



REGIONAL
HYDROGEOCHEMISTRY

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Scientific Notebook #094
Supporting the Regional
Hydrology Research Project

CONTROLLED

#094

This notebook is intended to
be used to document activities
related to the Regional
Hydrology Research Project

William M. Murphy is
identified by WMM

William M. Murphy
11/14/93

On 11/18/93 a rock sample was collected. Its description prepared by H.L. McKague is attached to page 3 of this book.

WLF 11/23/93

WLF 6/20/94

Rock sample HT(Hydrology Test) 11-18-93-001 was collected by Budhi Sagar on Nov. 18, 1993; at DOE well WT-7 at the south end of Boomerang Ridge, Nye County, Nevada. The sample was taken from the outcrop at the west edge of the drill pad. The rock is a devitrified densely welded ash-flow tuff from the Tiva Canyon Formation. Secondary vein filling mineralization occurs on two exterior surfaces and as a 1/8 th inch thick vein. The secondary mineralization in the vein and on the smaller exterior surface seem to consist of relatively pure carbonate. The secondary mineralization on the larger exterior surface appears to contain secondary silica mineralization, brecciated Tiva Canyon Tuff in addition to the carbonate mineralization.

WLF 11/23/93

Pages 1-4 of this notebook were reviewed for compliance with QA-001 in response to CR 94-02.

Verified by: Randy Jolck
SWRE-QA
7/12/94

WJL
2/24/95

Analysis of water distribution in UZ-16

Porosity : 30% *

Saturation : 95% *

* from Withmeyer/Flint for the peditized Calico Hills

UZ-16 is near USW H-4

H-4 is at 4096.5' elevation and UZ-16 is at 4001' elevation per GIS data base

Calico Hills formation is from 1205.5' to 1465.9' depth in UZ-16 per Withmeyer/Flint

Corresponding depths in USW-H4 are from 1301' to 1560.5'.

WJL
2/24/95

Zeolites from 1312' depth
in H.4 are given by
Braxton et al 1986

A typical one is

~~K~~ 6.16 $\text{Al} 2/24/95$

K Na Ca
6.33 6.22 0.43

Al Fe Si Ti
13.91 0.61 72.5 0.03

O
173.545 = n H₂O

Based on 24 structural oxygens

~~K~~ Na Ca Al Fe Si O
0.88 0.86 0.06 1.92 0.08 10.03 24
• n H₂O

$\text{Al} 2/25/95$

Bish et al (1989) gives
mineralogy of H.4.
A clinoptilolite zone
extends from 1300' depth
to 1550' depth. These
rocks are 30 to 80%
clinoptilolite.

An End member rock from
this zone given in A.

% weight fraction

	A	B
Clinopt.	78±10	70±14 + 9±2 mordenite
Quartz	3±2	17±5 Tr $\text{Al} 2/24/95$
Crystobalite	9±4	-
Alkali fs	10±5	3±1
Opal-CT	<u>101</u>	17±3

however there is not Opal-CT
given in this older analysis.
Rocks from the same stratum from
VFZ 25a#1 typically have 20% opal.
A silica rich-zeolite rich variety
is given in B.

$\text{Al} 2/24/95$

A lower clinophilitite left
from the Calico Hills of
H-4 is

smectite	3 ± 2	% weight
clinopt	29 ± 10	
mordenite	16 ± 5	
quartz	11 ± 3	
crystalite	10 ± 4	
alk. fs.	31 ± 10	

Given 1 dm³ (= 1 liter) of
rock $.95 \times 300 \text{ cm}^3 = 285 \text{ cm}^3$
of water in porosity

Assuming weight % \approx volume %
of mineralogy

$$78\% \text{ wt \% clinopt} \times 700 \text{ cm}^3 \\ = 546 \text{ cm}^3 \text{ clinophilitite}$$

Using a density of 2.16 for
Na,K clinophilitite (from Deer Howie + Z.)

$$546 \text{ cm}^3 \times \frac{2.16 \text{ g}}{\text{cm}^3} = 1179 \text{ g}$$

Wll 9/24/95

Assuming a stoichiometric 8 H₂O
per 24 structural oxygens
the molar mass of our H-4
zeolite is 906.8 g/mole.
Wll 9/24/95
906 g/mole.

So our 1179 g of
clinophilitite \approx 1.3 moles

$$1.3 \text{ mole clinopt.} \\ \times 8 \text{ moles H}_2\text{O} / \text{mole clinopt.} \\ \times 18 \text{ g H}_2\text{O} / \text{mole H}_2\text{O} \\ = 187.2 \text{ g H}_2\text{O}$$

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See page attached

Excerpt from
Broxton et al. 1986

APPENDIX E (cont)

Drill Hole USW H-4

Depth (ft) ^a	1312	1312	1312	1312	1312	1312
Unit ^a	Tht	Tht	Tht	Tht	Tht	Tht
SiO ₂	73.9	74.3	73.4	74.0	74.0	74.0
TiO ₂	0.04	0.04	0.02	0.05	0.05	0.01
Al ₂ O ₃	12.0	12.1	12.0	12.3	12.0	12.0
Fe ₂ O ₃	0.67	0.83	0.86	0.77	0.83	0.73
MgO	0.00	0.00	0.00	0.00	0.00	0.00
CaO	0.43	0.41	0.43	0.45	0.49	0.43
BaO	0.00	0.00	0.00	0.00	0.00	0.00
Na ₂ O	3.27	3.29	3.03	3.38	3.28	3.16
K ₂ O	4.91	5.09	5.39	4.96	5.37	5.25
Total	95.2	96.1	95.1	95.9	96.0	95.6
Cation Percent						
Si	72.7	72.5	72.4	72.2	72.2	72.6
Ti	0.03	0.03	0.01	0.04	0.04	0.01
Al	13.92	13.91	13.95	14.15	13.80	13.87
Fe	0.50	0.61	0.64	0.57	0.61	0.54
Mg	0.00	0.00	0.00	0.00	0.00	0.00
Ca	0.45	0.43	0.45	0.47	0.51	0.45
Ba	0.00	0.00	0.00	0.00	0.00	0.00
Na	6.24	6.22	5.79	6.39	6.20	6.01
K	6.16	6.33	6.78	6.17	6.68	6.57
Si/(Al+Fe)						
	5.04	4.99	4.96	4.91	5.01	5.03
(Al+Fe)/(2Mg+2Ca+2Ba+Na+K)						
	1.08	1.08	1.08	1.09	1.04	1.07
Mol% Exchangeable Cations						
K	47.9	48.8	52.0	47.4	49.9	50.4
Na	48.5	47.9	44.5	49.0	46.3	46.1
Ca+Mg	3.5	3.3	3.5	3.6	3.8	3.5
Oxides Recalculated to 100%						
SiO ₂	77.61	77.35	77.16	77.16	77.07	77.42
TiO ₂	0.04	0.04	0.02	0.05	0.05	0.01
Al ₂ O ₃	12.60	12.60	12.61	12.82	12.50	12.55
Fe ₂ O ₃	0.70	0.86	0.90	0.80	0.86	0.76
MgO	0.00	0.00	0.00	0.00	0.00	0.00
CaO	0.45	0.43	0.45	0.47	0.51	0.45
BaO	0.00	0.00	0.00	0.00	0.00	0.00
Na ₂ O	3.43	3.42	3.19	3.52	3.42	3.31
K ₂ O	5.16	5.30	5.67	5.17	5.59	5.49

my 2/25/9.

This research project was formally closed on January 19, 1996. This notebook contains a brief outline of geochemical speciation studies conducted for groundwater collected and analyses conducted by others. Most of the work described herein is summarized in detail in the semi-annual research reports.

Gordon Wittmeyer
Principal Investigator

2/5/96