

Test Plan:
Characterization of Chemical and Corrosion Effects
Potentially Occurring Inside a PWR Containment
Following a LOCA

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	July 20, 2005	July 20, 2005

Acknowledgement

This Test Plan was prepared for the Nuclear Energy Institute Sump Performance Task Force and U. S. Nuclear Regulatory Commission with input from the Westinghouse Electric Company, LLC, U. S. Nuclear Regulatory Commission and the Electric Power Research Institute (EPRI). The support of the Westinghouse Owners Group and the Babcock & Wilcox Owners Group in soliciting information, and the participation of individual plants in providing information to support the development of the Test Plan and is gratefully acknowledged.

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LIST OF ADDENDA

Revision Designation	Date	Affected Pages	Description
12	8/26/04	All	Initial release of the test plan.
12.a	10/6/04	Page 12	Amended text to correctly refer to “lithium concentration” rather than “LiOH”. Amended upper limit of lithium concentration in primary coolant to be 3.5 ppm.
		Page 14	Added requirement for a single (1) side stream filter to be installed in the loop piping.
		Page 15	Amended chemistry parameters to specify LiOH concentration to be as required to reach 0.7 ppm lithium as LiOH.
		Page 18	Added text setting limits on the flow rate and total flow to be passed through the side filter.
		Page 22	Added text describing minimum inspection requirements associated with side filter.
12.b	2/8/05	Page 16	Added text clarifying process for considering minor changes or additions to confirmatory tests.
		Page 17	Added text describing the dimensions of concrete coupons.
		Pages 17	Added text describing the preparation and addition of latent debris to test run.
		Pages 17 through 21	Amended section on test guidelines including: <ul style="list-style-type: none"> • Added text under Section 4.4.3, “Test Operation,” to note a summary of test operation guidelines is given in the following section. • Identified “Test Guidelines” as Section 4.4.3.1. • Revised presentation of the “Test Guidelines” section from bullet format to numbered format. • Added text clarifying frequency of taking test samples. • Added text describing guidance on analyzing sediment.
		Page 21	Added text describing guidance on adding TSP to test facility as Section 4.4.3.2.
12.c	3/30/05	Page 16	Swapped test sequence (3 → 4 and 4 → 3) to accommodate Industry request.

13	7/12/05	Page 10	Amended Section 4.1.31, including Table 4, “pH Levels of Sump Solutions,” to include Sodium Tetraborate as a buffering agent in the ICET program. Identified that the pH of the test would be determined by mixing a Boric Acid solution having 2800 ppm Boron with a Sodium Tetraborate solution having 2100 ppm Boron to obtain a mixture Boron concentration of 2400 ppm.
		Page 11	Added explanatory text describing that the initial test fluid will contain Sodium Tetraborate buffering agent; no attempt will be made in the test to simulate the ice melt of an ice condenser plant. (Also see page 38, amended Section 5.5.1.)
		Page 14	Corrected a typographical error in Item 14.
		Page 15	Amended “Chemistry Parameters” to include Sodium Tetraborate
		Page 16	Amended Table 5, “Test Run Conditions,” to include Sodium Tetraborate as the buffering agent for Test 5.
		Pages 19 and 20	Amended Item 4.4.3.1.5.3 to include a 4-hour spray duration at the initiation of Test 5.
		Page 21	Corrected a typographical error in Item 4.4.3.2.1.
		Page 22	Added Section 4.4.3.3, “Additional Guidance for Sodium Tetraborate Tests.”
		Page 37	Amended the last sentence of Section 5.4, “Test Temperature,” to identify 60° C (140° F) as being representative of long term pool temperatures for ice condenser containments.
		Page 38	Amended Section 5.5.1, “Spray Fluid pH,” to identify that the test will be initiated with Boric Acid/Sodium Tetraborate mixture having an initial Boron concentration of 2400 ppm.
		Page 38	Amended Section 5.5.2, “Sump Fluid pH,” to identify the pH of the simulated containment pool is derived from mixing Boric Acid and Sodium Tetraborate solutions to obtain a resulting mixture having a Boron concentration of 2400 ppm.

1 BACKGROUND

Pressurized Water Reactor (PWR) containment buildings are designed to both contain radioactive materials releases and to facilitate core cooling in the event of a Loss of Coolant Accident (LOCA). The cooling process requires water from the break and from containment spray to be collected in a sump and recirculated. The sump contains a screen that protects system structures and components in the Containment Spray and Emergency Core Cooling flow paths from the effects of debris that could be washed into the sump. There has been concern that fibrous insulation could form a mat on the screen that would obstruct flow. The flow through the fibrous mat would be further impeded if particles collect on the mat forming a dense filter cake.

Concerns have been raised about the potential for corrosion products to significantly block a fiber bed and increase its head loss. Among the materials that are found inside containment and are susceptible to corrosion and degradation by the post-LOCA solution, one can name aluminum, zinc, carbon steel, copper and non-metallic materials such as paints, thermal insulation and concrete.

A number of studies have been completed on the subject of zinc and aluminum corrosion in containment with regard to hydrogen generation. However little information is available on corrosion product release with representative post-LOCA conditions, and no studies have explored the possible interaction between the corrosion products (e.g. formation of gelatinous material, of agglomerates etc.) and the effects of those products on filtration. Further study is needed.

2 OBJECTIVE

This test plan addresses two (2) objectives:

- Determine, characterize and quantify the chemical reaction products that may develop in a representative post-LOCA containment sump environment.
- Determine and quantify any gelatinous material that develops during testing.

If gelatinous material is observed to develop in significant quantities during testing, the amount and location will be reported to the NRC and industry project managers for consideration for future debris bed head loss testing. Discovery of gelatinous material in quiescent flow zones may not be an issue as the likelihood of transport to sump screens is small. It is not intended that

this series of chemical effects tests be interrupted to conduct debris bed head loss tests if gelatinous material is found.

3 FUNDAMENTAL BASES FOR TEST PLAN DEVELOPMENT

This test plan is developed using the following as the bases for the plan:

1. The chemical corrosion codes identified by NRC (OLI Systems Inc., 2002a and 2002b) may be used, where applicable, to extend test data to plant conditions that may be beyond those explicitly covered in the test program.
2. The evaluation of the data collected will be directed at:
 - 2.1. First, determining if corrosion products form and/or if leaching of materials (from fiberglass, calcium silicate, concrete, etc.) occurs in a representative post-accident sump fluid inventory, and,
 - 2.2. If formed, characterizing and quantifying the corrosion products and leached solids to support evaluation of their impact on post accident sump head loss.
3. Epoxy-based protective coatings (paints) will not be included in the testing described in this test plan.
4. Measurement of head loss across a fiber bed is not considered as part of this test plan.
5. The test loop will be operated within a time-temperature-chemistry profile representative of PWR post-LOCA operation, except that the loop will be operated at a constant temperature of 60° C (140° F).

4 GENERAL APPROACH

This test plan addresses the following four (4) topical areas:

1. Definition of test parameters
2. Definition of the test loop
3. Test Performance
4. Characterization of test samples

4.1 Identification of Test Parameters

Tests will be conducted using justifiable proportions of non-metallic, metallic, and cementitious materials exposed to the warm, slightly basic pH liquid of the containment pool and spray environment. The specific parameters identified in the test plan are based on a review of readily available documentation and the results of surveys of U. S. nuclear power plants. The test plan logic is to conduct testing with representative material surface areas and sump volumes and chemical constituents to provide test conditions simulating the post-LOCA sump environment. Further justification for specific test parameters can be found in Section 5 to this document.

A limited number of test runs will be performed. For each test run, the parameters will be set to realistic levels that represent the conditions prevailing in containment after a LOCA. The test parameters and the bases for their selection are developed below.

4.1.1 Tested materials

The materials to be included in the test are:

- Zinc (in galvanized steel and in zinc-based protective coatings)
- Aluminum (valve actuator components, scaffolding)
- Copper (containment fan cooler fins)
- Carbon steel (untopcoated structural components)
- Concrete (representing exposed concrete surfaces and concrete dust particles)
- Insulation material (fiberglass, calcium silicate)

The amounts of each material are described below in the form of material surface areas to water volume ratios, with the exception of concrete dust, which will be represented as a mass to water volume ratio, and fiberglass and calcium silicate, which will be represented as a fiberglass or calcium silicate volume to water volume ratio. The bases for the values presented in Table 1 below are detailed in Section 5 of this document.

Table 1: Material Quantity/Sump Water Volume Ratios Planned to be Tested

<u>Material</u>	<u>Value of Ratio Tested (ratio units)</u>
Zinc in Galvanized Steel	8.0 (ft ² /ft ³)
Inorganic Zinc Primer Coatings (non-top coated)	4.6 (ft ² /ft ³) ¹
Inorganic Zinc Primer Coatings (top coated)	0.0 (ft ² /ft ³) ²
Aluminum	3.5 (ft ² /ft ³)
Copper (including Cu-Ni alloys)	6.0 (ft ² /ft ³)
Carbon Steel	0.15 (ft ² /ft ³)
Concrete(surface)	0.045 (ft ² /ft ³)
Concrete(particulate)	0.0014 (lbm/ft ³)
Insulation material ³ (fiberglass or calcium silicate)	0.137 (ft ³ /ft ³)

¹ This value addresses both untopcoated zinc-rich primer applied as an untopcoated system as well as zinc-rich primer exposed as a result of delamination of topcoat.

² Topcoated inorganic zinc coatings are protected against exposure to both containment spray and the liquid inventory of the containment pool by the topcoat. Therefore, they do not contribute to the development of corrosion products. Also, epoxy-based protective coatings provide for small quantities of leachable material, typically less than 200 ppm of the applied coating. Therefore, epoxy topcoats are judged to not contribute to the corrosion product mix post-accident and are not included in this test program.

³ Two tests are to be conducted using 100% fiberglass as the insulation material. Two additional tests are to be run with 80% calcium silicate and 20% fiberglass as the insulation material. In both cases, the same ratio of insulation material-to-sump liquid inventory will be used.

4.1.2 Physical Parameters

4.1.2.1 Simulated Sump Temperature

Previous studies have demonstrated the importance of temperature in the corrosion process of aluminum and zinc (References 1 and 2):

- Corrosion rate quickly increases with temperature, and,
- While the solubility of oxidized Al increases with temperature, the solubility of oxidized Zn decreases with temperature.

The predicted temperature history in the sump post-accident depends on the accident scenario, the operation of the plant, and the input values assumed for the calculation. A representative predicted temperature versus time profile for a large break LOCA gives the following:

- A maximum expected sump temperature of about 130° C (266° F), achieved less than 1 minute after the break.
- A cool-down of liquid temperatures in the sump such that, within 1 hour, the temperature decreases to approximately 65° C (149° F), and,
- Within 24 hours, a steady state value of 55° C (131° F) is predicted.

See Section 5.4, “Test Temperature,” for additional discussion and Figures 2 through and including 7 for representative design basis sump water temperature calculations. These calculations use input assumptions designed to maximize containment sump temperature calculations.

Performing the tests in a high temperature, high-pressure facility is not proposed. This is based on a thermodynamic simulation study conducted using Environmental Simulation Program (ESP) Version 6.6® (OLI Systems, Inc., 2002a) and StreamAnalyzer Version 1.2® (OLI Systems, Inc., 2002b) (Reference 12). The calculations performed indicated the amount of corrosion and leaching products that might be expected is dominated by the 14day low temperature phase rather than the 30 minute high temperature phase of the LOCA.⁴

⁴ Corrosion and leaching rates used in the study were based on open literature data and selected based on conservative values obtained between a pH 7 and 10 in borated water. The influence of pressure, temperature, and pH on chemical speciation was studied by speciating a fixed amount of each component based on its corrosion rate. The simulation study assumed certain corrosion and leaching rates of the metallic, concrete, and insulation materials as a function of temperature in a borated alkaline solution. An experimental study is being conducted to validate the corrosion and leaching rates used in the study.

The time of containment high-temperature operation post-accident approximately corresponds to the post-LOCA injection phase. The high temperature ($> 100^{\circ}\text{C}$) portion of the transient is over at the time recirculation from the containment sump is initiated. This is typically 20 to 30 minutes after the accident. The pH conditions during the injection phase differ from those during the recirculation phase (see section on pH below). The effect of high temperatures and a different pH on the resulting corrosion and leachant products were estimated for individual materials using the OLI systems, Inc. thermodynamic modeling suite of programs (Reference 12). The resultant calculations indicated that an elevated temperature effect is not expected to reduce the potential formation of silica-based gelatinous solids.

However, exposure of some materials to a high pH fluid which may be present during the initial spray injection for plants using NaOH buffering agent may significantly affect amount of corrosion and leachant products in the sump water, and hence affect the chemical evolution of the water. Therefore, the NaOH test will include a high pH spray injection phase to simulate this condition as described later in this document. In other words, this high pH spray injection phase will aid in the corrosion and leaching process, and potentially accelerate the formation of solid species, which may include gels, that may be formed after reaching their solubility limit.

4.1.2.2 Simulated Containment Temperature

Containment transient analysis results show that containment temperature is normally slightly below the sump temperature. Based on engineering judgment, it is estimated that the small difference in temperature between the containment vapor space and sump will have a very minor impact on the test results. Therefore the containment vapor space temperature will not be simulated independently of the sump temperature.

4.1.2.3 Simulated Sump Recirculation Flow

Velocities over samples shall be representative of post-LOCA fluid velocity conditions in PWR containment pool – which range from near zero to 3 cm/sec. A velocity profile map of the submerged portion of the test chamber will be developed so that the approximate velocity that a metallic or concrete coupon, fiberglass or calcium silicate sample is subjected to in the test can be ascertained. However, the volume of the test tank and test loop beyond the coupon samples shall be constructed so as not to allow corrosion product particulates to settle. The corrosion product particulates, if they exist, are to be collected from the grab sample line, as shown in Figure 1.

4.1.2.4 Simulated Containment Spray Flow

The ratio of spray flow to containment cross section area will be used as a simulation parameter. The value of containment spray flow and the containment cross-sectional diameter are plant specific parameters. However, a preliminary evaluation suggests that values in the table below are representative for PWR designs and are recommended for use in the test program. The suggested duration for simulating containment spray is given in Section 4.1.2.6.

Table 2: Ratio of Spray Flow to Containment Cross Sectional Area

<u>Spray Flow (ft³/Hr)</u>	<u>Containment Diameter (ft)</u>	<u>Flow/Area Ratio (ft/Hr)</u>
25,000	135	1.75

4.1.2.5 Submergence of Test Samples

The amount of material that will be submerged long term post-accident during the operation of the ECCS and CSS in the recirculation mode is a plant specific value that is dependent on the post-accident flood-up level for each plant. Based on a preliminary assessment of several representative PWRs, the split between submerged and non-submerged samples as shown in Table 3 is recommended for this test.

4.1.2.6 Test Duration

The high pH phase for NaOH spray injection testing is to last 30 minutes. The spray portion of the test will last for 4 hours including the aforementioned 30 minutes. (Note: The range of time for spray termination based on pressure control for a large dry Westinghouse 4-loop reactor is generally an hour or less. It is acknowledged that some plants continue spray operation to control dose. The 4 hours of spray operation is conservatively representative of the extended spray operation for dose control.)

Maximum duration of any test is limited to 30 days. Duration of subsequent test runs following the initial run will be determined after evaluating the results of the first run, and will consider establishing steady state conditions.

It is recommended that, metallic test specimens will be aged to allow a thin (μm thick) air oxide film to form, however, due to time constraints this may not be possible in all cases, less than

complete aging will result in samples being more susceptible to corrosion because of the absence of a protective oxide film and yield “conservative” results in this testing.

Table 3: Percentage of Surface Areas Above and Below Containment Flood Levels

<u>Material</u>	<u>Submerged</u> <u>%</u>	<u>Non-Submerged</u> <u>%</u>	<u>Comment</u>
Zinc Galvanizing	5	95	<ul style="list-style-type: none"> The submerged value accounts for grating and duct work that might be submerged.
Zinc Coatings (topcoated)	0	0	<ul style="list-style-type: none"> Epoxy-based topcoats preclude interaction of the zinc primer with containment sump inventory and containment spray. Exposure of zinc primer to containment sump and containment spray fluids due to local failures of epoxy-based topcoats is accounted for in the untopcoated zinc coatings.
Zinc Coatings (untopcoated)	4	96	<ul style="list-style-type: none"> Addresses both untopcoated zinc primer applied as an untopcoated system as well as zinc primer exposed as a result of delamination of topcoat.
Aluminum	5	95	<ul style="list-style-type: none"> Aluminum is generally not located at elevations inside containment where it may be submerged.
Copper	25	75	<ul style="list-style-type: none"> Majority of surface from CRDM coolers and instrument air lines.
90-10 Cu/Ni	25	75	<ul style="list-style-type: none"> Majority of surface present in containment fan coolers.
Concrete	34	66	<ul style="list-style-type: none"> The submerged value accounts for limited damage to floor and wall surface areas that will be submerged due to primary RCS piping being elevated above the containment floor.
Carbon Steel	34	66	
Fiberglass	75	25	<ul style="list-style-type: none"> The submerged value accounts for most of the fiberglass to remain in areas where it will wash down into the sump pool.
Calcium silicate	75	25	<ul style="list-style-type: none"> The submerged value accounts for most of the cal-sil to remain in the areas where it will wash down into the sump pool.

4.1.3 Chemical Parameters

4.1.3.1 pH

The pH of the sump solution and containment spray solutions will have a large effect on corrosion and precipitation reactions. Hydrated trisodium phosphate (TSP) ($\text{Na}_3\text{PO}_4 \cdot 12\text{H}_2\text{O}$) and sodium hydroxide (NaOH) are the standard chemicals used for pH control in post-LOCA solutions. For plants using TSP, the TSP is stored in baskets on the containment floor and is dissolved by the post-LOCA solution within a certain time. For plants using sodium hydroxide, the NaOH is injected and mixed directly with the containment spray flow. During the approximately 30 minute initial NaOH injection into the spray stream, pH values in this stream can be as high as 12. Typical values of sump solution pH are shown in Table 4 below⁵.

Table 4: pH Levels of Sump Solutions

<u>Sodium Hydroxide (NaOH)</u>	<u>Trisodium Phosphate</u>	<u>Sodium Tetraborate</u>
10	7	T. B. D. (See explanation below.)

For tests using NaOH as a buffering agent, the maximum pH of the recirculation solution is 10 when pH adjustment is made with sodium hydroxide and the Boron concentration in the RWST is low (2300 ppm). When TSP is used, its required quantity is calculated to reach a minimum pH of 7.0 at high boric acid concentration in the RWST. For the test using Sodium Tetraborate, the pH is a consequence of achieving a target Boron concentration as described below.

Therefore three sets of pH conditions will be simulated:

- pH = 10, with NaOH.
- pH = 7.0, with TSP.
- pH = Initial pH will be determined by combining and mixing appropriate quantities of Boric Acid solution having a Boron concentration of 2800 ppm and Sodium Tetraborate solution having a concentration of 2100 ppm Boron to produce an initial test solution having a Boron concentration of 2400 ppm.

⁵ pH of a post-LOCA containment pool is initially determined by the pool constituents. For ice condenser plants, this includes the ice-sodium tetraborate solution (2100 ppm Boron), the reactor coolant system (variable, depending on time in core life), and the ECCS/RWST (boric acid) Boron concentrations. An initial pH of ~ 8.1-8.5 in the test

No adjustment of the initial pH will be made after the commencement of the test.

At a pH range of 6.5 to 7.0, corrosion of aluminum is minimal. It is known that corrosion of aluminum increases with increasing pH (References 1 and 2).

Similarly, zinc corrosion reaches a minimum at a pH range of 7.0 to 8.0 and increases outside that range (References 2, 7 and 8). It is noted that Reference (2) shows this minimum corrosion to be in the pH range of 8.5 to 9.5.

It should also be noted that the solubility of aluminum oxides increases with pH while the solubility of zinc oxides decreases with pH (Reference 1).

pH conditions during the injection phase are different from those during the recirculation phase.

- For plants with TSP as pH control agent, the minimum pH during injection (pH ranging from 4.7 to 5) corresponds to the boric acid concentration in the RWST (2000-2800 ppm boron). It is noted that sump pH will not reach a steady state until the TSP is completely dissolved (one to two hours). For the pre-conditioning of test coupons, a conservatively low value of 4.7 is selected corresponding to a concentration of 2800 ppm boron in the RWST.
- For plants with NaOH addition, the spray pH can be higher – up to 12 – during the injection phase, when sodium hydroxide is mixed with the spray water.
- Plants that use sodium tetraborate as a pH control agent are ice condenser plants. For these plants, the minimum pH during injection (pH ranging from about 4.7 to 5) corresponds to the boric acid concentration of the RWST (2000-2800 ppm boron). Sodium tetraborate is added to the sump liquid inventory as the ice condenser ice bed melts. Thus, the sump pH will not reach its final value until the ice bed is completely melted (generally within about one hour after initiation of the large break LOCA).

For tests using NaOH as a buffering agent, the short operating period in plants with a high pH value (about 30 minutes of containment spray) will be simulated. For the tests using TSP as the buffering agent, a gradual introduction of the buffering agent over an hour will be simulated. For the test using Sodium Tetraborate as a buffering agent, the test will be initiated with a solution

loop is expected when the Sodium Tetraborate and Boric Acid solutions are mixed, not considering other effects such as CO₂ ingestion or reactions with the sample materials.

representing the boron concentration at the completion of ice bed melt. That is, no attempt will be made to simulate the addition of Sodium Tetraborate due to the phenomena of ice melting. See Section 5.5.1 for additional discussion on this approach.

Hydrochloric acid (HCl) can be formed from the degradation of cable insulation material (Reference 5). The test will provide for the conservative treatment of this degradation by having an initial concentration of 100 ppm of HCl in the fluid simulating the sump inventory.

4.1.3.2 Aeration

Dissolved oxygen is known to accelerate corrosion. Therefore, all tests will be conducted in fully aerated (e.g., air saturated) conditions. It is also recognized that some reduction of pH may result from entrainment of CO₂ in the fluid. No attempt will be made to increase or maintain pH beyond the initial value of ~10 for NaOH injection tests to compensate for this effect.

4.1.3.3 Other

Several chemical species will be set at the same initial value from test to test. Specifically, this pertains to boric acid, lithium hydroxide and pre-existing surface corrosion.

- For boric acid, a maximum boron concentration of 2800 ppm is selected. This value is chosen in recognition of the current trend to increase boron concentration in the RWST as core designs move to more reactive cores. (It is noted that some plants in the US already operate with 2900 ppm boron in the RWST). The initial boron concentration of 2800 ppm will be used during all test runs as this value is set largely by the RWST tank boron concentration and does not vary significantly from plant to plant.
- The lithium cation will affect zinc and aluminum corrosion primarily through an indirect pH effect. Lithium concentration typically varies between 0 and 3.5 ppm in the RCS, so its concentration would be less than 0.7 ppm in the post-LOCA recirculation solution. The impact is obviously negligible when compared to the NaOH concentration (more than 2500 ppm). However, to preclude the possibility of this assumption being challenged after the testing has been completed, this minimal concentration (0.7 ppm Li as LiOH) will be incorporated into the Test Plan for completeness.

4.2 Applicable ASTM Standards and Standard Practices

The following ASTM Standards and Standard Practices should be used, as applicable, in conjunction with the specific instructions offered below.

- G 1-90 (1999), Practice for Preparing, Cleaning, and Evaluating Corrosion Test Specimens
- G 4-01, Guide for Conducting Corrosion Coupon Tests in Field Applications
- D 3370-95a (1999), Standard Practices for Sampling Water from Closed Conduits
- G 16-93 (1999), Guide for Applying Statistics to Analysis of Corrosion Data
- G 31-72 (1999), Standard Practice for Laboratory Immersion Corrosion Testing of Metals

To minimize the number of individual coupons used in testing, the coupon sizing may depart from those specified in the procedures above. This is considered acceptable as determination of corrosion rate data is considered to be of secondary importance to the stated test objectives.

4.3 Test Loop Functional Requirements

The functional requirements for the test loop are described in this section. A schematic of a suitable test loop is shown in Figure 1

1. The central component of the system is a test tank. The test tank shall be designed to preclude the formation of sedimentation in the test tank.
2. The test tank shall be capable of maintaining both a liquid and vapor environment as would be expected in containment post-LOCA.
3. The test loop shall be capable of temperature control of the liquid phase to within $\pm 5^{\circ}$ F.
4. The system shall be capable of circulating water at flow rates that simulate spray flow rate per unit area of containment cross section. Pump required flow and head will be determined later.
5. The test tank shall provide for water flow over submerged test coupons that will simulate the range of sump fluid velocities that may be related to conditions expected at plants.
6. Piping and related isolation valves are to be provided such that the parallel stream can be isolated during performance of the test.

7. The pump discharge line shall split in two: one branch being directed to the spray ramp located in the vapor space inside the reaction tank, the other branch returning to the liquid side of the tank. Each branch will be provided with an isolation valve and flow meter.
8. A flow meter shall be provided in the recirculating piping.
9. The pump circulation flow rate shall be controlled at the pump discharge to be within ± 5 per cent of the flow required to simulate fluid velocities in the test article. Flow control may be either automated or manual. The ability to manually control flow at the levels identified for testing is to be demonstrated prior to initiating testing.
10. The tank shall accommodate a rack of immersed sample coupons including the potential reaction constituents identified previously.
11. The tank shall also accommodate a rack of sample coupons that may be exposed to spray of liquid that simulates the chemistry of a containment spray system. Provision is to be made for visual inspection of the spray rack.
12. The tank shall provide for sufficient space between the test coupons as to preclude galvanic interactions⁶ among the coupons. As a minimum, different metallic test coupons shall be electrically isolated from each other and the test stand to prevent galvanic effects resulting from metal-to-metal contact between specimens or between the test tank and the specimens as outlined in procedures ASTM G4 and G31.
13. The fluid volumes and sample surface areas shall be based on scaling considerations to relate the test conditions to plants.
14. A cartridge or analytical filter shall be provided for in a side stream from the loop piping. The cartridge or analytical filter shall be capable of retaining particulates having a dimension of 0.45 microns or larger. The cartridge or analytical filter shall be constructed of material that will not chemically interact with the test fluid.
15. All components of the test loop shall be made of corrosion resistant material (for example, stainless steel for metallic components).

⁶ Galvanic reactions are local corrosion effects, occurring between two or more electrically coupled dissimilar metals with an electrolyte path between the metals. Due to the small scale of the test, the potential influence of the tank and the potential variable effects from possible metallic coupon arrangements, there is a concern that electrically coupling coupons could yield non-typical results. To preclude this occurrence, test coupons are to be electrically isolated from one another.

4.4 Test Performance: Program and Guidelines

The practices that govern the testing outlined below are given in ASTM G4 and G31.

4.4.1 Test Program

Based on the evaluation of parameters identified in Table 3, the liquid volume of the test facility used to perform the test will fix the specific values for each of the materials and the following test parameters:

Physical parameters:

- Water volume in the test tank: 949 l (250 gal.)
- Circulation flow: 0 – 200 l/min (0 – 50 gpm)
- Spray flow: 0 – 100 l/min (0 – 25 gpm)
- Sump temperature: 60° C (140° F)

Chemistry parameters:

- Boron concentration: 2800 ppm
- Na₃PO₄·12H₂O concentration: as required to reach pH 7 approximately 2 g/l in the simulated sump fluid
- NaOH concentration: as required to reach pH 10 approximately 6g/l in the simulated sump fluid
- Sodium tetraborate concentration: as required to reach a Boron concentration of 2400 ppm approximately 15 g/l in the simulated sump fluid
- HCl concentration: 100 mg/l
- LiOH concentration: as required to reach 0.7 ppm lithium as lithium hydroxide

The test program defined by this test plan includes five (5) test runs. The parameters of each run are described in Table 5.

An objective of Test 5 is to provide information regarding the chemical effects with a Sodium Tetraborate buffer, including evaluating resulting products of chemical reactions. The initial pH for Test 5 is expected to be between that achieved for Test 1, which used Sodium Hydroxide as a buffering agent, and for Test 2, which used Trisodium Phosphate as a buffering agent. Other test parameters for Test 5 (fluid temp, number and type of coupons, and insulation tested) are similar to those of Test 1 and Test 2.

Table 5: Test Run Conditions

<u>Run</u>	<u>Temp</u> (° C)	<u>Buffering</u> <u>Agent</u>	<u>pH</u>	<u>Boron</u> (ppm)	<u>Note</u>
1	60	NaOH	10	2800	100% Fiberglass insulation test. High pH, NaOH concentration as required by pH. (See Notes 1 and 2)
2	60	TSP (Na ₃ PO ₄ ·12H ₂ O)	7	2800	100% Fiberglass insulation test. Low pH, Trisodium Phosphate concentration as required by pH.
3	60	TSP (Na ₃ PO ₄ ·12H ₂ O)	7	2800	80% Calcium silicate / 20% fiberglass insulation test. Low pH, Trisodium Phosphate concentration as required by pH.
4	60	NaOH	10	2800	80% Calcium silicate / 20% fiberglass insulation test. High pH, NaOH concentration as required by pH. (See Note 2)
5	60	Sodium Tetraborate	8.0 to 8.5 (est.)	2400	100% fiberglass insulation test with sodium tetraborate buffer. The sump solution is to be targeted to simulate the solution mixture of a representative ice condenser plant after melting of the ice beds. (Note 3).

Notes:

- (1) The duration of Test 1 will be 30 days.
- (2) During the first 30 minutes of Tests 1 and 4, NaOH will be injected in the spray fluid. The quantity of NaOH injected in the spray solution is subject to the following constraints:
 - a. The pH of the spray fluid shall not exceed a value of pH = 12 during this initial 30 minute injection phase, and,
 - b. The target pH of the simulated sump fluid inventory at the termination of the of containment spray simulation (e.g., after the 30 minute NaOH injection phase), not considering pH effects due to CO₂ absorption and other chemical effects which may be occurring during NaOH injection, is a value of pH = 10.
- (3) For Test 5, the planned test loop solution is to be a mixture of Boric Acid solution having a Boron concentration of 2800 ppm and Sodium Tetraborate solution having a concentration of 2100 ppm to achieve a resulting Boron concentration of 2400 ppm. To the extent practicable the remaining run parameters will be consistent with those of Test 1 and Test 2. This includes the temperature of the test solution, number and types of coupons, and insulation samples. Spray will be initiated at the commencement of the test and continue for 4 hours, but no additional buffering chemical will be added to the spray stream.

4.4.2 Test Coupon Preparation

The following guidance is given for the preparation of coupons to be used in the tests identified in this test plan.

4.4.2.1 Metallic Coupons

Test coupons should be prepared using the recommended practices outlined in ASTM G1, as applicable. This practice describes accepted procedures for and factors that influence laboratory immersion corrosion tests, particularly mass loss tests. These factors include specimen preparation, apparatus, test conditions, methods of cleaning specimens, evaluation of results, and calculation and reporting of corrosion rates. This practice also emphasizes the importance of recording all pertinent data and provides a checklist for reporting test data. As noted previously, to minimize the number of individual coupons used in testing, the coupon sizing may depart from those specified in the procedures above. This is considered acceptable practice as determination of corrosion rate data is considered to be of secondary importance to other stated test objectives. However, exceptions and deviations from ASTM G1 are to be documented and justified by the test performer.

4.4.2.2 Fiberglass and Calcium Silicate

For fiberglass and calcium silicate, retention in multiple sample baskets fabricated from fine (~1/16-inch stainless steel mesh) screens with small, removable sample containers, is recommended. Most of the submerged fiberglass and calcium silicate should be inserted below the metallic coupons where flow can pass through it. Some of this material should be placed in areas of the test tank that are expected to be quiescent as well.

4.4.2.3 Concrete Samples

Concrete coupons are to be located in both the submerged and non-submerged region. To facilitate placement of the 12" x 12" x 2" concrete coupon only one coupon will be placed in the submerged rack. The area of this coupon will exceed the surface area requirement for submerged concrete (.51 ft²) and non-submerged concrete (.99 ft²). Since it is all submerged, it is conservatively exposed to the sump fluid for the entire duration of the test run versus the 4 hour spray interval.

4.4.2.4 Aging

It is recommended that, metallic test specimens will be aged in air under ambient conditions to allow a thin (μm thick) air oxide film to form. Due to time constraints this may not be possible in all cases. However, less than complete aging will result in samples being more susceptible to corrosion because of the absence of a protective oxide film. This will allow for greater corrosion of the test coupons and samples and yield “conservative” results in this testing.

4.4.2.5 Latent Debris

Particulate material simulating 200 lb_m of latent containment debris is to be added to the fluid in the test tank as follows:

- Concrete particulate [simulating 50 lb_m]: 21.2 grams
(0.75 ounces)
- Surrogate particulate debris (sand and clay) [simulating 150 lb_m]: 63.7 grams
(2.25 ounces)

The basis for the above amounts of debris identified above is as follows:

- The test plan assumed 50 lbs of concrete particulate debris in a representative containment, and provides a factor of 0.0014 lbs (test material) / ft³ of system volume.
- The mass of the concrete particulate sample is then calculated as
$$0.0014 \times 250 \text{ gallons} / 7.48 \text{ gal/ft}^3 = 0.0467 \text{ lbs} = 0.75 \text{ oz} = 21.2 \text{ gms}$$
- Assuming a total latent particulate loading in containment of 200 lbs, including the concrete particulate, the mass of the surrogate particulate would then correspond to 150 lbs or three times that for the concrete particulate or 2.25 oz or 63.7 gms.

The particulate material should be shaken into the test tank after all of the chemical additions have been completed and the tank temperature has reached 60 °C, but before insertion of the fiberglass and sample coupons.

Note: The concrete particulate matter is to be obtained by chipping off one corner of one of the concrete test coupons and grinding it into a fine powder, about the size of the surrogate particulate debris. Chipping and grinding shall be performed in a manner so as not to contaminate the concrete with extraneous material such as iron oxide or glass.

4.4.3 Test Operation

The general practices for testing given in ASTM G4 and G31 should be used, as applicable, to perform the test. Detailed test procedures or instructions shall be developed. Such procedures and instructions shall be approved by the NRC and EPRI project managers/test program leads prior to use. A summary of test operation guidelines are given below:

4.4.3.1 Test Guidelines

The following general sequence of events is to be followed for the testing:

- 4.4.3.1.1 The cleanliness of the test loop shall be verified. Between test runs the loop shall be cleaned in accordance with applicable portions of ASTM A 380-99, Standard Practice for Cleaning, Descaling, and Passivation of Stainless Steel Parts, Equipment, and Systems. No observable scale or sediment shall be present in the test tank or loop piping, and the loop water conductivity (after cleaning) shall be $< 50 \mu\text{S}/\text{cm}$ and turbidity shall be $< 0.3\text{NTU}$.
- 4.4.3.1.2 The test loop shall be filled with demineralized water and chemicals added and adjusted as required for the initial test run conditions.
- 4.4.3.1.3 The temperature of the system shall be adjusted to the normal test run operating temperature.
- 4.4.3.1.4 The sample coupons and other test material (insulation, debris) shall be placed into the test sample racks and inserted into the test loop in a preplanned manner consistent with scaling and other pertinent information.
- 4.4.3.1.5 Operation of the test loop shall be performed with conditions representative of postulated post-LOCA conditions, including:
 - 4.4.3.1.5.1 A temperature of 60°C (140°F), $\pm 2^\circ\text{C}$
 - 4.4.3.1.5.2 Fluid velocities over samples between 0 and 3 cm/sec (0.1ft./sec), except where specifically exempted (e.g., fiberglass samples near the drain line will be subjected to significantly higher velocities).
 - 4.4.3.1.5.3 For all tests (Tests 1, 2, 3, 4 and 5), spray duration is 4 hours.

For tests using NaOH as a buffering agent, the initial high pH phase of the spray simulation will last 30 minutes. This high pH phase is intended to simulate containment spray with NaOH addition during the initial injection phase of a LOCA sequence. The remaining $3 \frac{1}{2}$ hours simulate spray on recirculation.

For the Sodium Tetraborate test, the containment spray duration will last 4 hours. The test will be initiated with a solution having a Boron concentration of 2400 ppm that is achieved from mixing a Boric Acid solution having a Boron concentration of 2800 ppm and a Sodium Tetraborate solution having a Boron concentration

of 2100 ppm. Spray will be conducted with recirculated sump fluid with no additional chemicals added to the spray stream

- 4.4.3.1.5.4 Nominal duration of each test run is 30 days, however, this may be shortened if chemical equilibrium is observed. Any reduction in duration shall be approved by the project managers/test program leads.
- 4.4.3.1.6 Sampling shall be performed at frequencies necessary to obtain necessary information regarding the behavior of the test loop and for characterization of chemical reaction products which may occur:
 - 4.4.3.1.6.1 For all grab samples, the observable physical properties of the sample shall be noted as soon as possible after the sample is taken from the test loop. The properties include color, suspended solids, and kinematic viscosity.
 - 4.4.3.1.6.2 Samples shall be obtained at appropriate times and frequencies during each test run, and particularly before and after significant changes are made in loop operations and frequently when parameters are expected to be changing rapidly. For example, samples should be obtained before adding debris, after adding debris, before initiating spray, after terminating a spray phase and frequently during the first day or so of the testing.
 - 4.4.3.1.6.3 Characterization of particulate matter is particularly important. Where feasible, determination of the chemical constituency and compounds, size, density, specific surface area, and information relative to the microstructure of the material (crystalline or amorphous) shall be performed. Collection of samples for particulate characterization shall be performed at least following debris addition, 24 hours after test commencement and at weekly intervals thereafter unless sample analysis indicates that particulate inventory is stable at a low value.
 - 4.4.3.1.6.4 If potentially gelatinous material is observed or identified, the project manager/test program leads shall be informed promptly with a description of the material and other pertinent information.
 - 4.4.3.1.6.5 Analyses of fluids shall be performed to characterize dissolved material in the test loop and behavior of loop chemistry. For elements whose concentration is not expected to vary during the test such as B, Li, K, and Pb analyses shall be performed at the beginning, end, and mid-point of a test run. For elements whose concentration may vary during a test run, frequent analyses shall be performed. These elements include Al, Ca, Cu, Fe, Ni, Si, Na, and Zn. Collection of samples for these analyses shall be performed at least before coupon and insulation addition, after 30

minutes, after spray termination, at 2,4,8, and 24 hours, and at a nominal daily frequency thereafter. Sampling and analysis frequencies may be changed if a sufficient basis exists, subject to project manager/test program lead approval.

4.4.3.1.6.6 Post-run examinations include sediment characterization and examination of, weighing, photographing and storage of sample coupons and insulation samples. Except for samples undergoing detailed and possibly destructive examination, adherent deposits and/or corrosion products shall not be removed. Coupons and samples shall also be preserved in appropriate sealed storage containers for possible future examinations. Ultimate disposition of samples and material generated during testing shall be governed by instructions jointly issued by the NRC and EPRI project managers/test program leads.

4.4.3.1.6.7 Sediment characterization includes, as appropriate and feasible determining the mass and volume collected, determination of constituents (e.g., fiberglass, latent particulate, precipitate, etc.), determination of density and specific surface area, whether amorphous or crystalline, elemental composition and speciation. Specific direction concerning characterization shall be provided jointly by the NRC and EPRI project managers/test program leads.

4.4.3.1.7 The test matrix calls for four primary test runs and a confirmatory run. It is planned that one of the test runs will be repeated as a confirmatory test. However, minor changes or additions to the previous test run to be repeated may be made for the confirmatory test. The project managers/test program leads shall concur regarding which run will be repeated and any changes or additions to be made.

4.4.3.2 Additional Guidance for TSP Tests

The following clarify the requirements for TSP addition during appropriate test runs:

4.4.3.2.1 Benchtop analyses shall be performed to confirm the quantities, concentrations and pH values to be achieved with TSP buffer addition.

4.4.3.2.2 The initial 30-minute spray phase will not contain TSP, as this material will not be introduced into spray before commencing recirculation operations in a postulated post-LOCA sequence. TSP will dissolve gradually, and metered addition of TSP solution into the test loop during the remaining 3 ½ hour spray phase is appropriate.

4.4.3.2.3 HCl should be added toward the end of the TSP injection sequence, during the final two hours of spray operation. Adding HCl early in the sequence or before buffer addition would result in unrealistically low pH values in the test loop fluid.

4.4.3.3 Additional Guidance for Sodium Tetraborate Tests

The following clarify the requirements for sodium tetraborate addition during Test 5:

- 4.4.3.3.1 The target Boron concentration of the sodium tetraborate solution simulating the melting ice of an ice condenser is to be about 2100 ppm. During preparation and holding, the solution is to be maintained in constant recirculation to keep the borax in solution⁷.
- 4.4.3.3.2 The target initial Boron concentration of the solution mixture in the test loop inventory is to be 2400 ppm. This concentration is to be achieved by mixing a Boric Acid solution having a Boron concentration of 2800 ppm with a Sodium Tetraborate solution having a Boron concentration of 2100 ppm.
- 4.4.3.3.3 Benchtop analyses shall be performed to confirm the quantities, concentrations and pH values to be achieved by mixing a Boric Acid solution with a Boron concentration of 2800 ppm and a Sodium Tetraborate solution with a Boron concentration of 2100 ppm to obtain a mixture having a Boron concentration of 2400 ppm.
- 4.4.3.3.4 Containment spray will be simulated for the first 4 hours of the test. This is conservative for ice condenser plants as there is little material in the ice condenser upper containment that containment spray solution can chemically interact with.
- 4.4.3.3.5 HCl is to be added during the final two hours of spray operation. This addition time is representative of the earliest time that HCl would be generated in the plant.

4.4.4 Test Termination Criteria

Based on the evaluation of the particulates captured on filtered grab samples, a decision will be made to continue or terminate the test. The criteria for termination are based on the following:

1. The first test will run for 30 days.
2. Maximum duration of any test is limited to 30 days. Duration of subsequent test runs following the initial run will be determined after evaluating the results of the first run, and will consider establishing steady state conditions.

⁷ Achieving a super saturated solution of Sodium Tetraborate is not desired as it would be completely non-representative of the ice condenser chemistry. Moreover, this could produce precipitates or reaction products that would be test run artifacts not representative of a representative plant situation.

3. For subsequent tests, the following termination criteria will be used:
 - a. The corrosion process achieves an equilibrium or steady-state condition in less than 30 days,
 - b. The leaching of silica, should it occur, reaches a steady-state condition in less than 30 days,
(Note: for criteria [a] and [b], the test sponsors will be consulted prior to termination of the test to determine whether the test should indeed be terminated.)
 - c. The test duration reaches 30 days of continuous operation, or
 - d. Alternate termination criteria, which will be discussed with the test sponsors prior to their implementation.

4.5 EVALUATION OF TEST SAMPLES

In general, ASTM G16-93 (1999) may be used to evaluate the corrosion data. Specific evaluation requirements are listed below.

4.5.1 Test Coupon Evaluation

Coupons used in the test should be weighed and photographed before and after testing. Prior to weighing, the coupons shall be dried to remove moisture from the attached corrosion products. These records and the coupons are to be retained for later use.

4.5.2 Fiberglass and Calcium Silicate Sample Evaluation

Similarly, the fiberglass and calcium silicate samples use in the test should be weighed and photographed before and after testing. Specific attention is to be given to possible collection of gelatinous material on the surface of the fiberglass or calcium silicate. Prior to weighing, the coupons shall be dried to remove moisture. These records and the fiberglass or calcium silicate samples are to be retained for later use.

4.5.3 Sampling and Grab Sample Analyses.

Given below are specific guidelines for collecting grab samples. See ASTM Standard D 3370-95a (1999), "Standard Practices for Sampling Water from Closed Conduits," for additional general guidance on collecting grab samples.

In preparation for collecting a grab sample, the sample line shall be flushed with a minimum volume equivalent to three sample line lengths. To assure a representative sample, the flow rate used during line flushing and sampling shall be sufficient to assure flow is sufficient to maintain all species in suspension in the sample line. Excess solution removed from the loop during sampling shall be collected and returned to the loop.

The volume of sample to be removed shall be sufficient to measure pH, turbidity and boron concentration on the unfiltered sample and to allow for filtration of an appropriate volume of solution through a 0.45 micron filter for collection of suspended material. A small amount of the grab sample fluid will be set aside for other analyses, and the remainder will be re-introduced to the test loop.

The sample for filtration should be filtered as rapidly as possible after collection to assure that precipitation of any material has not resulted from a decrease in sample temperature. The filtered material should be dried and weighed until reproducible results are obtained. The filtrate shall be collected and stored at ambient temperature in a sealed container and observed for at least 1 week to determine if precipitates form upon standing.

A sample should be evaluated as rapidly as possible after collection for the existence of gelatinous material in the grab sample fluid. One possible method of evaluation is to measure the viscosity of the sampled fluid.

At a later time, a representative portion of the filtered material shall be digested for the purpose of elemental composition. The filtrate shall be re-filtered using a 0.45 micron filter to remove for determination of elemental constituents any material that has settled from solution. The acidified filtrate shall also be analyzed to determine elemental constituents.

As a minimum, the filtered and non-filtered grab sample species shall be analyzed for the chemical species containing the following elements:

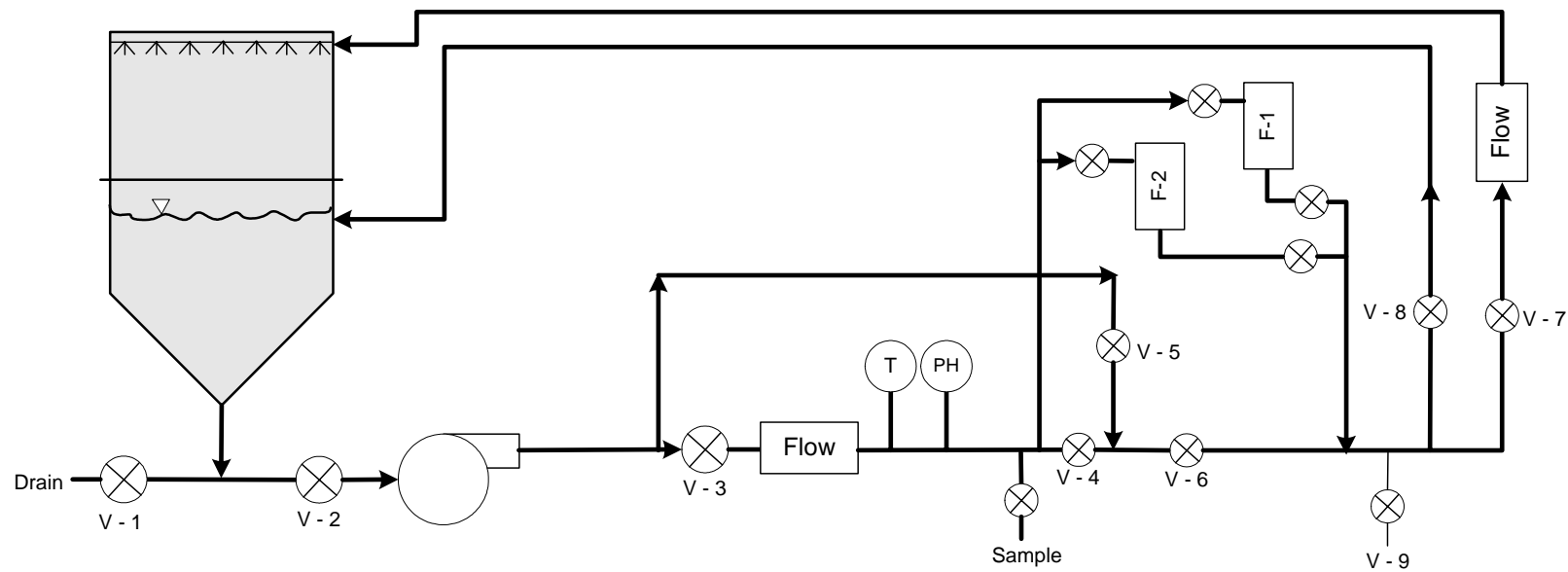
- | | | |
|------------|-------------|-------------|
| • Zinc | • Iron | • Lead |
| • Aluminum | • Nickel | • Sodium |
| • Silicon | • Calcium | • Potassium |
| • Copper | • Magnesium | • Chlorine |

As stated in Section 2, an objective of the tests described in this test plan is to determine and characterize the chemical reaction products that may develop in a representative post-LOCA containment sump environment. The characterization of these corrosion products will be used as input parameters to the NUREG-6224 head loss correlation to determine the effect of corrosion products on resulting head loss across a fibrous debris bed that might form on the containment sump screen.

In addition to the chemical species identified above, particulate corrosion products shall be analyzed to determine;

- The total mass of particulates from the grab sample on the filter surface
- X-ray diffraction analysis to identify the major compounds
- Density or specific gravity of each type of particulate, if possible
- Particle size distribution of particulates, e.g.
 - 1-10 microns
 - 11-25 microns
 - 26-50 microns
 - 51-75 microns
 - 76-100 microns
 - > 101 microns
- Evaluate the specific surface area of the particulates of each corrosion product, if possible.

As a minimum, the side stream filter is to be viewed under an optical microscope with the objective of characterizing the material that may have collected on the filter.

**Figure 1 – Test Loop Process Flow Diagram**

5 BASES FOR CONDITIONS INCLUDED IN THE GSI-191 POST-LOCA CORROSION TEST PLAN

5.1 Introduction

In order to develop the specific conditions, material types, and parameters to be included in a test program, and to assure that test results generated would be representative of all domestic PWRs, details of plant-specific conditions within containment (materials present, reactor coolant conditions, etc.) were necessary. To establish a representative set of detailed information, several sources of input were pursued. Westinghouse reviewed internally prepared plant-specific documents, such as Post-LOCA Hydrogen Generation Evaluations, other available plant documents (UFSARs) and issued survey questions to plant personnel. The plant survey responses formed the primary source of data for determining the parameters used to define the test conditions. The bases for selection of the parameters are discussed in this section.

5.2 Material Surface Areas

The primary concern for long-term corrosion in a post-LOCA environment is material that is submerged in reactor coolant water following containment flooding. Containment spray is actuated early during a design basis LOCA, but is terminated relatively early in the event (typically within 4 hours). Although some condensation may remain on equipment and material above the containment flood level, the amount of corrosion products contributed by this material is expected to be very small compared to the corrosion products generated by submerged materials. Therefore, the test parameters are primarily driven by the amount of material below the containment flood level at the point of switchover from injection to recirculation.

Representative values from the industry survey responses for material surface areas, with corresponding minimum post-LOCA sump volume of recirculation water, have been collected and are the basis for the material surface to sump water volume ratios identified in Table 1. Data that was collected from industry surveys and used to determine these values are shown later in Table 7. The surface area of each material type to be used in each test run will be the product of this ratio times the liquid volume in the tank of the test facility. The liquid volume of the test facility is 0.946 cubic meters (250 gallons). Although not expected at this time, if further refinement of the surface area to sump water volume ratio is warranted, based on additional information being made available prior to the start of testing, revised values should be substituted for those currently specified in Table 1.

Previous studies addressing post-LOCA corrosion (References 10 and 11) showed that aluminum and zinc, primarily in the form of galvanized carbon steel or as non-topcoated inorganic zinc based primer, will be the most corrosion susceptible materials in post-accident environment. Other less corrosion susceptible materials were also considered for this Test Plan. This was done because they were, in some cases, present in appreciable quantities (copper and copper based alloys), are known to be oxidants that can effect corrosion rates of other materials (copper ions), and may be sources of materials that may decrease the solubility of normally more soluble species when they are present (silica for fiberglass, calcium silicate and concrete). Table 7 contains the industry survey responses of exposed surface areas for a variety of materials within containment that would be susceptible to corrosion and/or chemical reaction when exposed to the post-LOCA chemical environment.

The goal in utilizing the collected data was to determine a representative area of individual material surfaces to be tested. These surface areas will then be exposed to a representative post-LOCA sump chemistry environment. This should result in a realistic quantity of dissolved and precipitated species in the simulated sump solution. These test results should be applicable to all PWRs.

For each material, Westinghouse attempted to determine a realistic upper bound estimate of the surface area to sump water volume ratio.

As discussed in Section 4.4.2, test coupons shall be prepared using the recommended practices outlined in ASTM G1. Exceptions or deviations to that guidance shall be documented and justified by the test performer.

5.2.1 Zinc

Zinc is present inside PWR containments in the form of both galvanized steel and zinc-based protective coatings. The following sections discuss the bases for the treatment of each of the two forms. The total area of zinc used in the test will be distributed among multiple coupons.

The surface area of zinc coupons to be used in each test run should be the product of the ratio of the surface area to liquid volume identified in Table 1, "Material Quantity/Sump Water Volume Ratios Planned to be Tested," times the test facility liquid volume. It is recommended that the coupons be 0.0625 inches thick⁸. This coupon thickness for zinc should be followed for all tests performed using this test plan.

⁸ Coupon thickness is to be 0.0625 (1/16) inches, except for those coupons to be coated with Inorganic Zinc (IOZ) primer, which are to be 0.0938 (3/32) inches thick.

5.2.1.1 Zinc in Galvanized Steel

Galvanized steel is present in the form of both electroplated and hot dipped.

A Westinghouse program to address Alloy 600 concerns calls for the addition of zinc acetate to the RCS. A preliminary assessment of the amount of zinc added under this program has determined that the amount of zinc added to the RCS inventory is small and is expected to have no impact on post-accident sump performance (Reference 10).

Zinc may be present in scaffolding stored in containment. However, since the scaffolding may be moved to an unsubmerged part of containment, or removed completely, the amount that may be present in scaffolding has not been accounted for in the surface to volume ratio used in the test.

5.2.1.2 Zinc Coatings

Generally, zinc coating has been topcoated with a DBA Qualified or “acceptable” epoxy, or a modified phenolic-epoxy topcoat. The industry survey returned very limited untopcoated zinc coatings in areas that could become submerged during a LOCA. However, untopcoated zinc will be included in the test to address concerns related to zinc primer exposed to post-accident conditions for those plants that reported having such coating systems inside containment.

The total surface area of zinc coatings inside containment available for reaction with the simulated sump solution does not reflect the surface area that is top coated with a qualified epoxy or modified phenolic epoxy based finish coat. These qualified topcoats protect the zinc primer from contact with reactive solutions and will reduce the surface area available for reaction with the simulated sump solution. Recently, however, NRC was made aware of the failure of coatings that were previously reported to be qualified. Therefore, a small amount of untopcoated zinc primer will be included in the test to account for the failure of coatings otherwise considered to be qualified.

5.2.2 Aluminum

Aluminum has been identified as the main contributor to hydrogen generation. Hence, the allowed quantity inside containment has typically been restricted (Reference 5) and tracked.

Aluminum may be present in scaffolding stored in containment. However, since the scaffolding may be moved to an unsubmerged part of containment, or removed completely, the amount that may be present in scaffolding has not been accounted for in the surface to volume ratio used in the test.

The surface area of aluminum coupons to be used in each test run should be the product of the ratio of the surface area to liquid volume identified in Table 1, "Material Quantity/Sump Water Volume Ratios Planned to be Tested," times the test facility liquid volume. It is recommended that the coupons be 0.0625 inches thick. This coupon thickness for aluminum should be followed for all tests performed using this test plan.

5.2.3 Copper and Copper Alloys

The major sources of copper inside containment are the containment fan coolers and CRDM coolers. Other potential, albeit smaller, sources are instrument air lines. The surface area of the latter is insignificant compared to the former. In some plants, such as ice condenser plants, the fan coolers are isolated during a LOCA; therefore, they are not considered to be a potential contributor of corrosion products for the post-LOCA sump solution. At other plants, the coolers may be partially submerged following an event.

In early testing (Reference 9), it was demonstrated that copper and its alloys corroded at low rates in a simulated post-LOCA environment. This testing also pointed out that alloying of copper with nickel further significantly lowered measured corrosion rates. Therefore, the Test Plan will conservatively test only Copper. Values for Copper-Nickel alloys were collected, and are combined with the values for Copper in Table 7 for the purpose of determining the surface to volume ratio for the test.

The surface area of copper coupons to be used in each test run will be the product of this ratio times the liquid volume in the tank of the test facility. ASTM G31 typically calls for coupons that are either 0.125 inch or 0.25 inch thick. However, 0.0625 inch (1/16 inch) thick coupons should be used for all tests performed using this test plan.

5.2.4 Carbon Steel

The value for carbon steel surface areas (both submerged and non-submerged in a postulated post-LOCA sequence) given in the test plan are representative of the US PWR fleet and are appropriate for use in chemical effects testing for the following reasons:

- Generally, carbon steel components and structures inside containment are either:
 - Protected by qualified coatings for protection (structural steel),
 - Located in portions of the containment that do not actively participate in the flow path for recirculating liquid from the sump (reactor vessel), or,
 - Encased in insulation not affected by the postulated pipe break (reactor vessel, components removed from and / or protected by cubicles and barriers).
- For B&W and CE design primary systems, primary system piping is constructed of inconel-clad carbon steel piping, encased in insulation. (Westinghouse design NSSS plants are constructed of stainless steel primary piping.) Also, steam generators are constructed of a carbon steel outer shell. For a postulated break, the insulation on primary system piping of B&W and CE NSSS designs, and on steam generators of all NSSS designs, would be removed within the Zone of Influence (ZOI), exposing the carbon steel outer surface. Depending upon the containment design, these surfaces may be exposed to containment spray during the active spray period (some containment designs will limit exposure of the carbon steel surface of these components).

Primary system piping and steam generators are not submerged post accident (they are above the post-accident flood-up level.) Therefore, these components are not subject to corrosion due to submergence in the containment pool. The duration of the exposure of the carbon steel piping to post-LOCA fluids is limited to the containment spray period. As noted above, some containment designs limit the direct exposure of primary system piping and steam generator shell surfaces to containment spray (for example, containments that are compartmentalized). The time duration of containment spray is short by comparison to the overall duration of the test (several hours versus several days), and both the corrosion rates and total corrosion are small (see the test plan for predicted containment steam space (non-submerged) temperature histories - time at temperatures above 200° F is relatively short). Moreover, examination of representative surfaces of components such as steam generators during in-service inspections indicates that there is practically no residual corrosion film on these components.

Therefore, carbon steel corrosion products are evaluated to not be a major contributor to the corrosion product mix in the post-accident sump of a PWR. This conclusion is supported by and consistent with the experimental results reported by Griess and Bacarella (Reference 9).

While being evaluated as a small contributor, the test plan does call for the inclusion of some bare, uncoated carbon steel surfaces. The surface area called for in the test plan ($0.15 \text{ ft}^2/\text{ft}^3$, 34%

submerged) is based on actual plant input obtained from a survey conducted of the PWR plants. Based on the discussion above, this value has been evaluated as an appropriate amount of carbon steel for use in the chemical effects test.

5.2.5 Concrete

A protective coating (paint) is generally applied to most concrete surfaces in containment. This protective coating is generally qualified for Design Basis Accidents (DBA's). A very small portion of concrete inside containment is expected to be untopcoated.

However, it is recognized that concrete will be eroded from the surfaces impacted by the jet around the initiating break location. For the purpose of the test, an assumed corrodible surface area will be defined as follows:

It will be assumed that a jet pressure of 10 psi is needed to erode concrete surfaces. A break diameter of 32 inches is assumed. Using the ANSI58.2-1988 jet expansion model, at the 10 psi isobar of the jet, the volume of the jet is calculated to be about 133,800 ft³. The radius of a sphere having the equivalent volume as the jet at the 10 psi isobar is calculated to be about 9.7 meters (31.7 feet). The surface area of this equivalent sphere is then calculated to be about 1174 m² (12,630 ft²). This approach provides for a realistically conservative approach to calculating the concrete surface area exposed due to the action of a jet from a postulated pipe break. As noted in Attachment A to the PWR Containment Sump Baseline Evaluation Method, protective coatings are observed to withstand pressures in excess of 1000 psi. If the coatings remain intact, the concrete beneath the coatings also remain intact. The approach taken in this test plan is to use a 10 psi isobar to determine a spherical ZOI. Then, the full surface area of the 10 psi isobar ZOI is taken to be the maximum area of the concrete surface that is exposed due to the action of the jet.

The actual area of concrete that comes into contact with the expanding jet depends on the configuration of the containment and the break location. The value of 1174 m² (12,630 ft²) is taken as a maximum surface area of concrete that will be exposed to erosion and will be used to set the concrete surface area to water volume ratio for the test.

In addition to the exposed concrete surface area resulting from the break, there may be initial quantity of concrete dust particulates in containment. The volume of this source of material will be accounted for as an initial mass of 22.7 kg (50 lbm) of concrete dust. This mass is based on an evaluation of current containment conditions. This ratio will apply to all plant types.

5.2.6 Insulation Material

Debris is generated within the Zone of Influence (ZOI) by the fluid escaping from the postulated break. Typical insulation materials include reflective metallic insulation (RMI), fiberglass and calcium silicate. In addition to deposition of the debris on the screen of a containment sump, this debris may be a potential source for nucleation of precipitants. Fiberglass and calcium silicate insulation debris may react at high pH values and release silica. Therefore, the tests will provide for the study of the possible interaction of insulation debris with the recirculation solution by chemical reaction.

The amount of fiberglass or calcium silicate inside containment depends on the specific plant design. Some plants use primarily reflective metallic insulation in the area that might be affected by a postulated large break in the primary piping. These plants have effectively no fiberglass or calcium silicate debris.

The largest component that may have fiberglass or calcium silicate insulation in the area that might be affected by a postulated large break is the steam generator. Based on the dimensions of a steam generator and accounting for a conservatively large ZOI volume, a preliminary calculation of the volume of fiberglass or calcium silicate insulation to be used to set the fiberglass to water ratio of this test has been estimated to be 141.6 m³ (5,000 cubic feet). This number is representative for a PWR that uses fiberglass or calcium silicate insulation in general and will be used for the test.

The fiberglass or calcium silicate inside containment will be either blown onto the containment floor by the jet, or upward into upper containment.

- Large pieces of fiberglass or calcium silicate insulation blown into upper containment may not be transported back to lower containment due to curbs and gratings. This insulation will be subjected to containment spray flow, but will not undergo long-term submergence.
- All fiberglass or calcium silicate insulation that is blown onto the containment floor, and fiberglass or calcium silicate insulation that is washed onto the containment floor due to the action of containment spray, will be submerged in or floating on the liquid on the containment floor.

The test will provide for fiberglass or calcium silicate insulation that is both subjected to containment spray, and submerged long term.

As noted above, the same ratio approach may be applied to calcium silicate as was applied for fiberglass.

5.2.7 Protective Coatings

Epoxy-based protective coatings (paints) will not be included in the testing described in this test plan. In the event of an accident such as a LOCA, epoxy-based protective coatings on structures, systems and components in Pressurized Water Reactor (PWR) containments may be exposed to severe chemical environments. Recognizing this, the US nuclear industry issued ANSI N5.9-1967, later replaced by ANSI N5.12-1974, "Protective coatings (paints) for the nuclear industry," which is a screening methodology for candidate reactor containment coatings. Section 5 of ANSI N5.9/5.12, entitled "Chemical-Resistance Tests" provides, "...a common basis for methods and procedures for the evaluation of the resistance of coating systems to chemical environments." Candidate epoxy-based protective coating samples, for consideration for use in "severe" exposures such as reactor containment, are immersed for 5 days in 5% solutions of nitric acid, sulfuric acid, hydrazine, sodium hydroxide, ammonium hydroxide, and/or sodium borate as appropriate (see Subsection 5.3 of ANSI N5.12-1974). The samples are visually examined at 24 hour intervals during testing and again at 5 days, the completion of testing. Test samples are evaluated based on the following test standards or criteria:

- ASTM D772, "Standard Method of Evaluating Degree of Flaking (Scaling) of Exterior Paints" -
Delamination - none permitted
- ASTM D714, "Standard Method of Evaluating Degree of Blistering of Paints" -
Discoloration - will be permitted
- Other effects - noted for evaluation on an individual basis

Evaluation results are documented in writing (test reports). These test reports are retained and are available for the various coating manufacturers and plant licensees. Only epoxy-based protective coatings which have been screened in accordance with ANSI N5.9/N5.12 are subsequently DBA tested in accordance with ANSI N101.2, "Protective Coatings (Paints) for Light Water Nuclear Reactor Containment Facilities" for potential designation as "DBA qualified."

Epoxy coatings have been tested for leachable materials. These tests have shown that the amount of leachable material, primarily chlorides and fluorides, resulting in concentrations in the order of parts per million or less.

For plants to be licensed in the future, the chemical effects testing provisions of ANSI N5.9/N5.12 have been replaced by ASTM D 3912, "Test Method for Chemical Resistance of Coatings Used in Light-Water Nuclear Power Plants." ASTM D 3912 is identical in function and requirements to the chemical effects testing provisions of ANSI N5.9/N5.12. As with epoxy-based protective coatings DBA qualified for earlier plants, only coatings which have been screened in accordance with ASTM D 3912 are subsequently DBA tested in accordance with ASTM D 3911, "Test Method for Evaluating Coatings Used in Light-Water Nuclear Power Plants at Simulated Design Basis Accident (DBA) Conditions" for potential designation as "DBA qualified."

It is also noted that epoxy-based protective coatings are used in highly basic and highly caustic environmental commercial applications where leaching of protective coating material is undesirable. Examples of specific applications include direct contact with food, potable and non-potable water, sanitation applications and BWR torus coatings. Epoxy coatings have been tested by coatings manufacturers for leachable materials. These tests have demonstrated that the amount of leachable material in epoxy coatings, primarily chlorides and fluorides, is small and results in leached concentrations in the order of parts per million or less⁹ of the applied coating. Therefore, epoxy-based coatings are not considered as a source of leachable materials and are not included in this test program.

The above discussion pertains to DBA qualified coatings. The Electric Power Research Institute (EPRI) is performing testing on representative unqualified coatings used in nuclear power plants to determine if these coatings fail. The information from those tests will be combined with information from this test. Therefore, inclusion of coatings in the current test program is either not warranted for qualified or DBA qualified coatings, or is redundant for unqualified coatings.

5.2.8 Other Materials

Other materials have been identified and evaluated for contribution to corrosion products in a post-LOCA environment inside containment.

⁹ Testing for leachable materials is performed using ASTM Standard D1179, ASTM Standard F1277 and ANSI NSF-61. A typical upper limit of leachable materials from epoxy-based coatings is 200 ppm of the applied coating.

- Nickel is bound inside non-corrodible stainless steel. Nickel is also a constituent of crud that forms on fuel. Based on plant measurements, the average nickel release for a Westinghouse 4-loop PWR is 3500 grams, or 7.7 pounds, which equates to a concentration of about 0.16 ppm in the sump fluid inventory. This concentration is judged to be sufficiently small that it may be ignored for this test.
- Calcium, magnesium and silicon will be present from concrete and insulation material dissolution. Their levels will not be controlled directly, but will be allowed to evolve as dictated by the variables that effect concrete dissolution such as temperature and pH.
- Likewise, a variety of corrosion products may be added to solution from the corroding aluminum and zinc coatings and the corrosion of underlying exposed steel. These concentrations will be measured but will be controlled only by chemical and physical conditions that effect corrosion.

5.3 Material Surface Submerged and Exposed to Spray

Table 3 contains estimates of the percentages of each material that would be submerged and exposed to spray at maximum post-accident sump volume. In each case, material not submerged was assumed to be exposed to spray. The values in the Test Plan were established based on observations and experienced judgments of knowledgeable senior members of the Westinghouse engineering staff, as well as from responses to the industry survey.

As with the surface area to sump water volume ratio values, additional verification will be requested of the responding utility personnel that were outside the Test Plan value. The percent submerged value for these materials may be further refined prior to the start of testing.

For concrete and fiberglass the percent submerged values were based on the Zone of Influence calculation described in the Test Plan.

5.4 Test Temperature

Figures 2 through 7 are calculated post-LOCA temperature profiles for various PWR containment types.

These temperature profiles are based on LOCA Containment Integrity analyses performed to demonstrate that the containment heat removal systems are adequate. These analyses usually appear in FSAR Chapter 6.2 and are based on the Westinghouse LOCA M&E model described in WCAP-10325-P-A. Since these analyses assume the operation of a single train of

Containment Spray and Emergency Core Cooling Systems, these assumptions provide for the calculation of a conservatively high containment pressure, containment atmospheric temperature, and sump water temperature. Therefore, use of the attached curves, developed for the most limiting break, Double Ended Pump Suction break assuming a loss of offsite power and failure of the Diesel Generator to start, will provide a conservatively high estimate of the sump water temperature post-LOCA.

A comparison of Figures 2 through 7 shows that these calculated temperature profiles vary considerably with containment design. However, they all show the same initial temperature elevation early in the accident, followed by a gradual downward trend in temperature with time. From the figures, it is noted that some plants, like those with the ice condenser design, drop below 60° C (140° F) very rapidly. Others, such as the 4-loop designs, drop to 62° C (150° F) in approximately 4 days, reach 60° C (140° F) in less than 9 days, and continue to decrease thereafter. Still others like the small three loop designs seem to remain at elevated temperature much longer.

There are general “rules of thumb” that state that for every 10° C increase in reaction temperature the rate of reaction will double. From solubility considerations, it is known that while the solubility of aluminum oxides increase with increasing temperature the reverse is true for zinc oxides. So, it is evident that selecting a temperature for the testing will have a significant impact on test results.

A constant temperature of 60° C (140° F) was selected for the test. The rationale for the specific value is that the temperature profiles shown in Figures 2 through 7 are conservatively large values, as described above. Realistic analysis assumptions will provide for the calculation of lower values earlier in the transient for both the 3-loop and 4-loop plants. The use of a single representative, but conservative, value simplifies test operation. Thus, considering these factors and the effects of temperature on chemical reaction rates, a constant temperature of 60° C (140° F) is chosen to be representative of long term sump flood conditions across the fleet of PWRs, including ice condenser plants, for the purposes of this test.

5.5 pH

The following sections identify the pH requirements for the containment spray simulation and the sump pool simulation of this test.

5.5.1 Spray Fluid pH

The pH of containment spray for plants using TSP as a buffering agent is the pH of the RWST tank and has been calculated to vary from about 7 to about 8.5. For preconditioning test coupons and fiberglass samples, a pH of 7 is selected to represent plants using TSP as a buffering agent because at the lower pH, the potential for forming precipitants is greater than with a pH of 8.5. Thus, using a lower pH will increase the probability of the formation of precipitants.

The pH of containment spray for plants using NaOH as a buffering agent is calculated to be about 12 during the NaOH injection phase. Therefore, to simulate conditions for plants using NaOH, the initial 30 minutes of containment spray simulation will have a maximum pH of 12. See Section 4.1.3.1, "pH," and Note (2) to Table 5, "Test Run Conditions," for additional discussion on the pH of the NaOH spray simulation.

Containment spray will include the Sodium Tetraborate buffering agent mixed into the test loop inventory at the start of test. No attempt is made in the test to simulate the ice melting process of an ice condenser containment, which would require the gradual introduction of Sodium Tetraborate into the recirculating fluid. Not performing this gradual introduction results in the non-submersed coupons being subjected to a spray solution having a pH associated with the complete dissolution of the buffering agent in the recirculating flow at the start of test.

5.5.2 Sump Fluid pH

The initial pH of containment sump water for plants using TSP as a buffering agent has been calculated to vary from about 7 to about 8.5. A pH of 7 was selected to represent plants using TSP as a buffering agent because at the lower pH, the potential for forming precipitants is greater than with a pH of 8.5. Thus, using a lower pH will increase the probability of the formation of precipitants.

An initial pH of containment sump water for plants using NaOH as a buffering agent has been calculated to vary from about 8 to 11. A pH for 10 was selected to represent plants using NaOH as a buffering agent.

The pH of for the Sodium Tetraborate test will be determined from achieving a Boron concentration of 2400 ppm from mixing a Boric Acid solution having a Boron concentration of 2800 ppm with a Sodium Tetraborate solution having a Boron concentration of 2100 ppm. This Boron concentration is representative of the containment pool inventory at the time of the ice bed melt and will yield a pH representative of ice condensers at that time in the postulated LOCA event.

6 REFERENCES

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4-LOOP (412) PWR AT 3600 MWT RATING DEPS WITH MINIMUM SAFEGUARDS

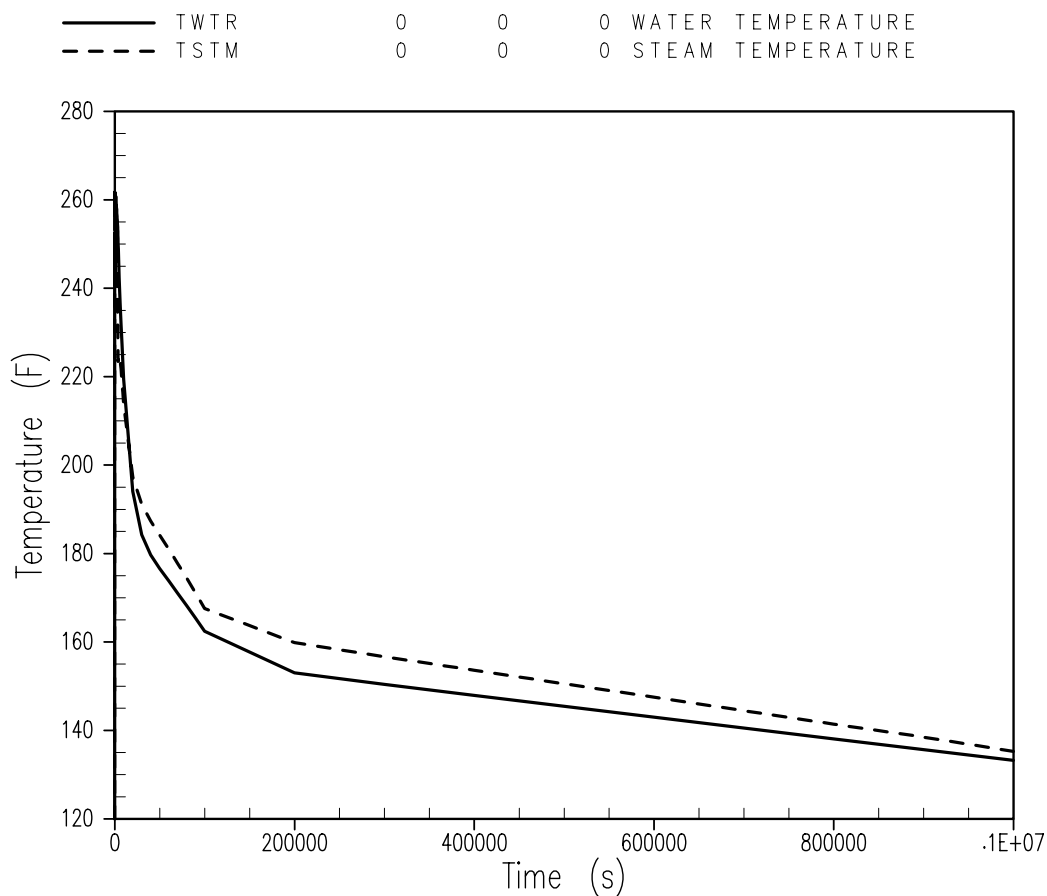


Figure 2: Four Loop Post-LOCA Temperature Profile (Linear Time Scale).

4-LOOP (412) PWR AT 3600 MWT RATING DEPS WITH MINIMUM SAFEGUARDS

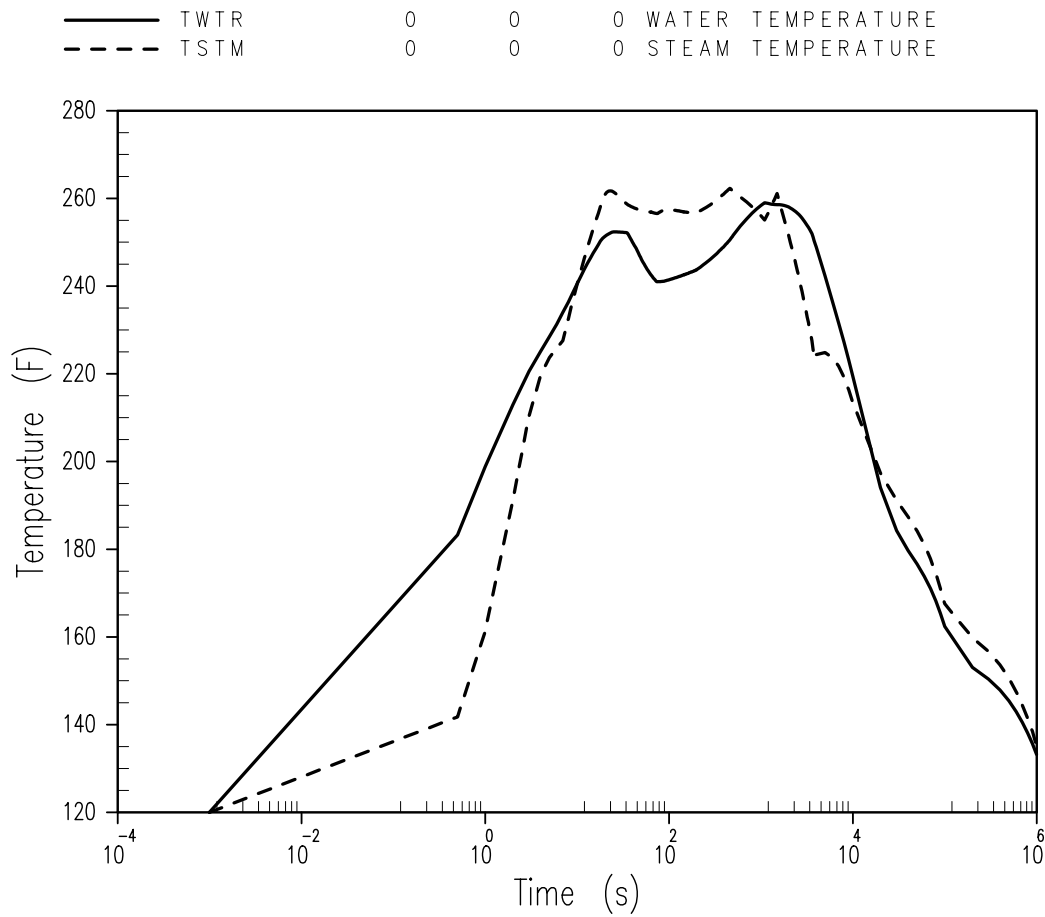


Figure 3: Four Loop Post-LOCA Temperature Profile (Logarithmic Time Scale).

SMALL 3 LOOP PWR RATED AT 2300 Mwt DEPS WITH MINIMUM SAFEGUARDS

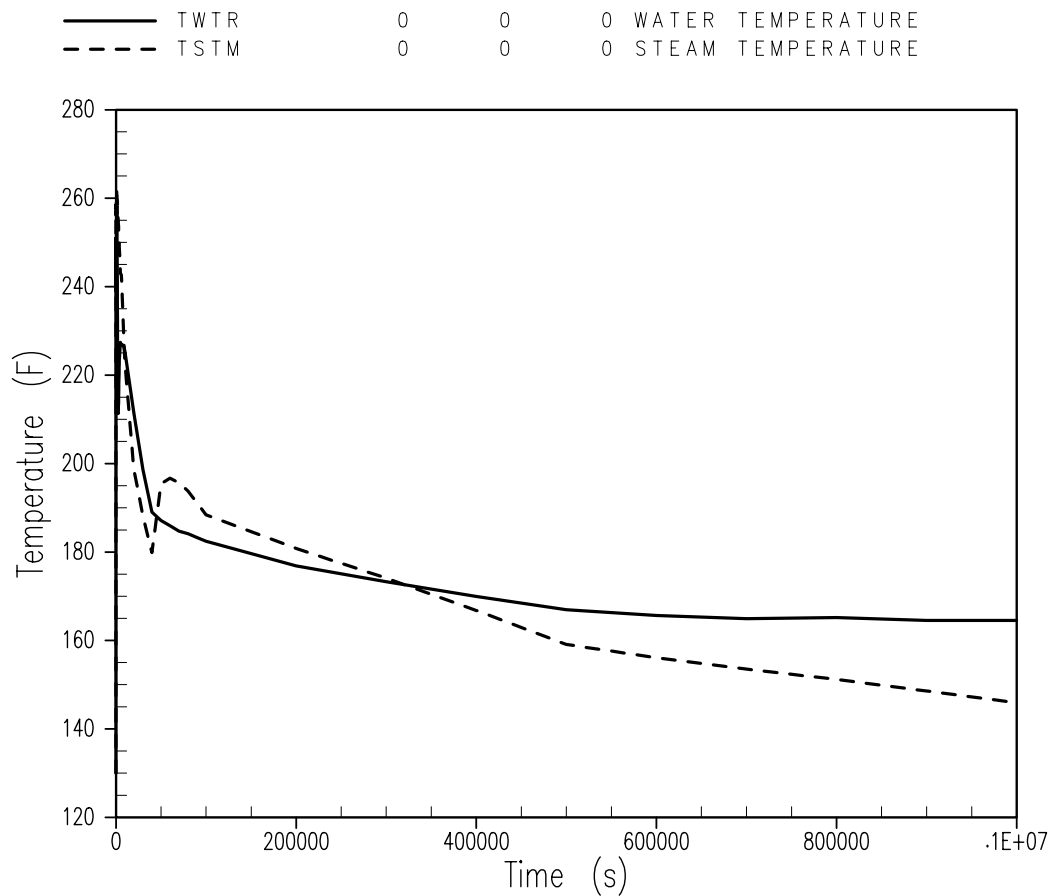


Figure 4: Three Loop Post-LOCA Temperature Profile (Linear Time Scale).

SMALL 3 LOOP PWR RATED AT 2300 Mwt DEPS WITH MINIMUM SAFEGUARDS

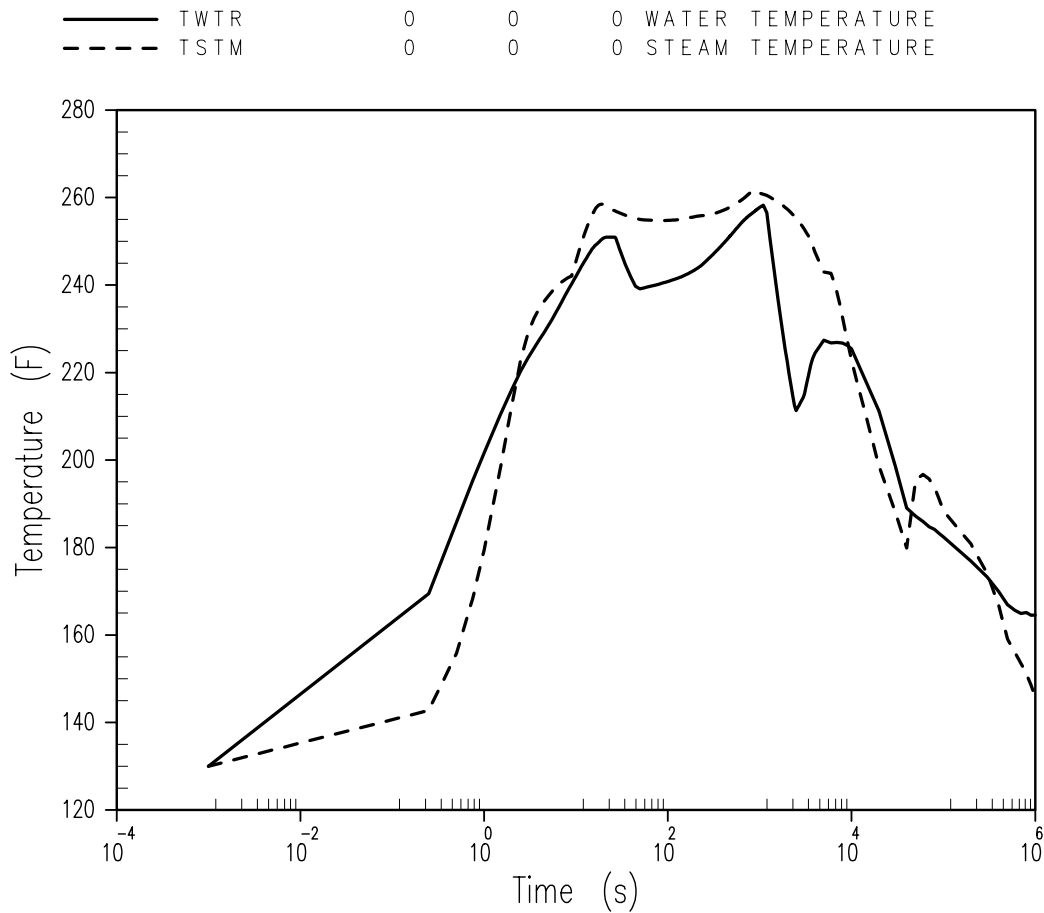
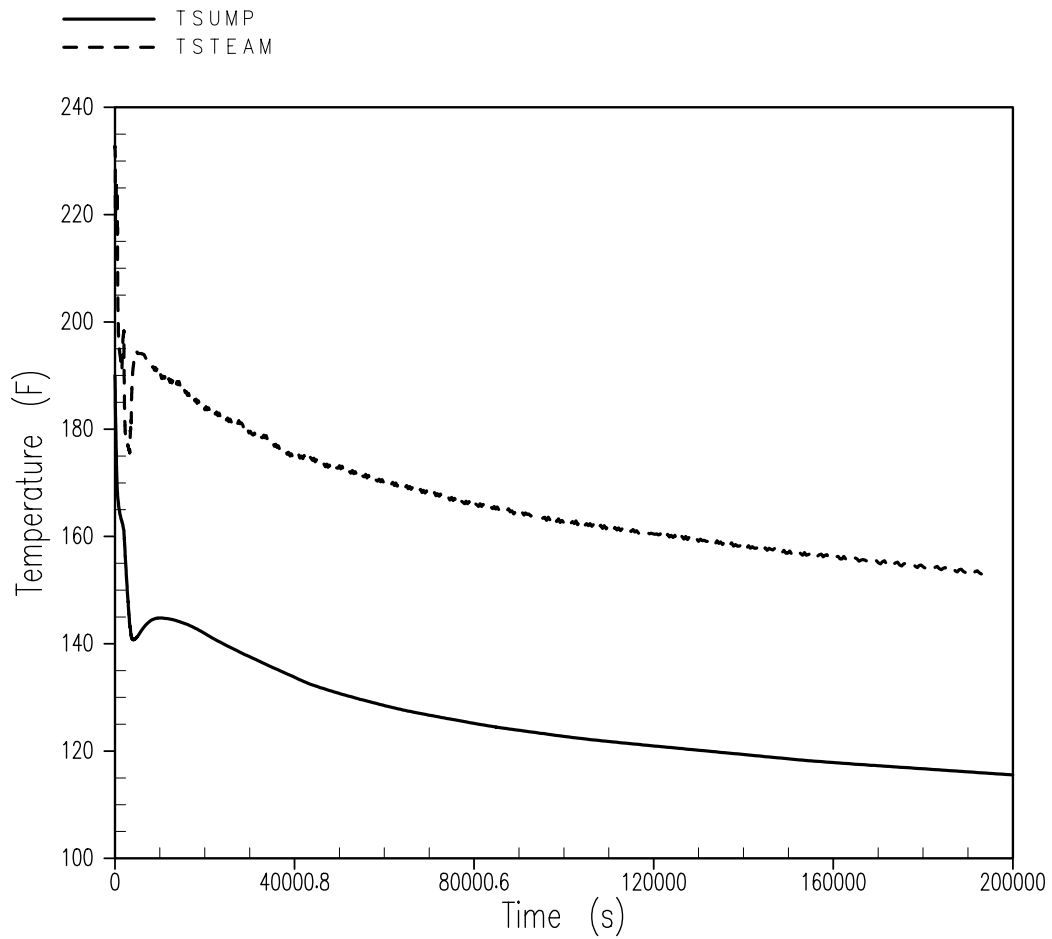


Figure 5: Three Loop Post-LOCA Temperature Profile (Logarithmic Time Scale).

4 LOOP ICE CONDENSER PLANT ICE WEIGHT OPTIMIZATION PROJECT DEPS WITH MINIMUM SAFEGUARDS



**Figure 6: Four Loop Ice Condenser Plant Post-LOCA Temperature Profile
(Linear Time Scale).**

4 LOOP ICE CONDENSER PLANT ICE WEIGHT OPTIMIZATION PROJECT
DEPS WITH MINIMUM SAFEGUARDS

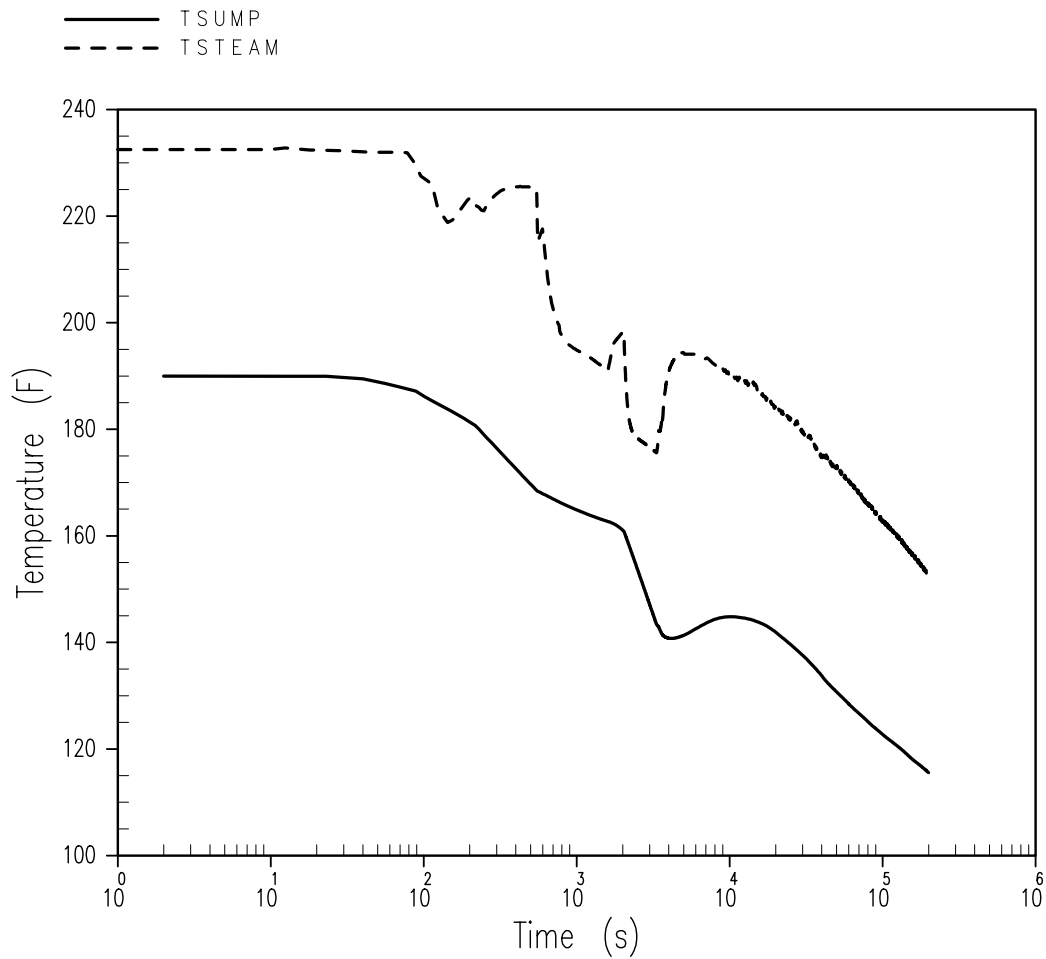


Figure 7: Four Loop Ice Condenser Plant Post-LOCA Temperature Profile (Logarithmic Time Scale).

Table 7: Industry Survey Response Data

Plant	Plant Type	Containment Type	Min Sump Vol (ft ³)	Total Mat'l (ft ²)	Submerged %	Submerged Mat'l	Mat'l Exposed to Spray	Ratio (Total)	Ratio (Submerged)	Ratio (Sprayed)
Galvanized Zn										
T	B&W	Dry	37,100.00	65,000.00	1	650.00	64,350.00	1.75	0.02	1.73
U	CE	Dry	55,000.00	105,979.00	1	1,059.79	104,919.21	1.93	0.02	1.91
J	3 Loop	Small Sub-Atm	68,809.00	128,845.00	N/A					
K	3 Loop	Small Sub-Atm	126,472.00	176,257.00	N/A					
Q	4 Loop	Large Dry	50,418.28	106,300.00	5	5,315.00	100,985.00	2.11	0.11	2.00
BB	B&W	Dry	36,682.00	8,000.00	1	80.00	7,920.00	0.22	0.00	0.22
N	2 Loop	Small Dry	41,073.00	63,596.00	5	3,179.80	60,416.20	1.55	0.08	1.47
JJ	4 Loop	Large Dry	159,740.00	325,215.00	3.5	11,382.53	313,832.48	2.04	0.07	1.96
S, KK, LL	B&W	Dry	52,600.00	43,970.00	N/A			0.84		
R	CE	Dry	40,758.00	74,540.00	5	3,727.00	70,813.00	1.83	0.09	1.74
O, P	2 Loop	Small Dry	44,200.00	60,000.00	10	6,000.00	54,000.00	1.36	0.14	1.22
RR	4 Loop	Large Dry	42,131.00	355,000.00	3	10,650.00	344,350.00	8.43	0.25	8.17
QQ	3 Loop		35,134.00	24,859.39	0	0	24,859.39	0.71	0.00	0.71
X	4 Loop	Large Dry	61,209.00	127,520.00	3	3,825.60	123,694.40	2.08	0.06	2.02
Max Ratios								8.43	0.25	8.17
Topcoated Zn Coatings										
T	B&W	Dry	37,100.00	N/A						
U	CE	Dry	55,000.00	N/A						
J	3 Loop	Small Sub-Atm	68,809.00	85,099.00	N/A			1.24		
K	3 Loop	Small Sub-Atm	126,472.00	263,163.00	N/A			2.08		
Q	4 Loop	Large Dry	50,418.28	131,443.00	1	1,314.43	130,128.57	2.61	0.03	2.58
BB	B&W	Dry	36,682.00	N/A	N/A					
N	2 Loop	Small Dry	41,073.00	139,555.00	5	6,977.75	132,577.25	3.40	0.17	3.23
JJ	4 Loop	Large Dry	159,740.00	11,886.00	5	594.30	11,291.70	0.07	0.00	0.07

Plant	Plant Type	Containment Type	Min Sump Vol (ft ³)	Total Mat'l (ft ²)	Submerged %	Submerged Mat'l	Mat'l Exposed to Spray	Ratio (Total)	Ratio (Submerged)	Ratio (Sprayed)
S, KK, LL	B&W	Dry	52,600.00	374,500.00	N/A			7.12		
R	CE	Dry	40,758.00	129,797.00	15	19,469.55	110,327.45	3.18	0.48	2.71
O, P	2 Loop	Small Dry	44,200.00	146,900.00	5	7,345.00	139,555.00	3.32	0.17	3.16
RR	4 Loop	Large Dry	42,131.00	4,000.00	3	120.00	3,880.00	0.09	0.00	0.09
QQ	3 Loop		35,134.00	0	0					
X	4 Loop	Large Dry	61,209.00	289,200.00	1	2,892.00	286,308.00	4.72	0.05	4.68
Max Ratios								7.12	0.48	4.68
Untopcoated Zn Coatings										
T	B&W	Dry	37,100.00		1					
U	CE	Dry	55,000.00		1					
J	3 Loop	Small Sub-Atm	68,809.00	N/A	N/A					
K	3 Loop	Small Sub-Atm	126,472.00	30,326.00	N/A			0.24		
Q	4 Loop	Large Dry	50,418.28	228,657.00	0.1	228.66	228,428.34	4.54	0.00	4.53
BB	B&W	Dry	36,682.00	100.00	10	10.00	90.00	0.00	0.00	0.00
N	2 Loop	Small Dry	41,073.00	7,345.00	5	367.25	6,977.75	0.18	0.01	0.17
JJ	4 Loop	Large Dry	159,740.00	N/A	N/A					
S, KK, LL	B&W	Dry	52,600.00	0	N/A					
R	CE	Dry	40,758.00	N/A	N/A					
O, P	2 Loop	Small Dry	44,200.00	N/A	N/A					
RR	4 Loop	Large Dry	42,131.00	N/A	N/A					
QQ	3 Loop		35,134.00	0	0					
X	4 Loop	Large Dry	61,209.00	76,500.00	0.05	38.25	76,461.75	1.25	0.00	1.25
Max Ratios								4.54	0.01	4.53
Aluminum										
T	B&W	Dry	37,100.00	10,750.00	1	107.50	10,642.50	0.29	0.00	0.29
U	CE	Dry	55,000.00	1,206.00	1	12.06	1,193.94	0.02	0.00	0.02
J	3 Loop	Small Sub-Atm	68,809.00	1,559.00	N/A			0.02		
K	3 Loop	Small Sub-	126,472.00	1,559.00	N/A			0.01		

Plant	Plant Type	Containment Type	Min Sump Vol (ft ³)	Total Mat'l (ft ²)	Submerged %	Submerged Mat'l	Mat'l Exposed to Spray	Ratio (Total)	Ratio (Submerged)	Ratio (Sprayed)
		Atm								
Q	4 Loop	Large Dry	50,418.28	2713.39 lbm	5					
BB	B&W	Dry	36,682.00	3,000.00	1	30.00	2,970.00	0.08	0.00	0.08
N	2 Loop	Small Dry	41,073.00	203.20	1	2.03	201.17	0.00	0.00	0.00
JJ	4 Loop	Large Dry	159,740.00	18,979.00	1	189.79	18,789.21	0.12	0.00	0.12
S, KK, LL	B&W	Dry	52,600.00	27,800.00	N/A			0.53		
R	CE	Dry	40,758.00	136,818.00	25	34,340.00	102,478.00	3.36	0.84	2.51
O, P	2 Loop	Small Dry	44,200.00	900.00	10	90.00	810.00	0.02	0.00	0.02
RR	4 Loop	Large Dry	42,131.00	1,800.00	3	54.00	1,746.00	0.04	0.00	0.04
QQ	3 Loop		35,134.00	670.00	0	0.00	670.00	0.02	0.00	0.02
X	4 Loop	Large Dry	61,209.00	854.00	5	42.70	811.30	0.01	0.00	0.01
Max Ratios								3.36	0.84	2.51
Cu/Cu Alloys										
T	B&W	Dry	37,100.00	191,400.00	1	1,914.00	189,486.00	5.16	0.05	5.11
U	CE	Dry	55,000.00	191,400.00	1	1,914.00	189,486.00	3.48	0.03	3.45
J	3 Loop	Small Sub-Atm	68,809.00	4,754.00	N/A			0.07		
K	3 Loop	Small Sub-Atm	126,472.00	4,824.00	N/A			0.04		
Q	4 Loop	Large Dry	50,418.28	116,207.60	0	0	116,207.60	2.30	0.00	2.30
BB	B&W	Dry	36,682.00	0	N/A			0.00		
N	2 Loop	Small Dry	41,073.00	60,844.00	0	0	60,844.00	1.48	0.00	1.48
JJ	4 Loop	Large Dry	159,740.00	35,495.00	N/A			0.22		
S, KK, LL	B&W	Dry	52,600.00	102,728.00	0	0	102,728.00	1.95	0.00	1.95
R	CE	Dry	40,758.00	66,526.00	25	16631.5	49,894.50	1.63	0.41	1.22
O, P	2 Loop	Small Dry	44,200.00	80,000.00	0	0	80,000.00	1.81	0.00	1.81
RR	4 Loop	Large Dry	42,131.00	2,618.00	0	0	2,618.00	0.06	0.00	0.06
QQ	3 Loop		35,134.00	7,610.00	0	0	7,610.00	0.22	0.00	0.22
X	4 Loop	Large Dry	61,209.00	162,051.00	0	0	162,051.00	2.65	0.00	2.65
Max Ratios								5.16	0.41	5.11