratio = $\frac{\text{Measured ND I-131}}{\text{Concentration (uCi/ml)}} \div \frac{\text{Measured NC I-131}}{\text{Concentration (uCi/ml)}}$

(4) _____ = _____ ÷ _____

- c) Calculate Z:

$Z = 1.0 + \frac{\text{Sump Volume} \cdot \text{Concentration Ratio}}{\text{NC Volume}}$

$Z = 1.0 + [(3) \cdot (4)] \div 92000 \text{ gal.}$

___ = $1.0 + [\text{_____} \cdot \text{_____}] \div 92000 \text{ gal.}$

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1. Dilution corrected concentrations:

Assume dilution Factor Z is a constant for all isotopes.

$$\begin{array}{l} \text{Dilution Corrected} \\ \text{Concentration } (\mu\text{Ci/ml}) \end{array} = \begin{array}{l} \text{Measured NC} \\ \text{Concentration } (\mu\text{Ci/ml}) \\ \text{(Enclosure 1)} \end{array} \cdot Z \text{ (Enclosure 5)}$$

<u>Isotope</u>	<u>Measured NC</u> <u>Concentration ($\mu\text{Ci/ml}$)</u>	<u>Z</u>	<u>Dilution Corrected</u> <u>Concentration ($\mu\text{Ci/ml}$)</u>
I-131	_____	_____	_____
Kr-87	_____	_____	_____
Xe-133	_____	_____	_____
Cs-134	_____	_____	_____
Te-132	_____	_____	_____
Ba-140	_____	_____	_____
_____	_____	_____	_____
_____	_____	_____	_____
_____	_____	_____	_____

NOTE: Individual Z Factors may be calculated for each isotope using the method of Enclosure 5, if desired. In this manner, an improved estimate of Z may be obtained.

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2. Density corrected concentrations:

Density Corrected Concentration ($\mu\text{Ci/ml}$) = Dilution Corrected Concentration ($\mu\text{Ci/ml}$) • X (Enclosure 4)

<u>Isotope</u>	<u>Dilution Corrected Concentration ($\mu\text{Ci/ml}$)</u>	<u>•</u>	<u>X</u>	<u>=</u>	<u>Density Corrected Concentration ($\mu\text{Ci/ml}$)</u>
I-131	_____		_____		_____
Kr-87	_____		_____		_____
Xe-133	_____		_____		_____
Cs-134	_____		_____		_____
Te-132	_____		_____		_____
Ba-140	_____		_____		_____
_____	_____		_____		_____
_____	_____		_____		_____
_____	_____		_____		_____

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3. Airborne activity corrected concentrations:

Concentrations = 0.0 if no containment air sample is available.

$$\text{Volume Correction Factor} = \frac{\text{Containment Volume}}{\text{NC Volume}} = 97.$$

$$\begin{array}{l} \text{Corrected Airborne} \\ \text{Concentration } (\mu\text{Ci/ml}) \end{array} = \begin{array}{l} \text{Containment Air} \\ \text{Concentration } (\mu\text{Ci/ml}) \\ \text{(Enclosure 3)} \end{array} \cdot 97.$$

Isotope	Containment Air Concentration ($\mu\text{Ci/ml}$)	\cdot	97.	$=$	Corrected Airborne Concentration ($\mu\text{Ci/ml}$)
I-131	_____		97.		_____
Kr-87	_____		97.		_____
Xe-133	_____		97.		_____
Cs-134	0.		97.		0.
Te-132	0.		97.		0.
Ba-140	0.		97.		0.
_____	_____		97.		_____
_____	_____		97.		_____
_____	_____		97.		_____

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4. Final corrected concentrations:

The corrected concentration is the sum of the density corrected and airborne concentrations.

Isotope	Density Corrected Concentration ($\mu\text{Ci/ml}$)	+ Corrected Airborne Concentration ($\mu\text{Ci/ml}$)	= Corrected Concentration ($\mu\text{Ci/ml}$)
I-131	_____	_____	_____
Kr-87	_____	_____	_____
Xe-133	_____	_____	_____
Cs-134	_____	0.	_____
Te-132	_____	0.	_____
Ba-140	_____	0.	_____
_____	_____	_____	_____
_____	_____	_____	_____
_____	_____	_____	_____

NOTE: No correction for decay has been included. A decay correction should be factored in if elapsed time from transient to analysis exceeds 2 days. In the case of Kr-87 (76 min. half-life), apply a decay correction after 2 hours.

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5. Comparison with concentration limits:

$$\text{Concentration Limit Fraction} = \frac{\text{Corrected Concentration}}{\text{Limit Concentration}}$$

Isotope	Corrected Concentration ($\mu\text{Ci/ml}$)	\div Limit Concentration ($\mu\text{Ci/ml}$)	Concentration = Limit Fraction
I-131	_____	4700.	_____
Kr-87	_____	1.0	_____
Xe-133	_____	5.0	_____
Cs-134	_____	300.	_____
Te-132	_____	300.	_____
Ba-140	_____	1000.	_____
_____	_____	_____	_____
_____	_____	_____	_____
_____	_____	_____	_____

NOTE: Limit for Kr-87 includes provision for a 1 - 2 hour sampling and analysis delay.