

308 --- Q200303210005  
Scientific Notebook # 382: Thermal Hydrology  
KTI Project (KEF)

Drift Scale Header Test  
Tetpi

11

21

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21  
150

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Melissa Hill	MA	X2636
Tom Prikyal	JP	X5647

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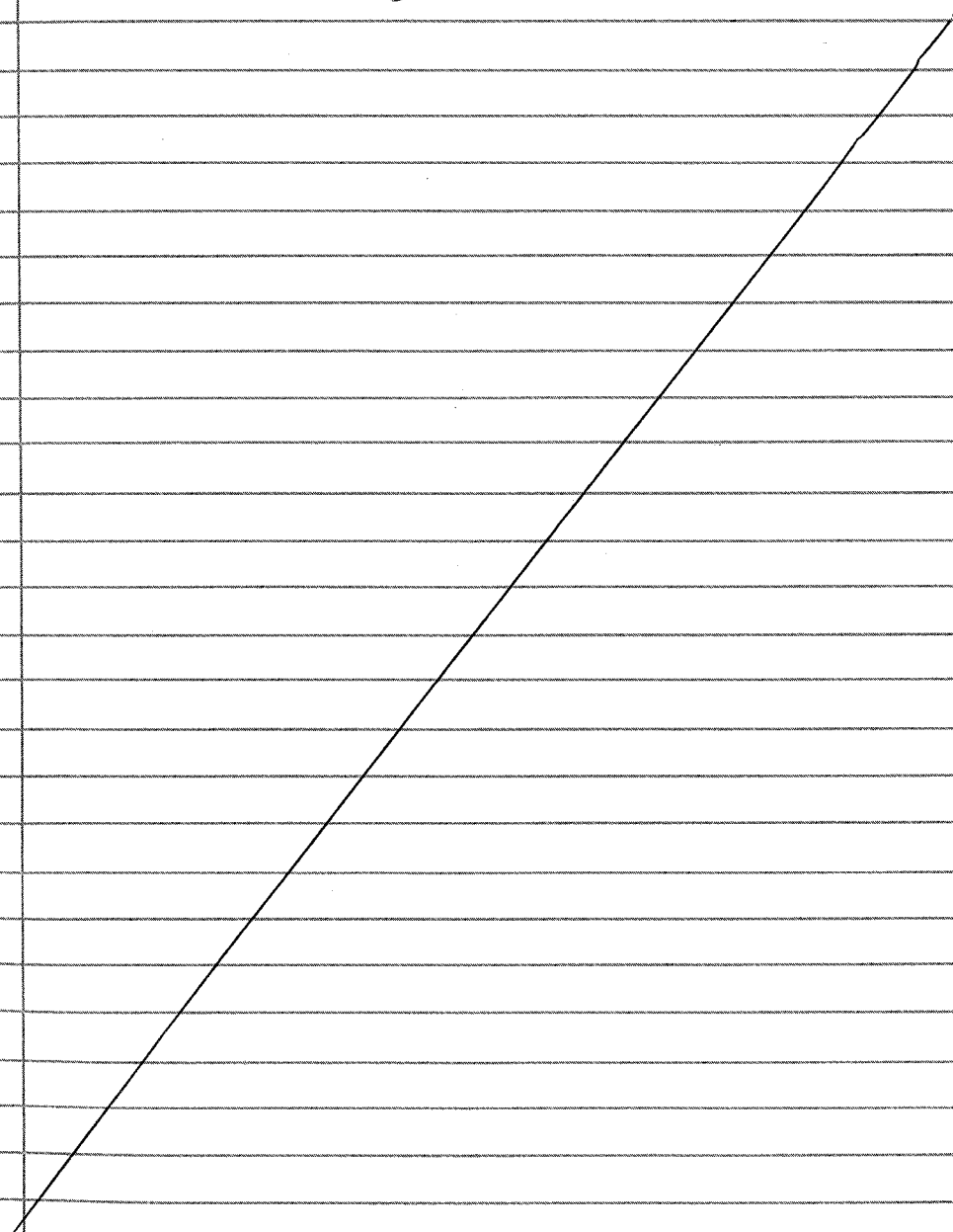
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12/21/99 This scientific notebook documents work performed on the Thermal Hydrology KTI Project (TEF). Specifically, this notebook documents work performed on the Drift Scale Heater Test investigating reflux/dripping/corrosion using the Topopah Spring Tuff.

Initial Entry on 12/21/99 by M. Hill



12/21/99


YMP-101-R3 06/06/94		YUCCA MOUNTAIN SITE CHARACTERIZATION PROJECT SAMPLE COLLECTION REPORT	
Date Collected: 12-7-99		Page 1 of 1	
Collector Name: DICK KOVACH			
Organization: LANL	Phone: (702) 295-6180		
Address: ESF PAD			
M/S 735			
Recipient Name: RON GREEN			
Organization: SOUTHWEST RESEARCH	Phone: (210) 522-5305		
Address: SOUTHWEST RESEARCH		Study Plan No.: NA	
BUILDING 51		Title: NA	
6220 CULDERA ROAD			
SAN ANTONIO, TX 78238			
Type of Sample: (circle one)	rock liquid	soil gas	alluvium caliche muck
Number of Samples Collected:		6 BARRELS/DRUMS	
Site Type: (circle one)	Trench	Outcrop	Surface
	Spring	Tunnel	Borehole
Other			
Describe Sample: <input type="checkbox"/> Q <input checked="" type="checkbox"/> Non-Q			
MUCK TRANSPORTED FROM NICHE #5, ECRB TO MUCK PILE VIA CONVEYOR BELT. TOPOL (Topopah-Spring lower lithophysal zone) TSW2			
Describe Site:			
MUCK PILE AT END OF CONVEYOR BELT, ESF PAD.			
Other Documentation (Field photos, maps, reports, etc.):			
SIX (6), 55 GALLON DRUMS/BARRELS			
Remarks:		SPL 00552582	
			
For SMF Use Only			
Record Verified By: Chris J. Hermes		Date: 12-7-99	
SMF TS			

Exhibit YAP-SII.4Q.1

01/17/00 JP

## Preparation of rock powders

Objective - prepare rock powders from samples of the Topopah Spring welded tuff (lower lithophysal unit) received from Yucca Mountain (TSW2)

Method - utilize mixer/mill to powder TSW2 samples.

## Equipment/Materials

- Spex mixer/mill Model 8000
- tungsten carbide vial
- plastic containers + bags
- rock hammer.

## Procedure

① Selected several hand sized specimens of TSW2 from the 55 gallon drums received from YM.

② Several <sup>less than 1.0</sup> centimeter sized pieces of rock were broken off from the hand sized specimens. These pieces were powdered using the Spex mixer/mill and tungsten carbide vial.



③ The resultant powder was placed in a snap cap plastic vial and labeled.

④ The remainder of the hand sized specimen was placed in a plastic bag and labeled.

Hand sized specimens were labeled as follows:

TSWZ-101

TSWZ-102

TSWZ-103

TSWZ-104

TSWZ-105

TSWZ-106

TSWZ-107

Powders in plastic container were labeled as follows:

TSWZ-101P

TSWZ-102P

TSWZ-103P

TSWZ-104P

TSWZ-105P

TSWZ-106P

TSWZ-107P

01/17/00 gp

Thin section preparation

Objective - have thin sections of TSWZ made for mineralogic examination

Method - prepare thin section blocks and send to Mineral Optics Laboratory.

Equipment/Materials

- trim saw
- sample bags.

Procedure

① Hand sized specimens TSWZ thru TSW7 were cut to a size (27x46mm) appropriate for standard thin section manufacture using a rock trim saw in Bldg 51.

② The blocks were placed in sample bags and sent to Mineral Optics lab in Winder VT.

- ③ The bags containing the blocks were labeled as follows:

TSW2-101TS  
 TSW2-102TS  
 TSW2-103TS  
 TSW2-104TS  
 TSW2-105TS  
 TSW2-106TS  
 TSW2-107TS

- ④ Samples sent to

Mineral Optics Laboratory  
 29 "A" Street  
 P.O. Box 828  
 Wilder, VT 05088

01/18/00 JP

### XRD Analysis of TSW2 rock powders.

Portions of TSW2 rock powder prepared on p 3-4 were placed in plastic vials. Samples were labeled and sent to Div for XRD analysis.

Standard procedures for XRD analysis were followed by Div to conduct the analyses. Analyses were performed using a Siemens D500 diffractometer. Data reduction and analysis was conducted using JADE XRD software.

01/27/00

## Bulk Chemical Analysis of TSW2 powders

Portions of the TSW2 powders prepared on p 3-4 were placed in plastic vials. The vials were labeled as shown below and sent to Texas Tech University, Geoscience Department for bulk rock chemical analysis by digestion + ICP analysis.

Sample	vial label
TSW2-101P	TS-101
TSW2-102P	TS-102
TSW2-103P	TS-103
TSW2-104P	TS-104
TSW2-105P	TS-105
TSW2-106P	TS-106
TSW2-107P	TS-107

To check quality of results 3 USGS rock standards were sent with the TSW2 powders. Portions of the USGS rock standards were placed in plastic vials and labeled as follows

USGS std	vial label
QLO-1 Quartz latite	TS-108
RGM-1 Glass Mt Rhyolite	TS-109
STM-1 Nepheline Syenite	TS-110

The rock standards were chosen based on their reported chemistry which should be similar to the TSW2 powders, especially for major elements such as Si, Al, Ca, Na, K, Fe, and Mg.

Samples were sent to the following address

Melanie Barnes  
Texas Tech University  
Geosciences  
Broadway and University  
Science Bldg, Rm 125  
Lubbock, TX 79409-1053

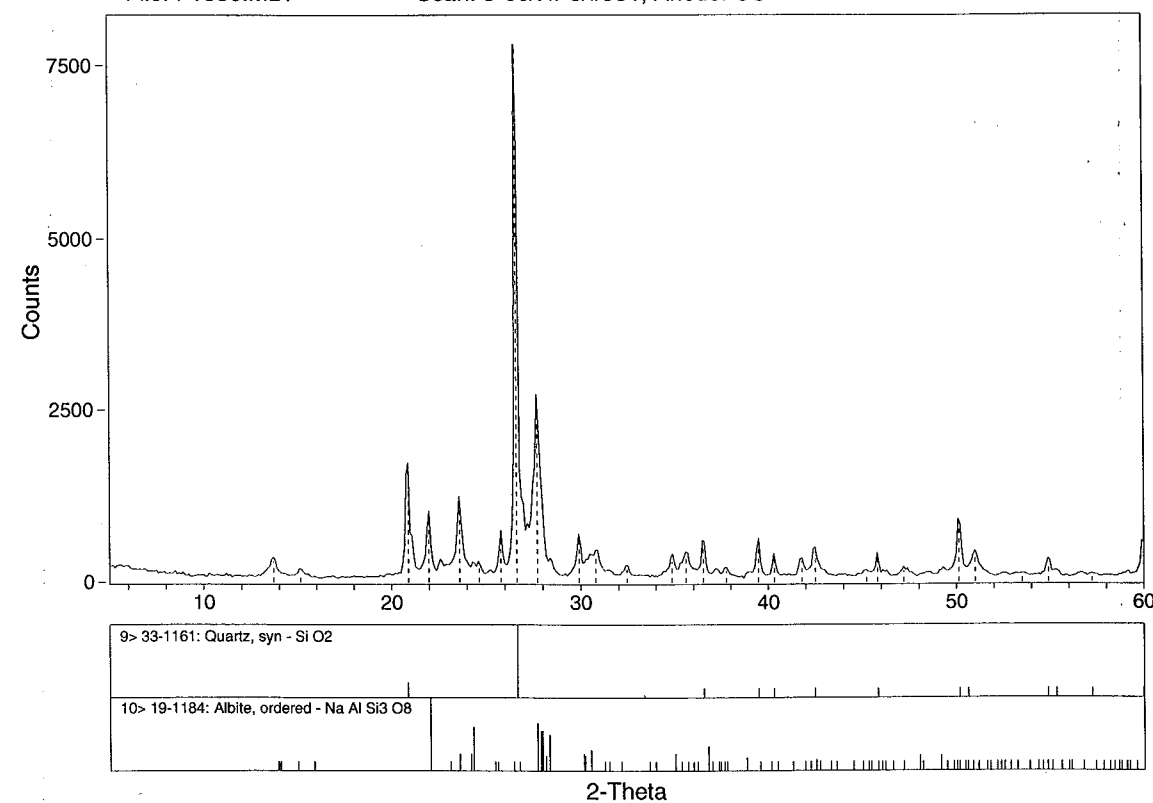
2/3/00 JP Results of XRD analysis of TSWZ powders.

XRD patterns for the TSWZ rock powder are shown below and on the following pages. Patterns indicate that quartz, cristobalite, and feldspars are the major mineral phases of TSWZ.

ID: TSW2-101P - 20.01402.661

File: F1330.MDI

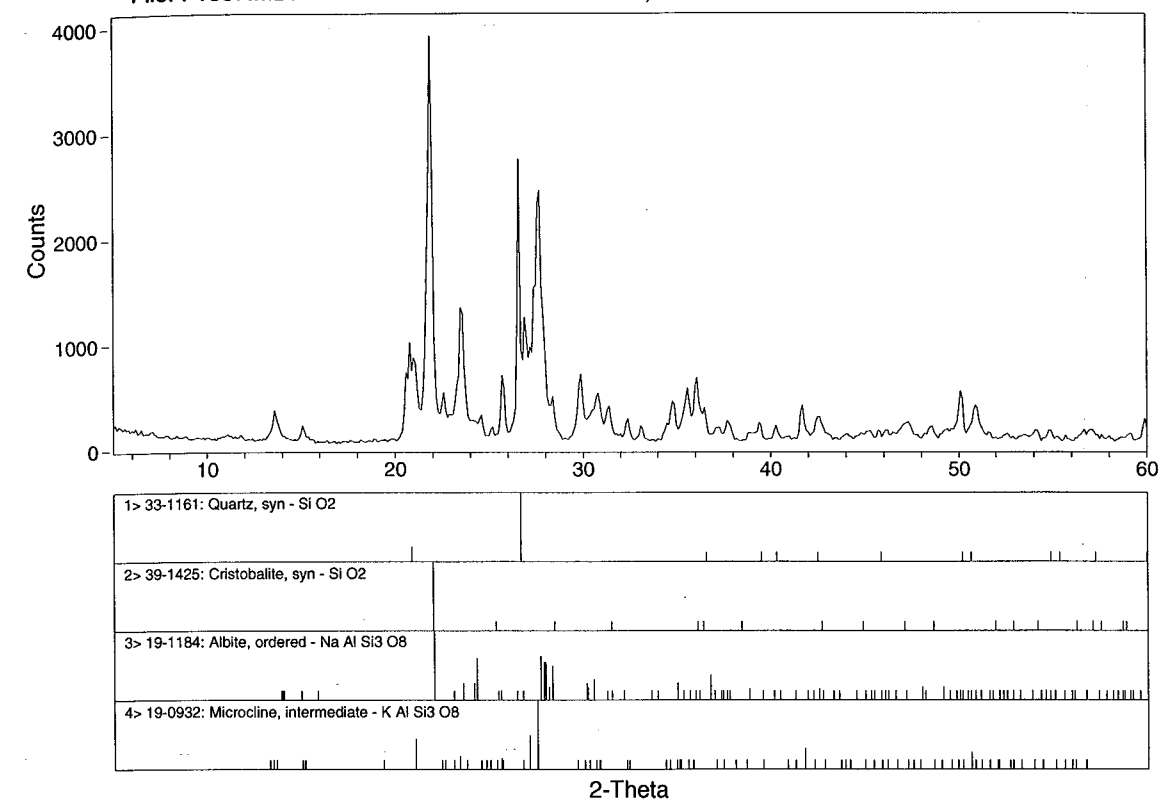
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ID: TSW2-102P - 20.01402.661

File: F1337.MDI

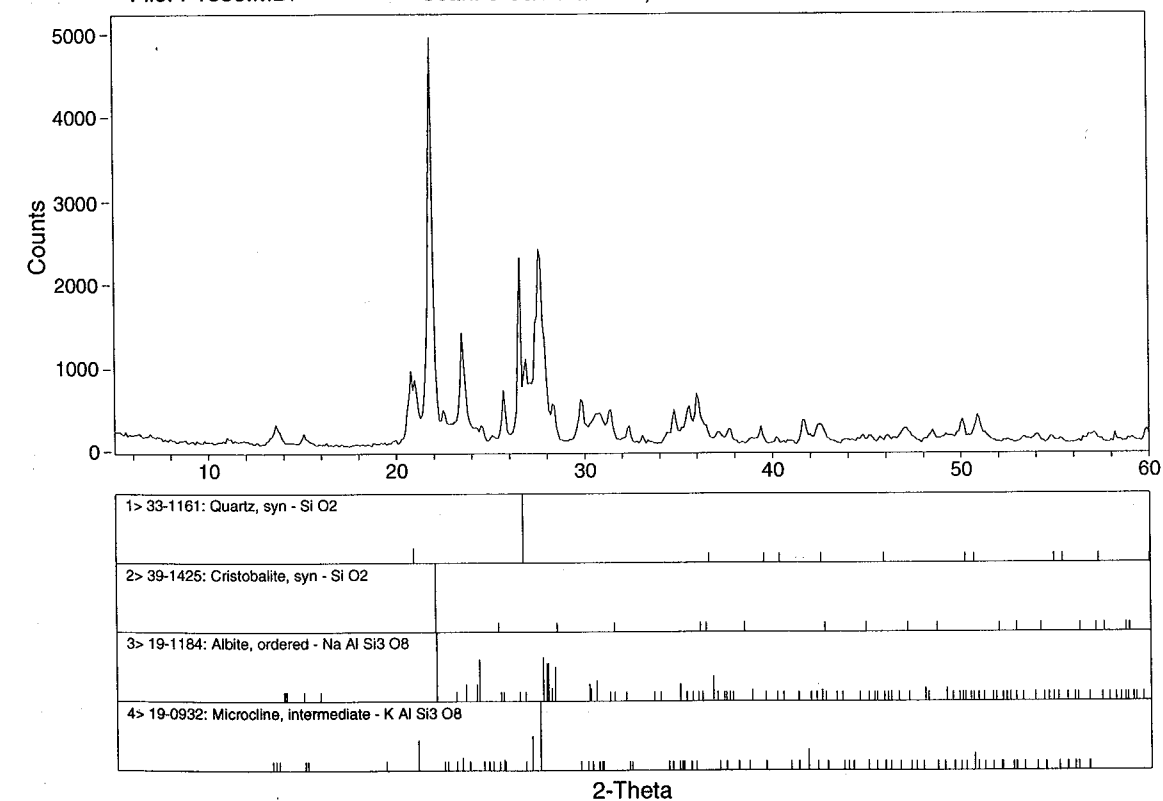
Scan: 5-60/.1/ 5/#551, Anode: CU



ID: TSW2-103P - 20.01402.661

File: F1335.MDI

Scan: 5-60/.1/ 5/#551, Anode: CU

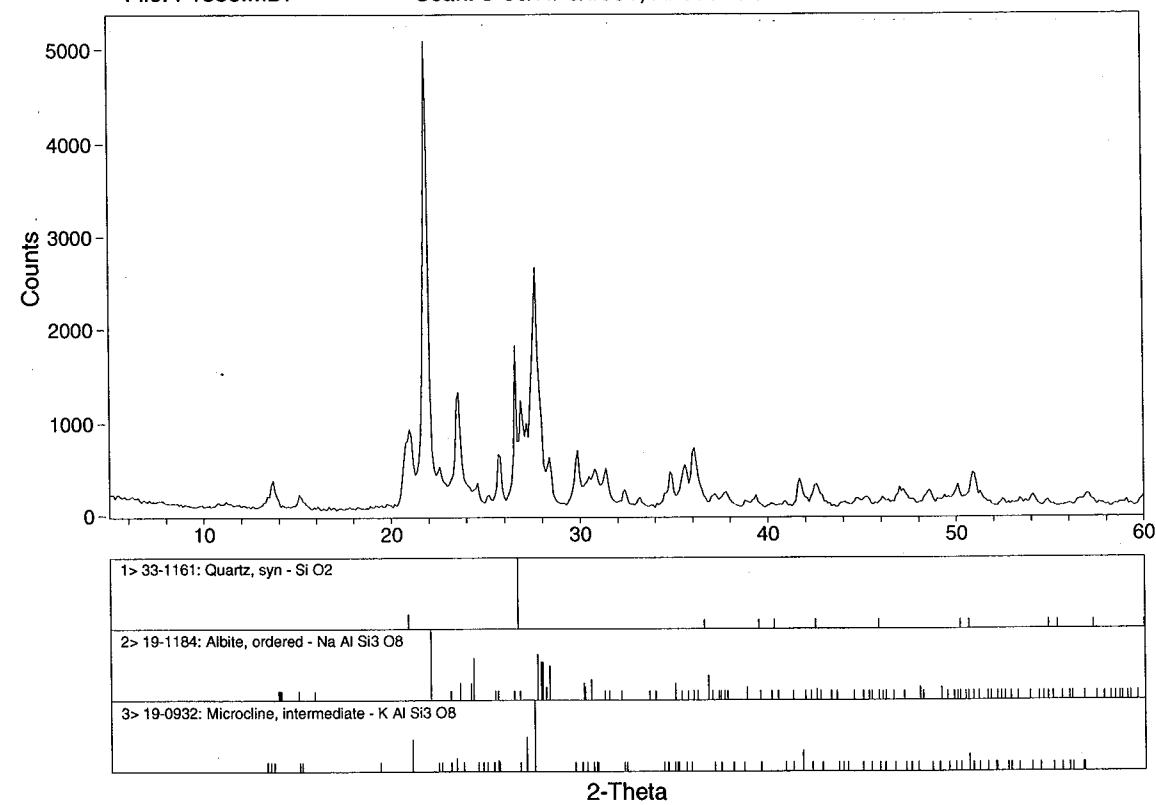




ID: TSW2-104P - 20.01402.661

File: F1333.MDI

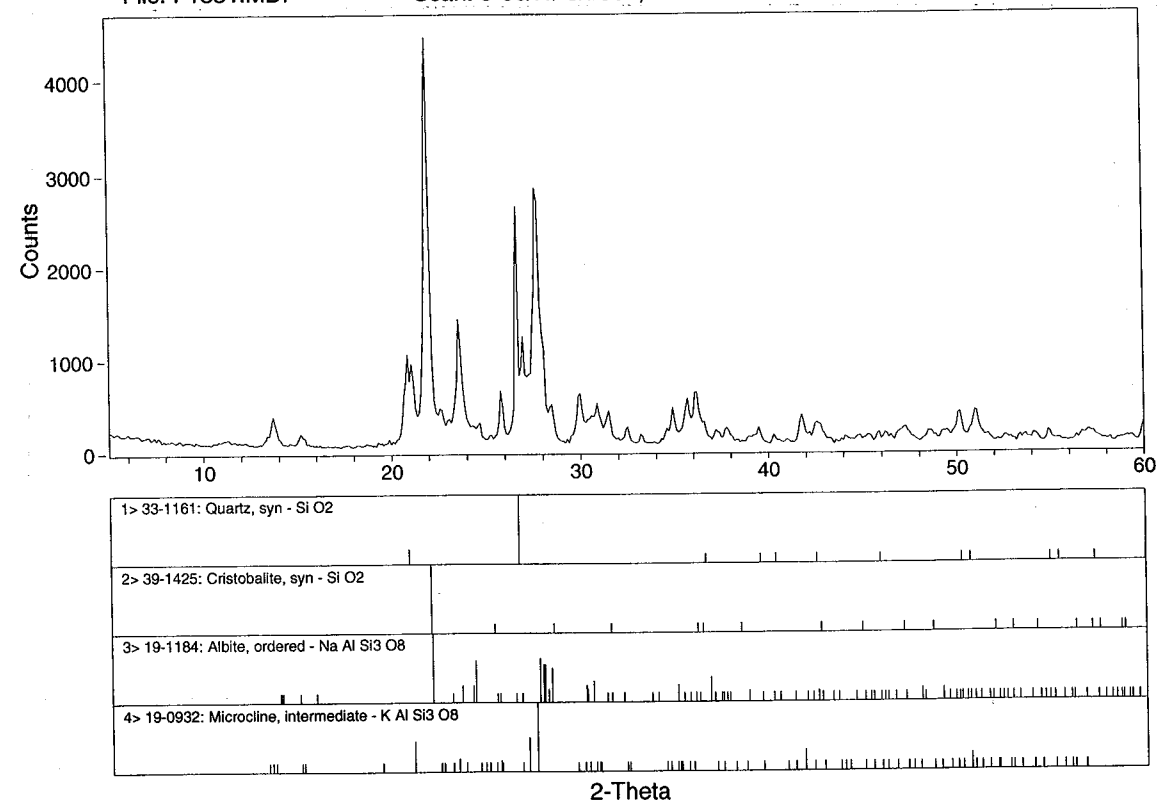
Scan: 5-60/.1/ 5/#551, Anode: CU



ID: TSW2-105P - 20.01402.661

File: F1331.MDI

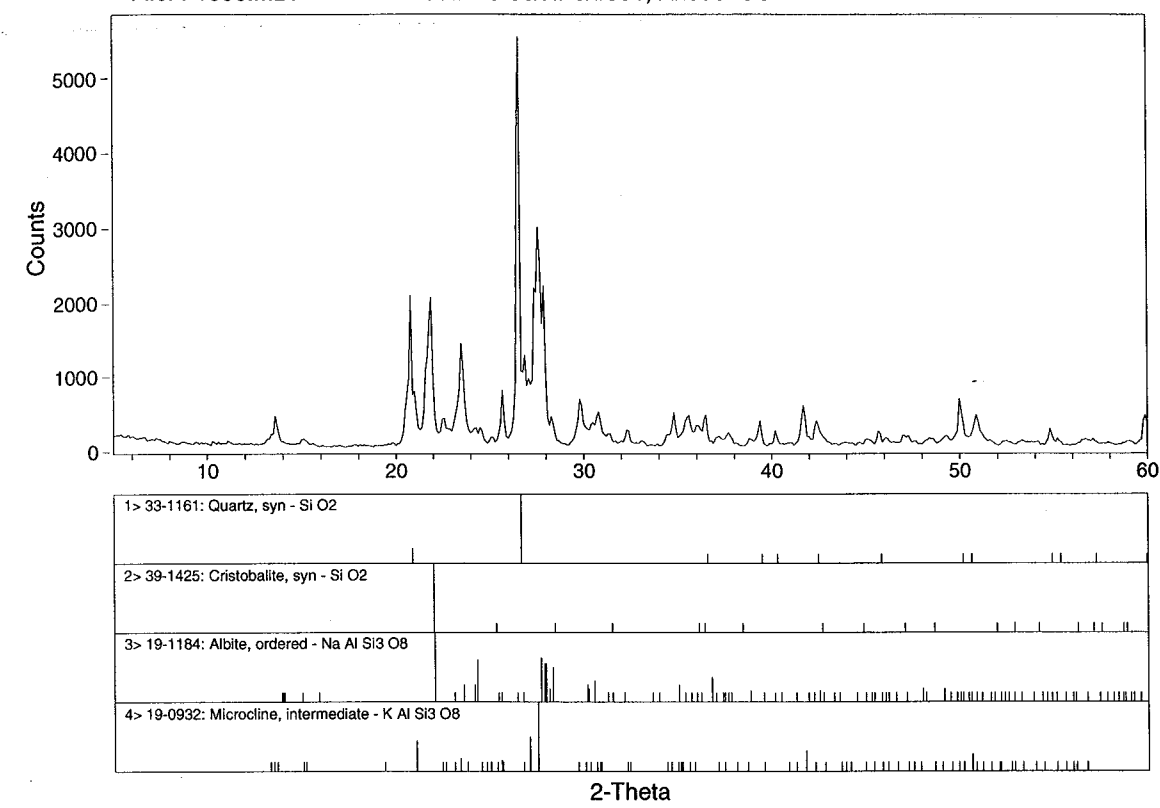
Scan: 5-60/.1/ 5/#551, Anode: CU



ID: TSW2-106P - 20.01402.661

File: F1336.MDI

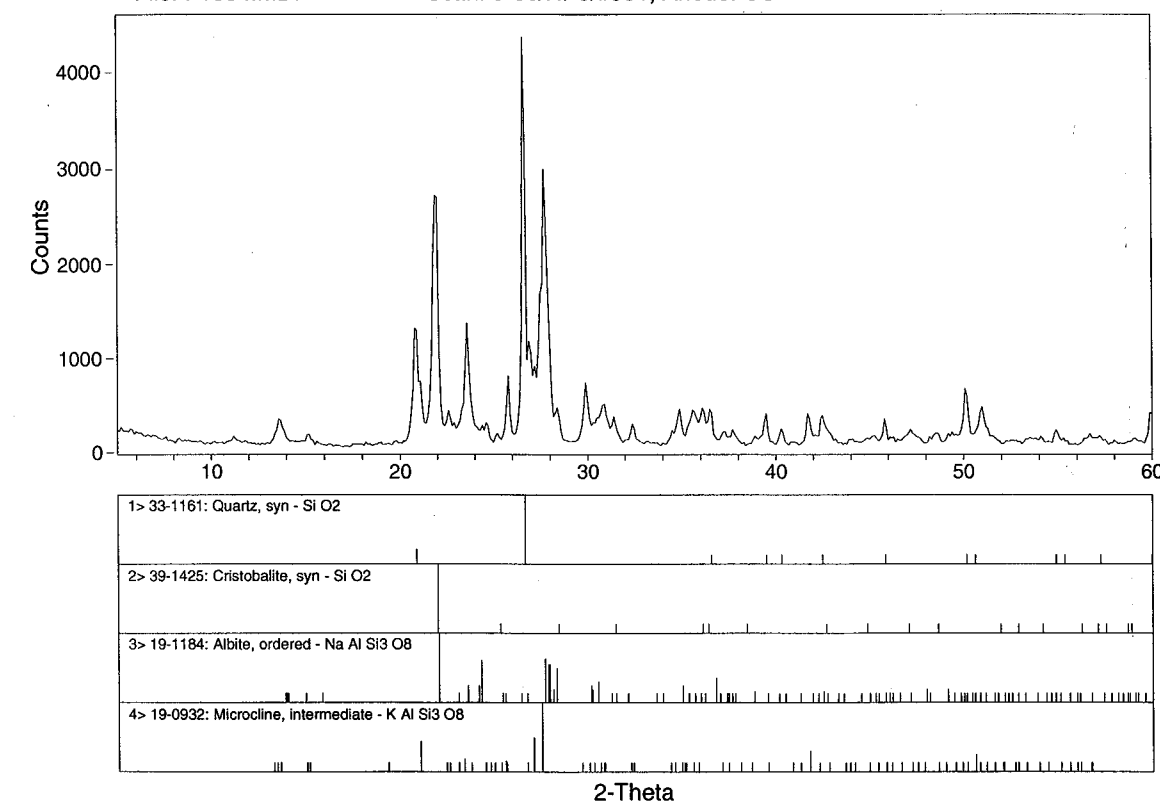
Scan: 5-60/.1/ 5/#551, Anode: CU



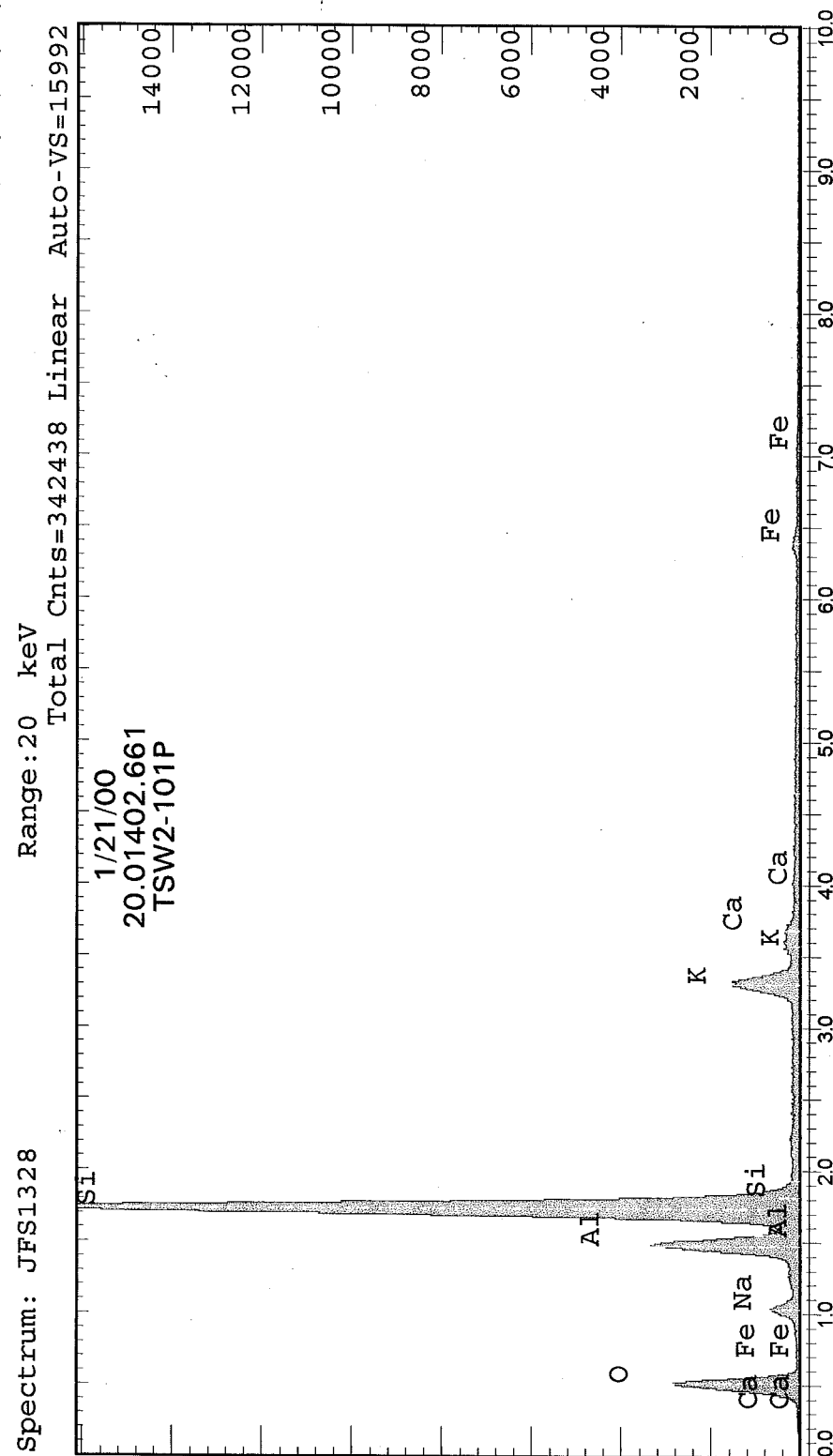
ID: TSW2-107P - 20.01402.661

File: F1334.MDI

Scan: 5-60/.1/ 5/#551, Anode: CU



2/3/00 JP An EDS analysis of TSW2-101P showing elemental composition is shown below.



2-28-00 JP

Separate TSW2 rock into size fractions

Obj - separate TSW2 into various size fractions for use in lab scale heater test #3.

Method - sieving and hand separation

Materials + Equipment

- brass sieves (1 in, 5/8", and 5/16")
- 30 gal polyethylene containers
- 50 gal steel drums.

Procedure

1) Use sieve and hand separation to separate TSW2 rock into 4 size fractions:

- > 1 inch (25 mm)
- < 1 inch (25 mm) to 5/8 inch (16 mm)
- < 5/8 inch (16 mm) to 5/16 inch (8 mm)
- < 5/16 inch (8 mm).

2) Place size fractions into separate containers; either 30 gal PE containers or 50 gal steel drums.

3/2/00 JP

Equilibrate water with TSW<sub>2</sub> rock

Obj - Equilibrate water with TSW<sub>2</sub> rock  
for use in lab scale heater  
test

Method - place TSW<sub>2</sub> rock in contact with  
D.I. water in 30 gal plastic  
containers and allow rock to  
react with water for ~2  
months.

#### Materials and Supplies

- > 1 inch TSW<sub>2</sub> size fraction
- < 5/16 inch TSW<sub>2</sub> size fraction
- 30 gal plastic containers
- D.I. water
- plastic bottles and containers  
as needed
- scale (Sartorius, 30 kg max)
- drying oven (Blue M)

#### Procedure

- ① Wash < 5/16 inch and > 1 inch TSW<sub>2</sub>  
size fractions several times with  
D.I. water to remove Li tag.
- ② After washing dry rock overnight  
in Blue M oven at ~85°C.

- (3) Into two 30 gal plastic containers the following quantities of rock were placed:

> 1" TSW2 - 30,000 gp 3/2/00  
 < 5/16" TSW2 - 11,800 g

- (4) Transfer 75 L of DI water to each of the two 30 gal plastic containers.

- (5) Lids were placed on containers to reduce evaporation.

- (6) Lids will be removed periodically to renew air in headspace of containers.

- (7) Allow water and rock to react for at least 2 months.

gp 4/2/00  
 4  
 5/2/00 gp

Thin sections of the Topopah Spring lower lithologized unit were received from Mineral Optics laboratory (see p 5-6).



4/17/00 JF

Results of chemical analyses of bulk  
rock powders of the Topopah Spring  
lower lithophysal unit are shown  
on the next page.

Analyses were performed by ICP at  
Texas Tech University.

Chemical Analyses of Topopah Spring rock powders.

	SiO <sub>2</sub>	TiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MnO	MgO	CaO	Na <sub>2</sub> O	K <sub>2</sub> O	P <sub>2</sub> O <sub>5</sub>	LOI	Total	Sr	Ba	Zr	Y	Sc	V	Cr	Ni	Cu	Zn	Nb	Be
TS-101	75.15	0.09	12.67	0.91	0.06	0.18	0.56	3.75	4.76	0.01	0.87	99.01	24	49	126	32.0	2.5	0	1	1	-1	54	19	3.0
TS-102	76.07	0.09	12.70	1.10	0.08	0.15	0.51	3.51	5.02	0.01	0.59	99.82	21	52	122	22.0	2.6	4	1	0	0	62	18	3.1
TS-103	76.84	0.09	12.56	0.91	0.06	0.15	0.52	3.79	4.76	0.01	0.71	100.40	22	46	126	28.8	2.5	2	1	7	0	41	18	2.8
TS-104	75.61	0.09	12.78	0.86	0.07	0.14	0.47	3.76	4.98	0.01	0.64	99.42	21	77	129	17.5	2.5	0	1	6	-3	34	18	2.7
TS-105	76.58	0.09	12.47	0.85	0.06	0.15	0.50	3.64	4.83	0.01	0.60	99.78	23	50	123	21.4	2.5	0	-1	3	-4	38	17	2.7
TS-106	75.85	0.09	12.51	0.84	0.06	0.16	0.48	3.78	4.88	0.01	0.58	99.24	21	50	122	30.2	2.5	1	3	2	-2	42	17	2.9
TS-107	76.23	0.09	12.61	0.89	0.06	0.17	0.50	3.56	4.99	0.01	1.60	100.70	23	47	123	25.1	2.6	-1	1	0	-2	46	17	2.8
TS-108	66.76	0.63	16.71	4.20	0.09	1.07	3.35	4.52	3.57	0.27	0.63	100.80	360	1365	190	22.1	8.6	45	1	2	29	66	11	1.6
TS-109	72.22	0.27	14.13	1.84	0.03	0.32	1.19	4.12	4.36	0.05	1.11	99.64	105	800	231	21.9	4.4	13	7	3	10	35	9	1.9
TS-110	59.13	0.13	18.65	5.12	0.21	0.15	1.15	8.96	4.31	0.16	1.55	99.52	705	570	1347	42.4	0.7	0	2	4	3	246	246	9.7

\*Oxides are in weight percent.

8/2/00

Remove Fines From Topopah Spring welded (TSW2) tuff.

Obj - Remove Fines From TSW2 which is to be used in lab scale heater test #3.

Method - wash and rinse with DI water.

Materials + Supplies

- sieved TSW2 rock
  - > 1 inch
  - < 1 inch to 5/8 inch
  - < 5/8 inch to 5/16 inch
- DI water
- plastic trays
- rubber containers.

Procedure

- ① Plastic 30 gal container and steel drums containing the TSW2 size fraction were filled with DI water.
- ② The rock was initially washed and rinsed inside the container. Fines were allowed to settle to the container bottom.
- ③ After initial rinsing + washing the rock was transferred to rubber containers filled with DI water. The rock was given a second rinsing.
- ④ The rock was then placed in plastic trays for drying and storage.

Note - the < 1 inch to 5/8 inch and < 5/8 inch to 5/16 inch size fractions were mixed together during the washing + rinsing procedure to create one size fraction consisting of < 1 inch to 5/16 inch rock pieces.

8/15/00 Lab Scale Heater Test using welded tuff  
from the Topopah Spring lower  
lithophysal unit

Obj - measure and monitor thermo-chemical  
parameters in a heated drift  
with water infiltrate from  
above.

Method - Construct and instrument a simulated  
adit surrounded by Topopah  
Spring welded tuff

### Materials and Equipment

- aluminum frame that will accommodate  
a 4' x 4' x 2' sample volume
- hexan plastic sheeting, 1/2" thick
- angle iron supports (8) - 2" angle  
1/4" wide x 52" long
- Fiberglass insulation
- 12" cartridge heater, 3/4" diameter,  
Omega brand
- Monostat cassette pump
- Hewlett Packard 34970A Data acquisition unit
- Data logger HP Backlink Software Version 1.1
- Staco Variable transformer, Type 3PN1510
- Viasalla HMP235 Temperature/Humidity probe
- Omega Type K precision thin wire thermocouples  
with 6' leads
- plastic tubing - various sizes as needed
- Teflon beakers - 600 ml, 250 ml, 100 ml

- Topopah Spring tuff lower lithophysal unit  
with fines removed.

> 1 inch size fraction

< 1 inch to 5/16 inch size fraction

- Plexiglass - 1/2" x 2' x 4' sheets (2)

- Spectra Mesh Nylon Filter (210  $\mu$ m mesh openings)

- 2' long x 6" diameter steel mesh

- Omega Type J thermocouple probe (12 inch long)

- porous ceramic tube (11 inch long)

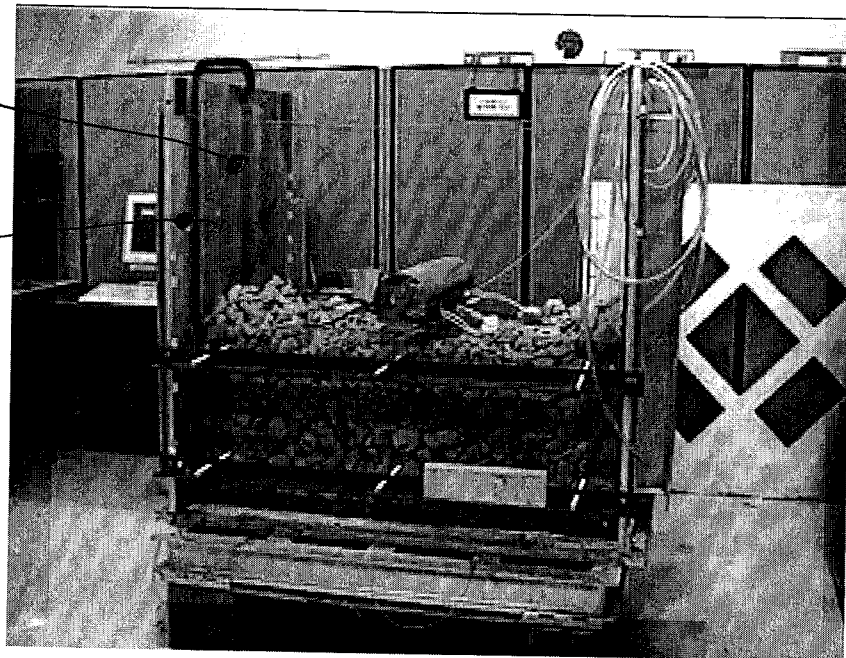
- Self Tune<sup>®</sup> plus temperature controller,  
Lone Control Corp.

8/15/00 <sup>JP</sup> Assembly

- ① A 4' high by 4' wide by ~3' deep aluminum frame shown in the photo below was used to construct the lab scale heater test. The front and back of the aluminum frame are open.

plexiglass sheet

Aluminum frame



8/16/00 <sup>JP</sup>

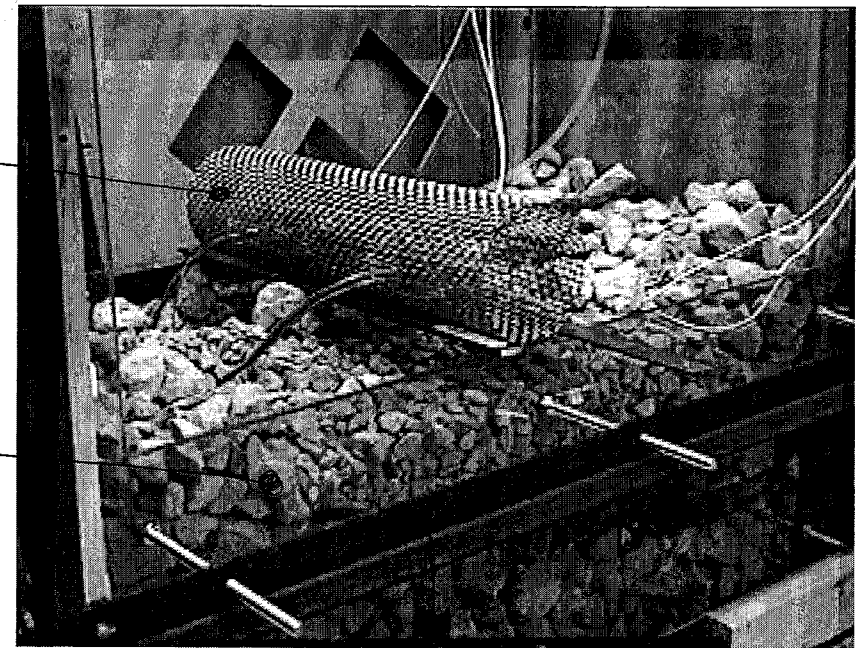
- ② 4' x 2' x 1/2" plexiglass sheets were secured to the inner right and left sides of the aluminum frame to provide a 2' wide spacing for emplacement of the tuff and simulated adit. Location of plexiglass sheets are shown in above photo.

8/17/00 <sup>JP</sup>

- ③ 4' x 21" x 1/2" sheets of laser plates were secured to the front and back of the aluminum frame using the angle iron supports. These sheets and the angle iron supports are shown in the photo below.

2' x 6" diameter  
steel  
enclosure

laser sheet



angle iron support.

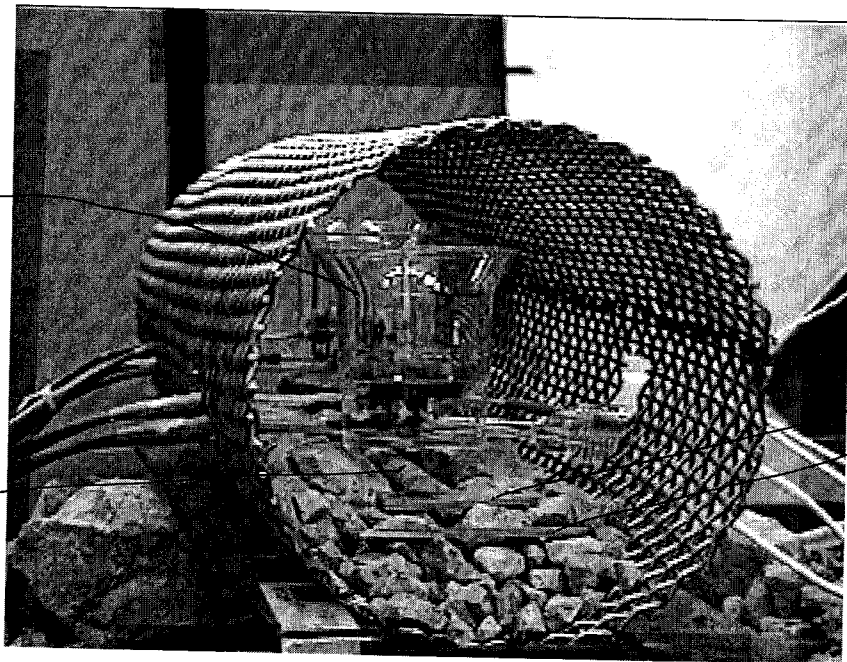
8/17/00 <sup>JP</sup>

- ④ The enclosure was filled with the >1 inch Topopah Spring tuff up to a level of about 16 inches. Above this level the enclosed area was filled progressively with >1 inch tuff at the right and left corners and with <1 inch to 5/16" tuff in the middle.



8/18/00 JP (5) The 2' long by 6" diameter steel enclosure (simulated adit) was placed at the center of the open sided alumin frame. See photo on previous page.

(6) The floor of the steel enclosure was filled with  $< 1$  inch to  $> 5 \frac{1}{16}$ " tuff. See photo below. A 12" cartridge heater with a  $\frac{3}{4}$ " diameter was placed to top of this floor.



lead wires for the cartridge heater were run under the steel enclosure and thru two holes drilled in the front lower corner. The heater was connected to a Staco variable transformer.

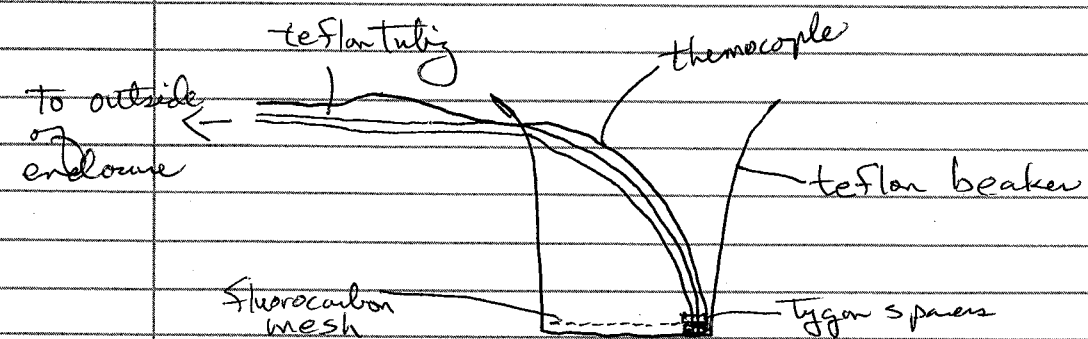
8/21/00 JP

(7) Instruments and metal samples were placed inside the adit to measure parameter related to corrosion. Two glass cups for measuring conductivity, Cl, corrosion potential, etc (and metal) coupons made of various materials were placed in the adit (see photo on previous page). The glass cups were designed to capture infiltrating water.

See scientific notebook # for a more detailed description of these instruments and samples.

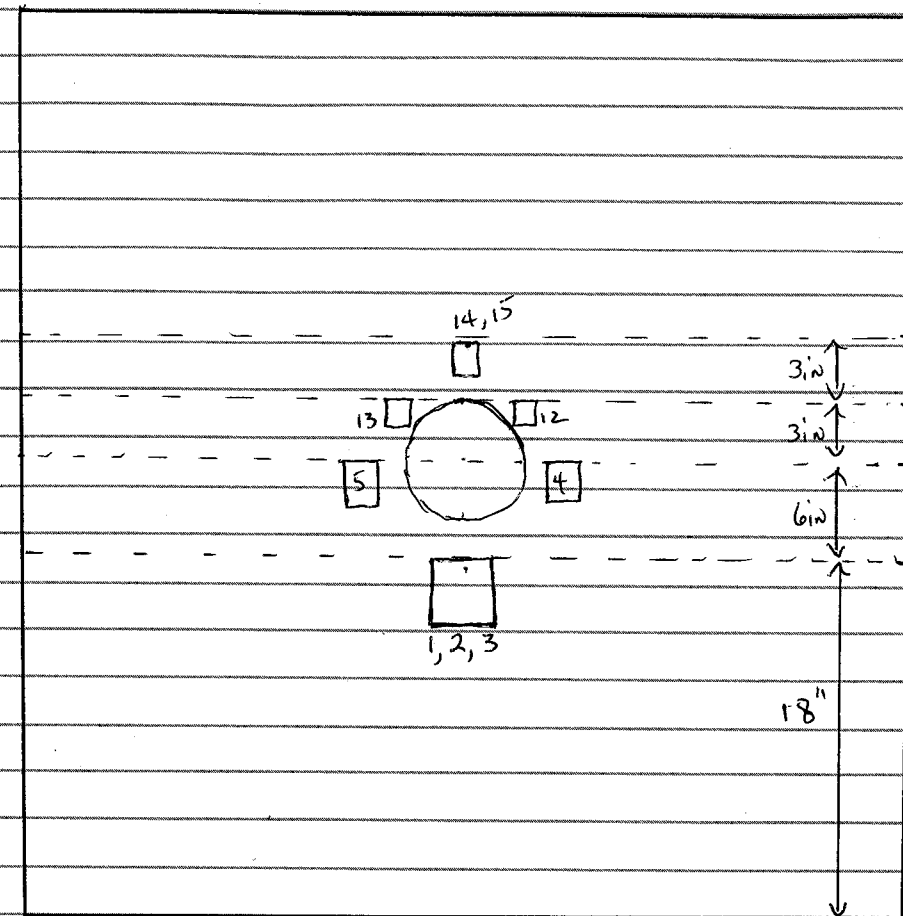
lead wires for the sample cups were run out the side of the steel enclosure and thru the holes drilled in the back lower corner.

- ⑧ Teflon sample cups were placed at various locations around the simulated adit as tuff was added to the enclosure. The purpose of the cups was to collect water as it infiltrated downward thru the tuff. The diagram below shows how each sampling cup was constructed.



Teflon tubing ran from the bottom of the sampling cup to outside of the enclosure. A syringe can be attached to the tubing outside the enclosure to draw out any water that collects in the bottom of the beaker.

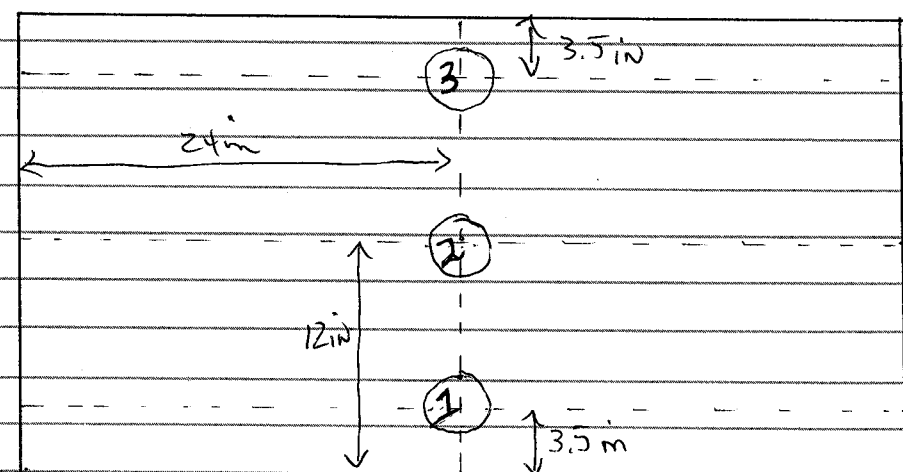
The location of each sampling cup are shown in the diagrams on the next pages.



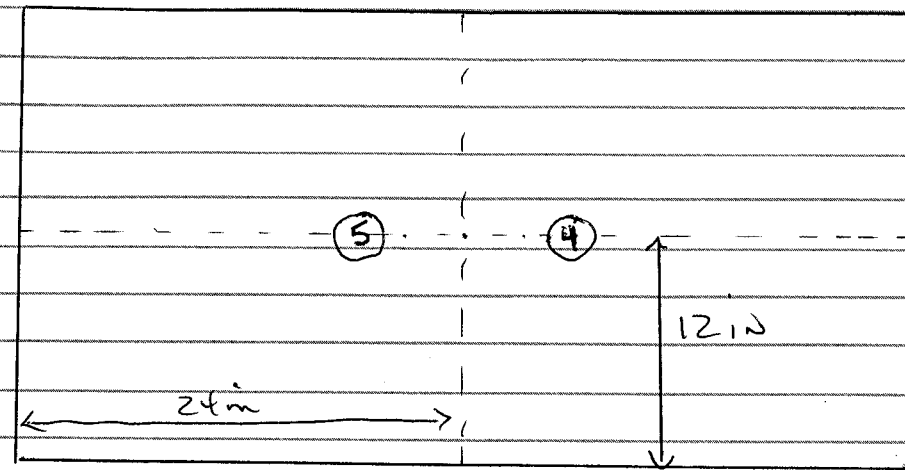
Front view

Plan views

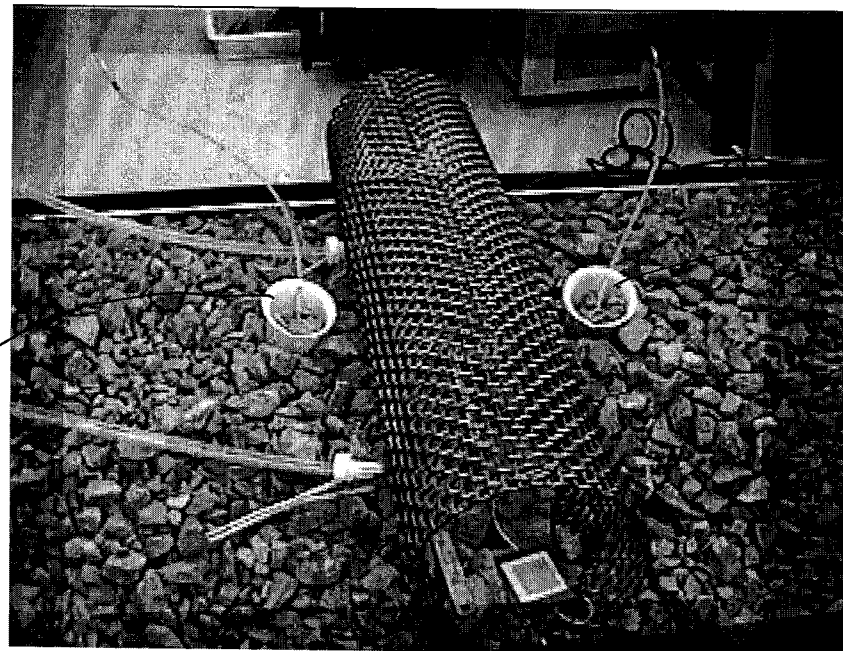
Sample cups 1, 2, + 3



Sample cups 4 and 5



Front

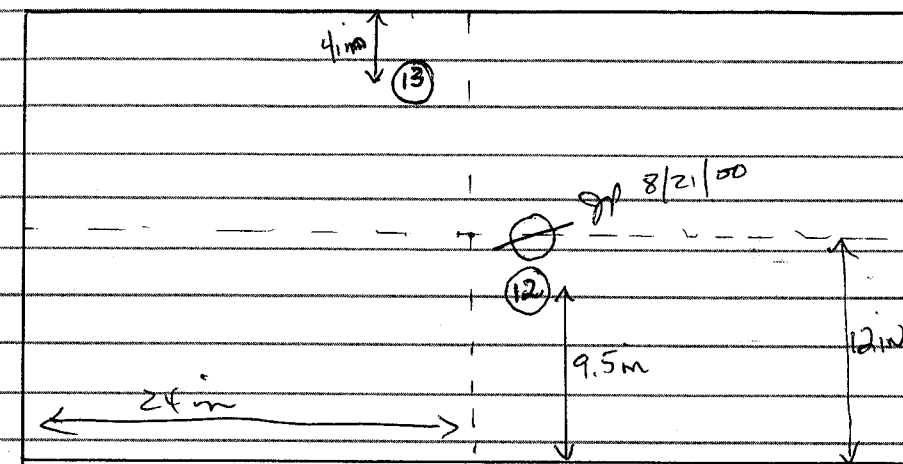


Sample cup 5

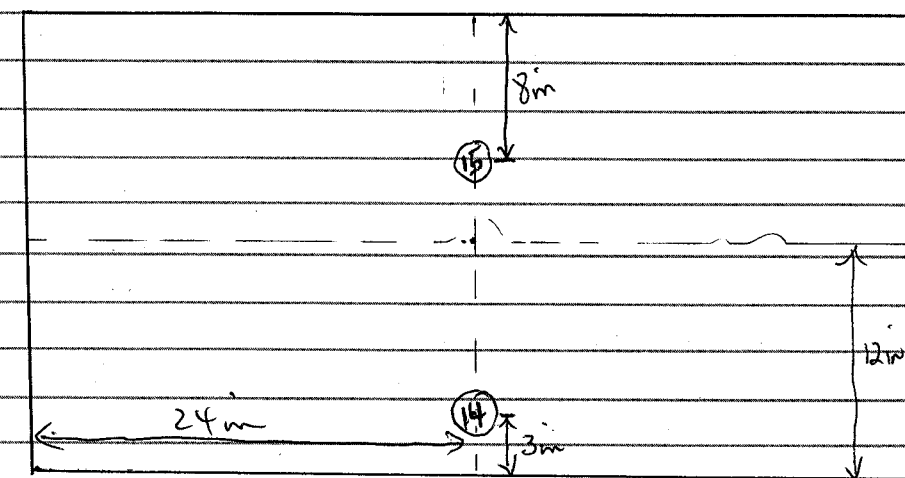
Back

Photo showing location of sample cups 4 and 5 in enclosure.

Sample cups 12+13



Sample cups 14+15



- (9) Thermocouples were placed in each supply cup and at other selected locations inside and outside the simulated adit. A total of 31 thermocouples (30 precision thin wire type K thermocouples and 1 type J thermocouple).

Before placement the thermocouples were calibrated using the procedure below.

### Thermocouple calibration

#### Equipment & materials

- cooking hot plate
- ice
- 1L glass beaker
- calibrated thermometer (S/N C96-833)
- HP34970A data acquisition unit
- Omega Precision fine wire thermocouples with 12 inch leads
  - Type K. (30)
- Omega 12" thermocouple probe
  - Type J. (1)

### Procedure

- (10) Thermocouples were placed in a beaker filled with ice along with the calibrated thermometer. The 30 Type K thin wire thermocouples were labeled T1-T30 and the Type J 12 inch thermocouple probe was labeled T34.

Temperatures are recorded below

Calibrated thermometer -0.7°C

T1 - 1.1	T16 - 0.7
T2 - 0.9	T17 - 0.7
T3 - 0.7	T18 - 0.9
T4 - 1.2	T19 - 0.5
T5 - 0.6	T20 - 0.4
T6 - 0.5	T21 - 0.9
T7 - 0.8	T22 - 0.8
T8 - 0.7	T23 - 1.1
T9 - 0.9	T24 - 1.1
T10 - 1.1	T25 - 0.7
T11 - 1.0	T26 - 0.8
T12 - 1.0	T27 - 0.6
T13 - 0.9	T28 - 0.5
T14 - 1.2	T29 - 0.6
T15 - 0.6	T30 - 0.6
	T34 - 0.9



- (9) Thermocouples were placed in a beaker of hot water ( $\sim 95^\circ\text{C}$ ) with the calibrated thermometer.

Temperatures are recorded below

Calibrated thermometer -  $95^\circ\text{C}$

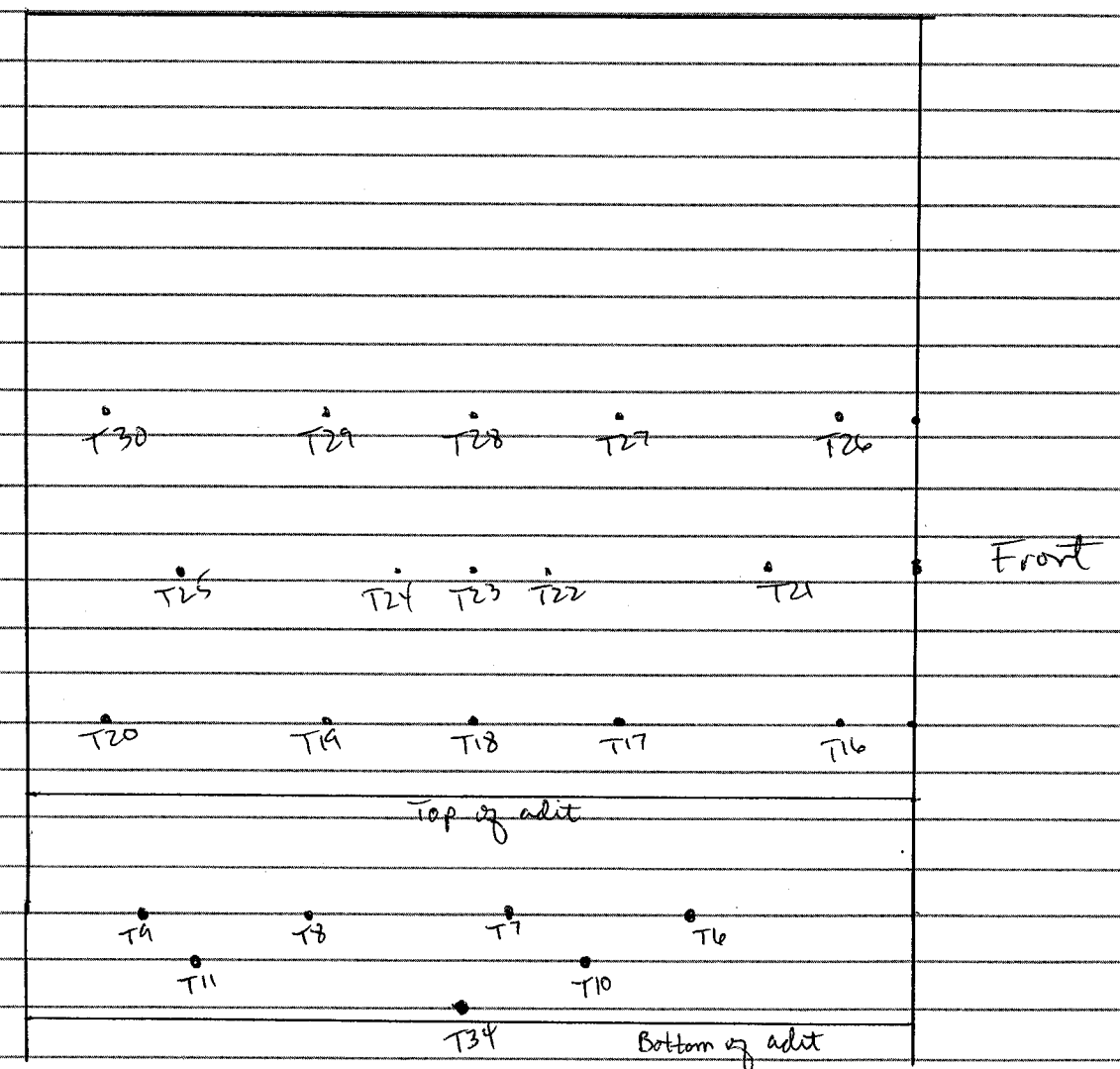
T1 - 95.3	T16 - 95.1
T2 - 94.7	T17 - 95.6
T3 - 94.7	T18 - 95.0
T4 - 95.0	T19 - 95.2
T5 - 95.3	T20 - 94.6
T6 - 95.6	T21 - 94.6
T7 - 94.7	T22 - 94.7
T8 - 94.8	T23 - 94.6
T9 - 94.8	T24 - 94.5
T10 - 94.7	T25 - 94.8
T11 - 95.1	T26 - 95.0
T12 - 95.1	T27 - 95.0
T13 - 95.0	T28 - 94.9
T14 - 94.9	T29 - 94.8
T15 - 94.8	T30 - 94.7
	T34 - 95.3

(10) 8/22/00 gpf

Diagram below shows the location of thermocouples inside the enclosure.

Thermocouples 1, 2, 3, 4, 5, 12, 13, 14, & 15 were placed inside the teflon sample cups. Their approximate locations are shown in the diagram on pp 31-33.

Back



Side View

Thermocouples 6, 7, 8, & 9 are inside the Tunnel at the approximate middle.

Thermocouples 10 and 11 were placed inside the glass sample cups.

Thermocouples T16-T30 are above the adit at the midpoint.

Thermocouple T34 is touching the heater inside the adit.

Thermocouple leads for type K thermocouples were run outside from the left front corner of the enclosure. The leads were attached to the HP 34970A data acquisition system.

The Type J probe was run thru a hole drilled in the front befan cover.

- (11) 4' x 27" x 1/2" sheets of lexan plastic were placed on the front & back of the alumin frame to complete the enclosure as tuff was added. These lexan sheets are also held in place by angle iron supports. The contact between the two befan sheets on the front & back of the assembly was sealed with RTV compound.

8/23/00 JP

- (12) The enclosure was filled to the top with tuff. The >1 inch tuff material was used along the right & left sides of the enclosure while the <1 inch to 5/16 inch tuff material was used around the simulated adit and the top middle of the enclosure. A photo of the filled enclosure is shown on the next page.



Temperature/  
humidity  
probe

- (13) A hole was drilled in the front lexan sheet into the adit to accommodate a Vaisala temperature/humidity probe. The probe can be seen in the photo above.

- (14) A porous ceramic tube was placed on the top of the tuff along the midline of the frame. The photo on the next page shows the ceramic tube after placement. The ceramic tube was connected to a Monostate cassette pump via plastic tubing.

- (15) The Monostate cassette pump was calibrated to determine the flow rate of water delivered from the porous ceramic tube at various Monostate settings.

The Monostate calibration is described on the following page.



porous  
ceramic  
tube

# Manostat calibration

## Equipment/materials

- Manostat cassette pump
- 11 inch porous ceramic tube
- DI water
- Volumetric flask.

① Plastic tubing was used to connect the cassette pump to a reservoir of DI water and the porous ceramic tube.

② The ceramic tube was placed in a Volumetric flask.

③ The cassette pump was run at different settings (10 to 3) for timed intervals and the water delivered by the ceramic tube was collected in the volumetric flask.

A table showing the volume of water collected at each setting and the time elapsed is shown on the following page.

Setting	Time Elapsed (min)	Volume Collected (ml)
10	15	16.5
9	15	16
8	15	15
7	15	13
6	15	10
5	15	8
4	15	7
3	36	14

The volume collected and time elapsed for each setting was used to calculate the volume of water that would be delivered by the tube in one day for the various settings. The formula is:

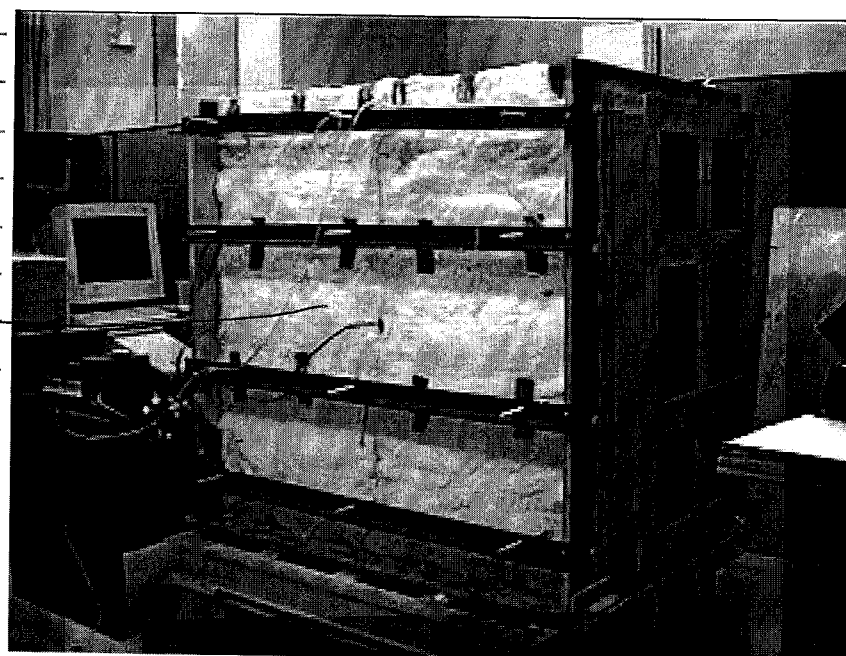
$$\frac{\text{Vol collected (ml)}}{\text{Time elapsed (min)}} \times \frac{1440 \text{ min}}{\text{day}} = \frac{\text{Vol (ml)}}{\text{day}}$$

Setting	Delivery Rate
10	1.58 L/day
9	1.54 L/day
8	1.44 L/day
7	1.25 L/day
6	.96 L/day
5	.76 L/day
4	.67 L/day
3	.56 L/day



8/24/00 JF

- (16) The front and back of the enclosure was covered with a layer of fiberglass insulation.



Fiberglass  
insulation

The top of the assembly was covered with a layer of  $\frac{1}{8}$ " neoprene followed by a layer of fiberglass insulation followed by a layer of  $\frac{1}{2}$ " lexan plastic.

- (17) The porous ceramic tube at the top of the assembly was connected to the Minotaur cartridge pump via plastic tubing. The Minotaur was also connected to a 20L reservoir of water that was equilibrated with the Tuff. (see p 17+18)

- (18) The 20L reservoir of water was spiked with NaCl to increase the Cl content of the water by 10 ppm.

- (19) Thermocouple #7 was attached to a temperature controller (Self Tune<sup>®</sup> Plus, Inco Controls Corp) to control temperature in the slit. The Staco variable transformer which controls the voltage supply to the cartridge heater was connected to the controller.



8/28/00 JP

## Start of lab scale heater test 3

① Voltage was applied to the cartridge heater by the variable transformer to heat up the system. The purpose is to raise the temperature of air in the adit to  $\sim 100^{\circ}\text{C}$  before adding water to the system.

② Power to the heater was controlled via a temperature controller with attachment to thermocouple #7 and the variable transformer. The variable transformer was adjusted until cycling was minimal. Minimal cycling of the heater allowed temperature inside the adit to remain relatively constant. Minimal cycling was achieved at a variable transformer setting of 60.

③ Thermocouple temperatures were monitored using HP Benchlink Datatogger software.

9/11/00 JP

④ Temperatures were monitored for about 2 weeks to ensure that system had reached a steady state before adding water. The temperatures of thermocouples after attaining steady state are shown below.

Data Grid

Channel	Reading	Time
<101>T1	26.07400 C	08:57:47.791 AM
<102>T2	26.68800 C	08:57:47.832 AM
<103>T3	26.12500 C	08:57:47.870 AM
<104>T4	50.56600 C	08:57:47.909 AM
<105>T5	48.10400 C	08:57:47.948 AM
<106>T6	71.26500 C	08:57:47.987 AM
<107>T7	96.34600 C	08:57:48.026 AM
<108>T8	101.4120 C	08:57:48.065 AM
<109>T9	70.47900 C	08:57:48.103 AM
<110>T10	109.9660 C	08:57:48.142 AM
<111>T11	74.57800 C	08:57:48.181 AM
<112>T12	94.18700 C	08:57:48.221 AM
<113>T13	71.90400 C	08:57:48.260 AM
<114>T14	85.00500 C	08:57:48.299 AM
<115>T15	105.4590 C	08:57:48.338 AM
<116>T16	76.15900 C	08:57:48.377 AM
<117>T17	101.9290 C	08:57:48.416 AM
<118>T18	111.4350 C	08:57:48.455 AM
<119>T19	104.1850 C	08:57:48.493 AM
<120>T20	79.58700 C	08:57:48.532 AM

Data Grid

Channel	Reading	Time
<121>T21	81.61800 C	08:59:48.559 AM
<122>T22	101.2280 C	08:59:48.598 AM
<123>T23	87.77900 C	08:59:48.637 AM
<124>T24	103.3190 C	08:59:48.676 AM
<125>T25	86.60800 C	08:59:48.715 AM
<126>T26	73.37600 C	08:59:48.754 AM
<127>T27	90.00600 C	08:59:48.793 AM
<128>T28	93.90100 C	08:59:48.832 AM
<129>T29	91.90200 C	08:59:48.870 AM
<130>T30	75.69300 C	08:59:48.909 AM
<134>T34	282.7360 C	08:59:48.948 AM

9/12/00

- ⑤ At 08:25 on 9/12/00 water addition at a rate of ~1L/day was started.

Temperature of thermocouples was acquired at 2 minute intervals

Temperature data is stored in the following files

test-3-1.csv (9/12/00)  
test-3-2.csv (9/13/00)  
test-3-3.csv (9/13-9/14)  
test-3-4.csv (9/14-9/15)

9/15/00 Temperature acquisition settings were changed to a 5 min interval.

test-3-5.csv (9/15)	test-3-22.csv (10/11-10/13)
test-3-6.csv (9/15)	test-3-23.csv (10/13-10/16)
test-3-7.csv (9/15-9/18)	test-3-24.csv (10/16-10/17)
test-3-8.csv (9/18-9/19)	test-3-25.csv (10/17-10/18)
test-3-9.csv (9/19-9/20)	test-3-26.csv (10/18-10/19)
test-3-10.csv (9/20-9/21)	test-3-27.csv (10/19-10/20)
test-3-11.csv (9/21-9/22)	test-3-28.csv (10/20-10/23)
test-3-12.csv (9/22-9/25)	test-3-29.csv (10/23-10/24)
test-3-13.csv (9/25-9/28)	test-3-30.csv (10/24-10/25)
test-3-14.csv (9/28-9/29)	test-3-31.csv (10/25-10/26)
test-3-15.csv (9/29-10/2)	test-3-32.csv (10/26-10/26)
test-3-16.csv (10/2-10/4)	test-3-33.csv (10/26-10/30)
test-3-17.csv (10/4-10/5)	test-3-34.csv (10/30-10/31)
test-3-18.csv (10/5-10/6)	test-3-35.csv (10/31-11/1)
test-3-19.csv (10/6-10/9)	test-3-36.csv (11/1-11/2)
test-3-20.csv (10/9-10/10)	test-3-37.csv (11/2-11/3)
test-3-21.csv (10/10-10/11)	Cont on page 53

Readings in mA from the temperature/humidity probe (Vaisala HMP235) were taken daily and converted to temperature and humidity using the formulas below.

$$\text{Relative Humidity} = \left( \frac{(\text{mA reading} - 4)}{16} \right) * 100$$

$$\text{Temperature} = \left( \left( \frac{(\text{mA reading} - 4)}{16} \right) * 200 \right) - 20$$

		mA readings			
		RH	Temp	RH	Temperature
9/12/00	1525	5.58	13.69	9.87%	101.1°C
9/13/00	1325	6.10	13.64	13.12%	100.5°C
9/14/00	0820	6.18	13.85	13.62%	103.1°C
9/15/00	0750	6.25	13.91	14.06%	103.9°C
9/18/00	0835	6.55	13.55	15.94%	99.4°C
9/19/00	0740	6.45	13.73	15.31%	101.6°C
9/20/00	0750	6.44	13.69	15.25%	101.1°C
9/21/00	0750	6.07	14.28	12.94%	108.5°C
9/22/00	0750	5.87	14.54	11.69%	111.7°C
9/25/00	0820	5.79	14.67	11.19%	113.4°C
9/28/00	0740	6.14	14.22	13.37%	107.7°C
9/29/00	0910	5.96	14.41	12.25%	110.1°C
10/2/00	0805	6.21	13.95	13.81%	104.4°C
10/4/00	0730	6.54	14.00	15.87%	105.0°C
10/5/00	0845	7.19	13.43	19.94%	97.8°C
10/6/00	0755	7.12	13.30	19.50%	96.2°C
10/6/00	1550	7.31	13.08	20.69%	93.5°C
10/9/00	0840	7.45	13.10	21.56%	93.75°C
10/10/00	0750	7.35	13.20	20.94%	95.0°C
10/11/00	0740	7.50	13.19	21.87%	94.9°C
10/11/00	1415	7.57	13.00	22.31%	92.5°C
10/13/00	0755	7.23	13.17	20.19%	94.6°C

9/29/00 JF

1430 hrs

Ceramic tube used to deliver water at top of apparatus was replaced. Then original tube appeared to be clogged.

10/13/00 JF

Temperature & Relative Humidity readings  
(continued from p 49)

		RH <sup>MA</sup>	Temp		RH	Temp
	10/13/00	1555	7.15		17.68%	95.4°C
	10/16/00	0840	6.76		17.25%	101.7°C
	10/17/00	0840	6.79		17.44%	101.1°C
	10/18/00	0730	6.74		17.12%	101.5°C
	10/19/00	0755	6.86		17.87%	100.0°C
	10/20/00	1100	6.84		17.75%	100.0°C
	10/23/00	0800	6.94		18.37%	100.4°C
	10/24/00	0900	6.86		17.87%	101.1°C
	10/25/00	0745	6.53		15.81%	105.1°C
	10/26/00	0900	6.50		15.62%	105.6°C
	10/26/00	1335	6.91		18.19%	100.1°C
MA	10/27/00	0845	6.86	MA	17.87%	99.1°C
	10/30/00	0745	6.97		18.56%	96.9°C
	10/31/00	0800	7.02		18.87%	98.7°C
	11/1/00	0815	7.02		18.87%	100.1°C
	11/1/00	1030	7.27		20.43%	97.2°C
	11/2/00	0730	7.29		20.56%	91.0°C
	11/3/00	1035	7.23		20.19%	91.6°C
	11/3/00	1340	7.41		21.31%	90.0°C
	11/6/00	0750	7.36		21.0%	90.0°C
	11/7/00	0800	7.75		23.44%	85.7°C
	11/7/00	1615	8.34		27.12%	81.0°C
	11/8/00	0745	8.46		27.87%	80.4°C
	11/9/00	0755	7.76		23.50%	84.0°C
	11/10/00	0730	7.71		23.19%	84.1°C
	11/10/00	1600	7.75		23.44%	84.0°C
	11/13/00	0725	8.16		26.0%	81.5°C
	11/13/00	1340	8.58		28.62%	78°C
	11/14/00	1015	8.27		26.69%	81.2°C
	11/15/00	0800	8.70		29.37%	82.4°C

(cont on p 56)

10/24/00 JP

1400 hrs The flow rate of water introduced to the system thru the ceramic tube was changed from ~1L/day to 2L/day

10/24/00 JP  
0920hrs

Controller that provides power to transformer was set from 106° (100°C) to 99°C. This will allow temperature in the adit to decrease by about 5°C.

11/1/00 JP  
0815hrs

Controller that provides power to transformer was set to 99°C.

\* Disregard above two lines, controller was not changed.

Instead the transformer setting was reduced from 56 to 52.

11/3/00 JP  
1240hrs

Transformer setting reduced from 52 to 50

11/6/00 JP

Cont From p 48

Temperature data is stored in following files

test-3-38.csv (11/3-11/6)	test-3-71.csv (10/3-10/4)
test-3-39.csv (11/6-11/7)	test-3-72.csv (10/4-10/5)
test-3-40.csv (11/7-11/8)	test-3-73.csv (10/5-10/8)
test-3-41.csv (11/8-11/9)	test-3-74.csv (10/8-10/9)
test-3-42.csv (11/9-11/10)	test-3-75.csv (10/9-10/10)
test-3-43.csv (11/10-11/13)	test-3-76.csv (01/10-01/11)
test-3-44.csv (11/13-11/14)	test-3-77.csv (01/11-01/12)
test-3-45.csv (11/14-11/15)	test-3-78.csv (01/12-01/15)
test-3-46.csv (11/15-11/16)	test-3-79.csv (01/15-01/16)
test-3-47.csv (11/16-11/19)	test-3-80.csv (01/16-01/17)
test-3-48.csv (11/19-11/22)	test-3-81.csv (01/17-01/18)
test-3-49.csv (11/22-11/27)	test-3-82.csv (01/18-01/19)
test-3-50.csv (11/27-11/28)	test-3-83.csv (01/19-01/22)
test-3-51.csv (11/28-11/29)	test-3-84.csv (01/22-01/23)
test-3-52.csv (11/29-11/30)	test-3-85.csv (01/23-01/24)
test-3-53.csv (11/30-12/1)	test-3-86.csv (01/24-01/25)
test-3-54.csv (12/1-12/4)	test-3-87.csv (01/25-01/26)
test-3-55.csv (12/4-12/5)	test-3-88.csv (01/26-01/29)
test-3-56.csv (12/5-12/6)	test-3-89.csv (01/29-01/30)
test-3-57.csv (12/6-12/7)	test-3-90.csv (01/30-01/31)
test-3-58.csv (12/7-12/8)	test-3-91.csv (01/31-02/01)
test-3-59.csv (12/8-12/11)	test-3-92.csv (02/01-02/02)
test-3-60.csv (12/11-12/12)	test-3-93.csv (02/02-02/05)
test-3-61.csv (12/12-12/13)	test-3-94.csv (02/05-02/06)
test-3-62.csv (12/13-12/14)	test-3-95.csv (02/06-02/07)
test-3-63.csv (12/14-12/18)	test-3-96.csv (02/07-02/08)
test-3-64.csv (12/18-12/19)	test-3-97.csv (02/08-02/09)
test-3-65.csv (12/19-12/20)	test-3-98.csv (02/09-02/12)
test-3-66.csv (12/20-12/21)	test-3-99.csv (02/12-02/13)
test-3-67.csv (12/21-12/22)	test-3-100.csv (02/13-02/13)
test-3-68.csv (12/22-12/28)	test-3-101.csv (02/14-02/15)
test-3-69.csv (12/28-01/02)	test-3-102.csv (02/15-02/16)
test-3-70.csv (01/02-01/03)	(cont on p 114)

11/6/00 JP  
0850 hr

Transducer setty reduced from 50 to 48.

11/7/00 JP  
1100 hr

To reduce water vapor loss from the top of the test cell a layer of sponge neoprene was placed over the rock and covered with a layer of  $\frac{1}{2}$ " Plexiglas. This created a better seal at the top boundary of the test cell. Fiberglass insulation followed by a layer of  $\frac{1}{2}$ " lexan plastic was then placed on top of the plexiglas layer.

11/8/00 JP  
1430 hr

The flow rate of water introduced at the top of the test cell was decreased from 2 L/day to 1 L/day

11/13 1300 hr JP

An additional layer of sponge neoprene was placed along the top margin of the apparatus to reduce heat and moisture loss. In addition 50 lb sacks of lead beads were placed on top of the cell to create a better seal between the neoprene and top cell margin.



11/15/00 JF

Temperature + Relative Humidity Readings  
(Continued from p 51)

		MA					
		RH	Temp			RH	Temp
	11/16/00	0535	8.80	12.31		30.0%	83.9°C
	11/16/00	1605	8.89	12.27		30.5%	83.4°C
MA	11/17/00	0900	8.94	12.19		30.87%	82.4°C
	11/17/00	1705	8.93	12.21		30.81%	82.6°C
	11/19/00	1030	9.01	12.16		31.31%	82.0°C
MA	11/20/00	0900	9.53	12.03		34.56%	80.4°C
	11/20/00	1705	9.43	12.04		33.94%	80.5°C
	11/22/00	0900	10.06	12.01		37.87%	80.1°C
MA	11/22/00	1705	9.45	12.11		34.06%	81.4°C
	11/27/00	0750	9.94	12.02		37.12%	80.2°C
	11/27/00	1500	10.46	11.79		40.4%	77.4°C
	11/28/00	0745	10.46	11.70		40.4%	76.2°C
	11/29/00	0715	11.44	11.37		46.5%	72.1°C
	11/30/00	0735	11.53	11.28		47.06%	71.0°C
	12/1/00	0805	12.42	11.06		52.6%	68.2°C
	12/4/00	0915	13.02	10.39		56.37%	59.9°C
	12/5/00	0755	12.75	10.98		54.69%	67.2°C
	12/6/00	0700	12.80	11.16		55.00%	69.5°C
	12/7/00	1000	11.46	11.33		46.62%	71.6°C
	12/8/00	0715	11.34	11.18		45.87%	69.7°C
	12/11/00	0830	10.40	11.41		40.00%	71.6°C
	12/12/00	0820	10.75	11.25		42.19%	70.6°C
	12/13/00	0935	10.52	11.35		40.75%	71.9°C
	12/14/00	0745	11.58	11.01		47.37%	67.6°C
	12/18/00	0801	15.50	10.04		71.87%	55.5°C
	12/19/00	0825	17.01	9.72		81.31%	51.5°C
	12/20/00	0800	17.73	9.69		85.81%	51.1°C
	12/21/00	0750	19.40	9.52		96.25%	49.0°C

MH  
MH  
MH

		RH	MA	Temp		RH	Temp
	12/22/00	0800	19.40	9.53		96.25%	49.1°C
	12/27/00	0945	18.54	9.50		90.87%	49.4°C
	12/28/00	0920	17.13	9.96		82.06%	54.5°C
	12/29/00	0940	17.26	10.09		82.87%	56.1°C
	01/02/01	0830	15.71	10.21		73.19%	57.6°C
	01/03/01	0745	16.06	10.31		75.37%	58.9°C
	01/04/01	0820	17.95	10.10		87.19%	56.2°C
	01/05/01	0805	18.14	9.82		88.37%	52.7°C
	01/08/01	0735	18.46	9.74		90.37%	51.7°C
	01/09/01	0950	18.56	9.76		91.00%	52°C
	01/10/01	0835	18.40	9.71		90.00%	51.3°C
	01/11/01	1000	17.99	9.80		87.44%	52.5°C
	01/12/01	0830	18.41	9.77		90.06%	52.1°C
	01/15/01	0735	15.31	10.01		70.69%	55.1°C
	01/16/01	0945	17.01	9.93		81.31%	54.1°C
	01/17/01	0945	15.55	9.97		72.15%	54.6°C
	01/18/01	0815	18.55	9.74		90.94%	51.7°C
	01/19/01	0955	19.10	9.68		94.37%	51.0°C
	01/22/01	0725	18.76	9.75		92.25%	51.8°C
	01/23/01	0940	18.77	9.78		92.31%	52.2°C
	01/24/01	0940	18.87	9.73		92.94%	51.6°C
	01/25/01	0720	18.38	9.95		89.88%	54.3°C
	01/26/01	0940	17.98	10.17		87.74%	57.1°C
	01/29/01	0720	18.18	10.13		88.62%	56.6°C
	01/30/01	0945	18.67	10.23		87.94%	52.8°C
	01/31/01	0935	17.76	10.25		86.0%	58.1°C
	02/01/01	0730	18.50	10.16		90.62%	57.0°C
	02/02/01	0730	18.04	10.19		87.75%	57.3°C
	02/05/01	0715	18.11	10.17		88.19%	57.1°C
	02/06/01	0930	16.90	10.49		80.62%	61.1°C
	02/07/01	0935	17.38	10.62		83.62%	62.7°C
	02/08/01	0900	16.54	10.96		78.37%	67.0°C
	02/09/01	0945	15.95	11.10		74.69%	68.7°C
	02/12/01	0715	16.90	11.02		80.62%	67.7°C

(Temp + RH Readings continued on p 112)

12/5/00 JF  
1300 hrs

Collection of water equilibrated with the Topopah Spring lower Tertiary rock and water equilibrated with Topopah Spring lower Tertiary rock and spiked with 10 ppm Cl<sup>-</sup>. These waters were collected for chemical analysis.

Approximately 125 ml of equilibrated Topopah Spring lower Tertiary water was placed in a 125 ml plastic container and labeled TSEQ.

Approximately 125 ml of equilibrated Topopah Spring lower Tertiary rock water spiked with 10 ppm Cl<sup>-</sup> was placed in a 125 ml plastic container and labeled TSEQCL.

12/5/00 JF  
1330 hrs

pH of water collected on previous page -  
TSEQ + TSEQCL

Obj - measure pH of water samples

#### Equipment & Supplies

- Orion pH electrode
- ATC probe
- Orion pH/mV/ISE meter Model 920A
- plastic sample cups
- Fisher buffer solution
  - pH 4 (lot 994523-24)
  - pH 10 (lot 987055-24)
- Corning stirrer
- Mini Teflon stir bar.

#### Procedure

- ① Calibrate pH electrode using Orion pH/mV/ISE meter and buffer solution.
- ② Place ~10 ml of sample (TSEQ or TSEQCL) in sample cups with mini stir bar.
- ③ Measure pH with calibrated pH electrode.
- ④ Discard pH sample.

## Results

Sample	pH
TSE Q	6.88
TSE QCL	6.83

12/5/00 JF  
1340 hrs.

Alkalinity of water collected on p58-  
TSE Q + TSE QCL

Obj - determine alkalinity of water sample.

Method - titration with standard acid

## Equipment &amp; Supplies

- Mettler D21 Titration
- Mettler Toledo PG111-SC pH electrode
- Plastic sample cups
- 0.1N HCl standard soln

Prepared by adding 8.2 ml concentrated HCl  
to 1L DI water in a 1L Volumetric  
flask.

- Fisher buffer solution  
pH 4 lot (994523-24)  
pH 10 lot (987055-24)

## Procedure -

- ① Calibrate pH electrode using Fisher buffer solution.
- ② Transfer 40 ml of water sample (TSE Q or TSE QCL) to plastic sample cups.

③ Set up Mettler DL21 titrator to perform a p- and/or m-value end point titration using 0.1 N HCl standard acid.

④ Attach sample cup to automatic titrator and perform titration.

⑤ Results of the titration to determine alkalinity are shown on the following pages.

# Sample TSEQ

===== METTLER DL21 Titrator date 05-12-00 =====

TITRATION	1	operator
METHOD	4011	2 04-12-0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.125	
pH0 pH	7.298	
SLOPE mV/pH	-57.4	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.43739	
M-VALUE mmol/L	1.0935	

$$\text{Alkalinity} = \frac{\text{ml std acid used} \times \text{normality std acid} \times 50000}{\text{ml sample}} =$$

$$= \frac{.43739 \times 0.1 \times 50000}{40} = 54.67 \text{ mg CaCO}_3/\text{L}$$

Sample TSEQCL

===== METTLER DL21 Titrator date 05-12 00 =====

TITRATION 2 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 6.847

pH0 pH 7.298

SLOPE mV/pH -57.4

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .41404

M-VALUE mmol/L  
1.0351

$$\text{Alkalinity} = \frac{.41404 \times .1 \times 50000}{40} = 51.75 \text{ mg CaCO}_3/\text{L}$$

12/5/00 JF  
1400 hr

Sample TSEQ and TSEQCL remaining after pH and alkalinity analysis were split into two equal aliquots and placed in 60 ml PP containers. One of the aliquots of each sample was acidified to pH < 2 with conc HNO<sub>3</sub>. These acidified samples were labeled TSEQ-A and TSEQCL-A. The unacidified samples kept their original labels of TSEQ and TSEQCL. These samples were refrigerated for later analysis.



12/6/00 JF  
1100 hrs.

The ceramic tube that delivers water to the system was repositioned. The tube was lowered about 2 inches into the cracked tuff.

12/14/00 JF  
0830 hrs.

Transducer was set to 30. This will reduce temperature of cartridge heater and thus the temperature in the adit. This was done to try to induce higher humidity and dripping in the adit.

12/19/00 JF 0900 hrs

Collection of water from sampling cups.  
See p 30-34 to see location of each cup.

Sample cup no	Approximate amount of water collected	Temp of water collected
1	—	
2	—	
3	—	
4	40 ml	30.7°C
5	40 ml	30.4°C
12	—	
13	—	
14	15 ml	37.2°C
15	20 ml	40.5°C

Water collected were placed in 60 ml PP bottles and labeled as follows.

Sample cup no	Label
4	S4
5	S5
14	S14
15	S15

12/19/00 JP 0930 hr

The pH of samples collected on previous page  
were measured using procedure on p 59.

## Results

Sample	pH
S4 JP 12/19/00	7.96
S5 JP 12/19/00	7.01
S14 JP 12/19/00	7.80
S15 JP 12/19/00	7.78

12/19/00 JP 1000 hr

The alkalinity of sample S5 was determined  
using procedure on p 61.

Results are shown below

----- METTLER DL21 Titrator date 19-12-00 -----

TITRATION	2	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	Sample S5
const	25.000	
START pH	7.856	
pH0 pH	7.281	
SLOPE mV/pH	-57.4	40 ml sample volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.85573	
M-VALUE mmol/L	2.1393	

$$\text{Alkalinity} = \frac{.85573 \times 1 \times 50000}{40} = 106.97 \text{ mg CaCO}_3/\text{L}$$

12/19/00 JF 1100hr.

After pH and alkalinity measurements samples S4, S5, S14, and S15 were split into two equal aliquots and placed in 30 ml PP bottles. One aliquot of each sample was acidified to pH < 2 with  $\text{HNO}_3$ . These acidified samples were labeled as follows:

S4-A

S5-A

S14-A

S15-A

The unacidified samples kept their original labels. All samples were refrigerated for later analysis.



12/19/00 gp 0330 hr

The following set of pictures show the progress of the wetty front in the lab scale heater tank. Digital images were taken of the front of the apparatus with the insulation removed to show the boundary of the wetty front with respect to the side. Pictures were taken on 12/5/00, 12/13/00, and 12/19/00.

Picture taken on 12/5/00 at 0900 hr.



Picture taken on 12/5/00 at 0900 hr.





Pictures taken on 12/13/00 at 11:15 hr.

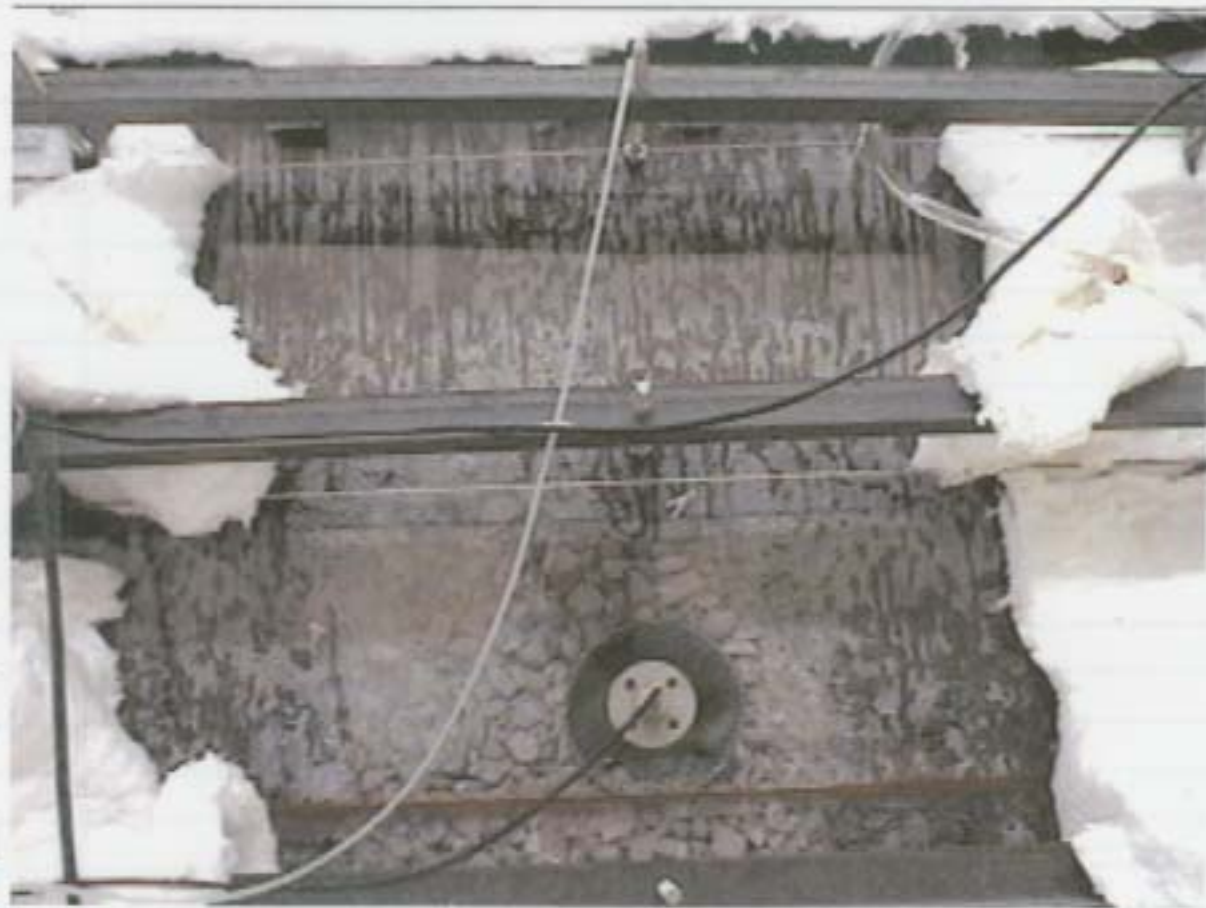


Pictures taken on 12/13/00 at 11:15 hr.

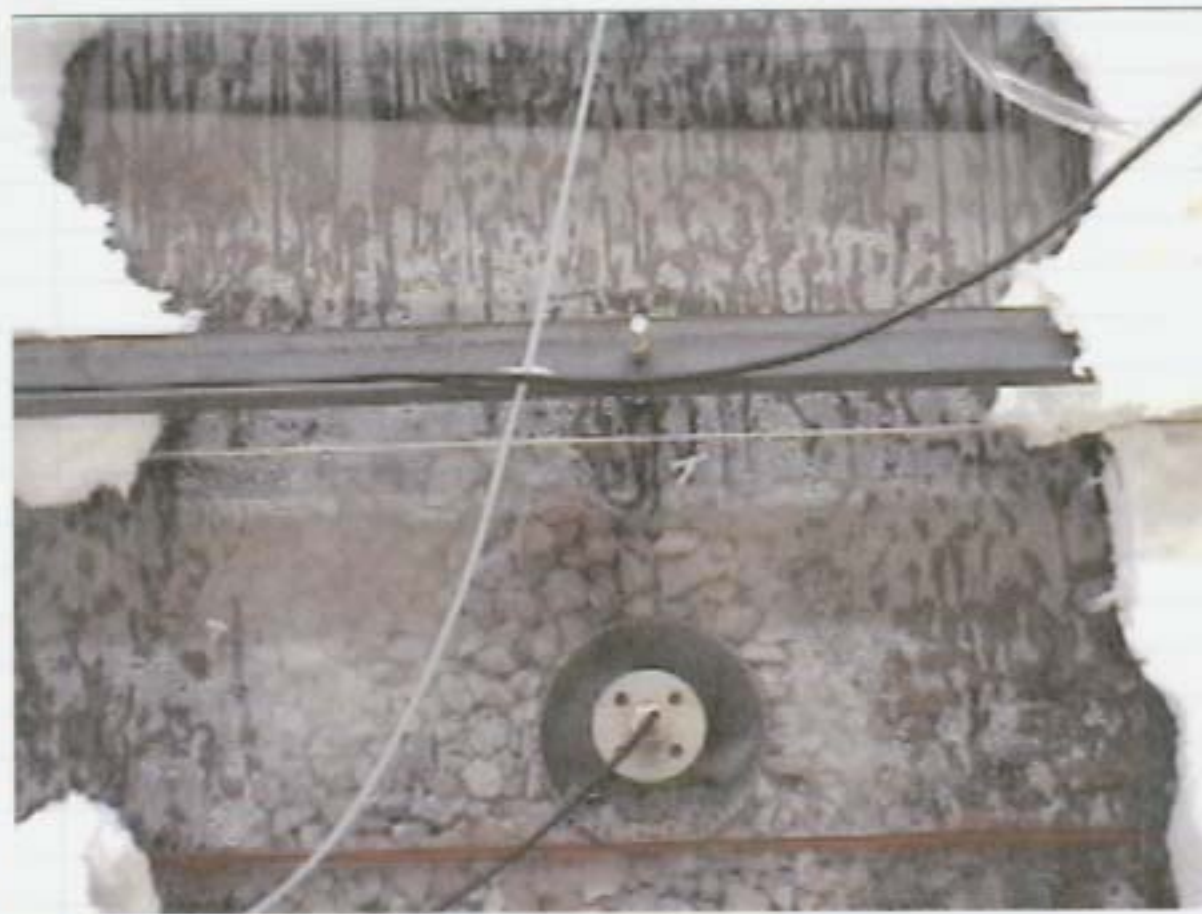




Picture taken on 12/19/00 at 1500 m.  
 7p 12/19/00 1500 m



Picture taken on 12/19/00 at 1500 m.



12/21/00 Jp 0835 hr.

Collection of water samples for alkalinity analyses

Sample Cup No	Approximate amount collected	Temp
4	40 ml	30.7°C
5	40 ml	30.3°C
14	15 ml	34.63°C
15	10 ml	40.36°C

Water were placed in 60 ml PP bottles and labeled as follows.

S4A	Sample cup 4
S5A	" 5
S14A	" 14
S15A	" 15

12/21/00 Jp 0845 hr.

Alkalinity of samples collected on previous page were determined by procedure on p. 61.

Results are shown below

===== METTLER DL21 Titrator date 21-12-00 =====

TITRATION	1	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	26.700	Sample S4
START pH	8.110	
pH0 pH	7.252	
SLOPE mV/pH	-57.5	37.5 ml sample volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.85649	
M-VALUE mmol/L	2.2868	

$$\text{Alkalinity} = \frac{.85649 \times 1 \times 50000}{37.5} = 114.20 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 21-12 00 =====

TITRATION 2 operator  
 METHOD 4011 2 04-12 0  
 WEIGHT g 1  
 IDENT 0 remarks

conc mol/L .10000  
 const 25.000  
 START pH 7.810

pH0 pH 7.252  
 SLOPE mV/pH -57.5

M-POINT pH 4.300  
 P-POINT pH 8.200

RESULT mL .80847

M-VALUE mmol/L  
 2.0212

$$\text{Alkalinity} = \frac{.80847 \times .1 \times 50000}{40} = 101.06 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 21-12 00 =====

TITRATION 3 operator  
 METHOD 4011 2 04-12 0  
 WEIGHT g 1  
 IDENT 0 remarks

conc mol/L .10000  
 const 66.700  
 START pH 7.607

pH0 pH 7.252  
 SLOPE mV/pH -57.5

M-POINT pH 4.300  
 P-POINT pH 8.200

RESULT mL .20949

M-VALUE mmol/L  
 1.3973

$$\text{Alkalinity} = \frac{.20949 \times .1 \times 50000}{15} = 69.83 \text{ mg CaCO}_3/\text{L}$$

1/8/01 JP 0905hr

Collection of water samples for pH and  
alkalinity analysis. And chemical analysis

Sample Cup No. + label	Approximate amount	Temp °C
52B	60 ml	24.83°C
54B	50 ml	31.13°C
55B	50 ml	30.86°C
512B	3 ml	39.85°C
514B	25 ml	37.5°C
515B	10 ml	44.9°C

1/8/01 JP 0930hr

The pH of samples collected on previous  
page were measured using procedure  
on p59.

## Results.

Sample	pH
52B	7.60
54B	8.06
55B	8.08
514B	8.16
515B	8.17



1/8/01 JP 1000 ml.

Alkalinity of samples on previous page were measured by procedure on p 61.

Results:

===== METTLER DL21 Titrator date 08-01-01 =====

TITRATION	1	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	52B
const	25.000	
START pH	6.922	
pH0 pH	7.259	40 ml sample volume
SLOPE mV/pH	-57.6	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.86801	
M-VALUE mmol/L	2.1700	

$$\text{Alkalinity} = \frac{.86801 \times .1 \times 50000}{40} = 108.5 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 08-01-01 =====

TITRATION	2	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	33.300	54B
START pH	7.463	
pH0 pH	7.259	30 ml sample volume
SLOPE mV/pH	-57.6	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.46869	
M-VALUE mmol/L	1.5607	

$$\text{Alkalinity} = \frac{.46869 \times .1 \times 50000}{30} = 78.115 \text{ mg CaCO}_3/\text{L}$$



===== METTLER DL21 Titrator date 08-01-01 =====

TITRATION	3	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	SBS
START pH	7.038	
pH0 pH	7.259	20ml
SLOPE mV/pH	-57.6	Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.33804	
M-VALUE mmol/L	1.6902	

$$\text{Alkalinity} = \frac{.33804 \times .1 \times 50000}{20} = 84.51 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 08-01-01 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	514B
START pH	7.694	
pH0 pH	7.259	20ml
SLOPE mV/pH	-57.6	Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.30800	
M-VALUE mmol/L	1.5400	

$$\text{Alkalinity} = \frac{.30800 \times .1 \times 50000}{20} = 77.0 \text{ mg CaCO}_3/\text{L}$$

1/8/01 JP 1115hr.

After pH and alkaliing mammary samples  
S2B, S4B, S5B, and S14B were split into  
two equal aliquots and placed in 30  
ml PP bottles. One aliquot of each  
sample was acidified to pH 2 with  
 $\text{HNO}_3$ . These acidified samples were  
labeled as follows.

S2B-A

S4B-A

S5B-A

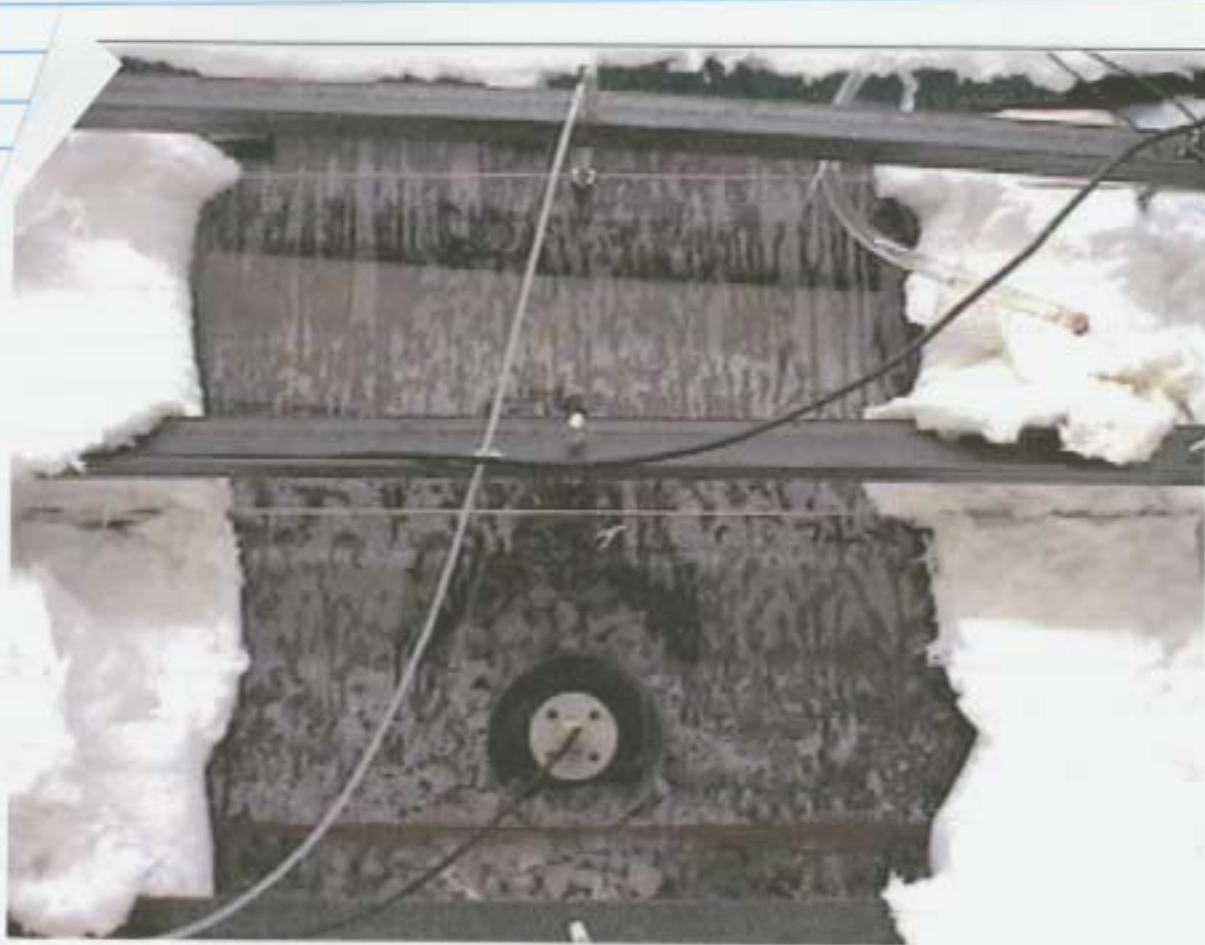
JP 1/8/01 S14B  
S15B-A

The acidified samples kept their sample  
labels. Samples were then refrigerated  
for later analysis

1/8/01 GP 1130 hrs.

Additional picture showing presence of wetty frant  
in the lab scale late test.

Picture taken at 3:00pm on 1/5/01



Picture taken at 3:00pm on 1/5/01



1/24/01 JJP 1000 km.

Collection of water samples for chemical analysis.

Sample Cup No or label	Approximate amount	Temp °C
S2C	60 ml	25.2 °C
S4C	50 ml	31.1 °C
S5C	50 ml	31.1 °C
S12C	15 ml	39.0 °C
S14C	25 ml	37.3 °C
S15C	15 ml	43.6 °C



1/24/01 gfp 1030 hr.

The pH of solution collected in previous page was measured using procedure on p 59.

### Results

Sample	pH
S2C	8.02
S4C	8.23
S5C	8.38
S12C	8.52
S14C	8.47
S15C	8.50

1/24/01 gfp 1045 hr.

Alkalinity of samples on previous page were determined using procedure on p 61.

### Results

===== METTLER DL21 Titrator date 24-01-01 =====

TITRATION	1	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	7.781	S2C
pH0 pH	7.310	40 ml
SLOPE mV/pH	-57.3	sample
		volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.89897	
M-VALUE mmol/L	2.2474	

$$\text{Alkalinity} = \frac{.89897 \times .1 \times 50000}{40} = 112.37 \text{ mg CaCO}_3 / \text{L}$$

===== METTLER DL21 Titrator date 24-01-01 =====

TITRATION 2 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 7.445

pH0 pH 7.310

SLOPE mV/pH -57.3

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .51222

M-VALUE mmol/L  
1.2806

$$\text{Alkalinity} = \frac{.51222 \times .1 \times 50000}{40} = 64.03 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 24-01-01 =====

TITRATION 3 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 50.000

START pH 7.567

pH0 pH 7.310

SLOPE mV/pH -57.3

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .34466

M-VALUE mmol/L  
1.7233

$$\text{Alkalinity} = \frac{.34466 \times .1 \times 50000}{20} = 86.16 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 24-01-01 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	SI4C
START pH	7.669	
pH0 pH	7.310	20 ml
SLOPE mV/pH	-57.3	Sample
		Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.30302	
M-VALUE mmol/L	1.5151	

$$\text{Alkalinity} = \frac{.30302 \times .1 \times 50000}{20} = 75.75 \text{ mg CaCO}_3/\text{L}$$

01/24/01 JJP 1130h.

After pH + alkalinity measurements samples were split into two equal aliquots + placed in 30 ml PP bottles. One aliquot of each sample was acidified to pH < 2 with conc HNO<sub>3</sub> (10 µl). These acidified samples were labeled as follows.

S2C-A

S4C-A

S5C-A

S12C-A

S14C-A

S15C-A

The unacidified samples kept their original sample labels. Samples were then refrigerated for later analysis.

01/24/01 JP 1300 hr.

Variable temperature was changed from a set of 33 to a set of 38. Flow rate of water introduced to the top of the apparatus was also increased to 3L/day.

01/25/01 JP 0730 hr

Variable temperature was changed from a set of 38 to a set of 40.

01/25/01 JP

Water samples of equilibrated Topical Spring tuffs were split into several aliquots. These samples were labeled TSEQ, TSEQ-A, TSEQCL and TSEQCL-A (see p 65). Each sample was split into 3 approximate equal aliquots and relabeled as shown below.

Original label	New labels
TSEQ	EQ1 EQ2 EQ3
TSEQ-A	EQ1-A EQ2-A EQ3-A
TSEQCL	EQ4 EQ5 EQ6
TSEQCL-A	EQ4-A EQ5-A EQ6-A

01/26/01 1000 hr.

Water samples were sent to Div 01 for chemical analysis. Analysis of major and minor cations will be done by ICP and anion analysis will be done by ion chromatography.

The sample list/chain of custody for the samples are shown on the following pages.

Client Name/Address		Client Purchase Order/Other ID		Site/Zone ID		SAMPLE CHAIN OF CUSTODY										Requested Turnaround:		SWRI Contact:		REMARKS	
John Patrick CNRRA/Div 20 BIO 57						Southwest Research Institute Chemistry and Chemical Engineering Division 6220 Culebra Road, San Antonio, Texas 78238-5168										<input type="checkbox"/> 1 Week <input checked="" type="checkbox"/> 2 Weeks (Normal) <input type="checkbox"/> 3 Weeks <input type="checkbox"/> Other:				Preservation a = HCl to pH <2 b = HNO <sub>3</sub> to pH <2 c = H <sub>2</sub> SO <sub>4</sub> to pH <2 d = NaOH to pH >12 e = Other (Specify)	
Sample ID	Sample Collection Date (mm/dd/yy)	Sample Collection Time (mm/dd/yy)	Matrix Type	Sample Type	# of Containers	Analyses Requested										Relinquished by (Signature)	Received by (Signature)	Relinquished by (Signature)	Comments		
S4B-A	11/3/01		W	DM	1																
S5A-A	11/3/01				1																
S14B-A	11/3/01				1																
S2B-A	11/3/01				1																
S4-A	12/15/00				1																
S5-A	12/15/00				1																
S14-A	12/15/00				1																
S15-A	12/15/00				1																
S2C-A	11/24/01				1																
S4C-A	11/24/01				1																

Matrix Types: A - Air, P - Product, S - Soil, T - Tissue, W - Water  
 Sample Types: DM - Dissolved Metals; ER - Equipment Rinseate; FB - Field Blank; MSD - Matrix Spike Duplicate; MS - Matrix Spike; TB - Trip Blank; TM - Total Metals; ES - Environmental Samples; FD - Field Duplicate  
 Relinquished by Sampler (Signature): *John Patrick*  
 Received by (Signature): *Wendy Williams*  
 Div 01 COC Form 01-01-001, Rev 1/97



Client Name/Address		SAMPLE CHAIN OF CUSTODY				Requested Turnaround:	
Jim Pirkayl CNWRA/Div 20 Bldg 57		Southwest Research Institute Chemistry and Chemical Engineering Division 6220 Culebra Road San Antonio, Texas 78238-5166				<input type="checkbox"/> 1 Week <input checked="" type="checkbox"/> 2 Weeks (Normal) <input type="checkbox"/> 3 Weeks <input type="checkbox"/> Other	
Client Purchase Order/Other ID		Site/Zone ID		Analyses Requested			
Sample ID	Sample Collection Date (mm/dd/yy)	Sample Collection Time (mm/dd/yy)	Matrix Type	Sample Type	# of Containers	REMARKS	
S5C-A	12/1/01		W	DM	1	Preservation a = HCl to pH < 2 b = HNO <sub>3</sub> to pH < 2 c = H <sub>2</sub> SO <sub>4</sub> to pH < 2 d = NaOH to pH > 12 e = Other (Specify)	
S12C-A	12/4/01				1	Project = nuclear	
S14C-A	12/4/01				1	Supply related	
S15C-A	12/4/01				1	10 F&W Part 21	
E01-A	12/5/00				1	Appendix B	
E02-A	12/5/00				1	Quintan Roo	
E03-A	12/5/00				1	Jim Pirkayl x 5167	
E04-A	12/5/00				1		
E05-A	12/5/00				1		
E06-A	12/5/00				1		
Matrix Types: A - Air; P - Product; S - Soil; T - Tissue; W - Water						Relinquished by (Signature)	
Sample Types: DM - Dissolved Metals; ER - Equipment Rinseate; FB - Field Blank; MSD - Matrix Spike Duplicate; MS - Matrix Spike; TB - Trip Blank; TM - Total Metals; ES - Environmental Samples; FD - Field Duplicate						Received by (Signature)	
Relinquished by Sampler (Signature)						Relinquished by (Signature)	
Received by (Signature)						Comments:	
Dw 01 COC Form 01-01-001, Rev 1/87						Page 1 of 1	

Client Name/Address		SAMPLE LIST/CHAIN OF CUSTODY				Requested Turnaround:	
Jim Pirkayl CNWRA/Div 20 Bldg 57		Southwest Research Institute Chemistry and Chemical Engineering Division 6220 Culebra Road San Antonio, Texas 78238-5166				<input type="checkbox"/> 1 Week <input checked="" type="checkbox"/> 2 Weeks (Normal) <input type="checkbox"/> 3 Weeks <input type="checkbox"/> Other	
Client Purchase Order/Other ID		Site/Zone ID		Analyses Requested			
Sample ID	Sample Collection Date (mm/dd/yy)	Sample Collection Time (mm/dd/yy)	Matrix Type	Sample Type	# of Containers	REMARKS	
S4B	12/1/01		W	DM	1	Preservation a = HCl to pH < 2 b = HNO <sub>3</sub> to pH < 2 c = H <sub>2</sub> SO <sub>4</sub> to pH < 2 d = NaOH to pH > 12 e = Other (Specify)	
S5B	12/1/01				1	Project = nuclear	
S14B	12/1/01				1	Supply related	
S2B	12/1/01				1	10 F&W Part 21	
S4	12/5/00				1	Appendix B	
S5	12/5/00				1	Quintan Roo	
S14	12/5/00				1	Jim Pirkayl x 5167	
S15	12/5/00				1		
S2C	12/1/01				1		
S4C	12/1/01				1		
Matrix Types: A - Air; P - Product; S - Soil; T - Tissue; W - Water						Relinquished by (Signature)	
Sample Types: DM - Dissolved Metals; ER - Equipment Rinseate; FB - Field Blank; MSD - Matrix Spike Duplicate; MS - Matrix Spike; TB - Trip Blank; TM - Total Metals; ES - Environmental Samples; FD - Field Duplicate						Received by (Signature)	
Relinquished by Sampler (Signature)						Relinquished by (Signature)	
Received by (Signature)						Comments:	
Dw 01 COC Form 01-01-001, Rev 1/87						Page 1 of 1	

SAMPLE CHAIN OF CUSTODY				Requested Turnaround:	
Client Name/Address: <b>Jim Pickett</b> <b>CWRA / Div 20</b> <b>BID 57</b>				<input type="checkbox"/> 1 Week <input checked="" type="checkbox"/> 2 Weeks (Normal) <input type="checkbox"/> 3 Weeks <input type="checkbox"/> Other:	
Client Purchase Order/Other ID: <b>SWRI Contact:</b>				Southwest Research Institute Chemistry and Chemical Engineering Division 6220 Culebra Road San Antonio, Texas 78238-5166	
Site/Zone ID: <b>Analyses Requested</b>				Preservation a = HCl to pH <2 b = HNO <sub>3</sub> to pH <2 c = H <sub>2</sub> SO <sub>4</sub> to pH <2 d = NaOH to pH >12 e = Other (Specify)	
Sample ID	Sample Collection Date (mm/dd/yy)	Sample Collection Time (mm/dd/yy)	Matrix Type	Sample Type	# of Containers
SSC	12/24/01		W	DM	1
S12C	12/24/01				1
S14C	12/24/01				1
S15C	12/24/01				1
EQ1	12/24/02				1
EQ2	12/24/02				1
EQ3	12/24/02				1
EQ4	12/24/02				1
EQ5	12/24/02				1
EQ6	12/24/02				1
Matrix Types: A - Air; P - Product; S - Soil; T - Tissue; W - Water Sample Types: DM - Dissolved Metals; ER - Equipment Rinseate; FB - Field Blank; MSD - Matrix Spike Duplicate; MS - Matrix Spike; TB - Trip Blank; TM - Total Metals; ES - Environmental Samples; FD - Field Duplicate					
Relinquished by (Signature):			Relinquished by (Signature):		
Received by (Signature):			Received by (Signature):		
Relinquished by (Signature):			Relinquished by (Signature):		
Comments:			Comments:		

2/5/01 JP 0815h

Collected another set of water samples for chemical analysis

Sample Cup No + label	Appropriate amount	Temp °C
S2D	60	25.5°C
S4D	50	33.7°C
S5D	45	32.7°C
S12D	20	42.3°C
S14D	25	40.2°C
S15D	15	46.5°C

2/5/01 JP 0830h

The pH of solutions collected above were measured using procedure on p 59.

Results Sample	pH
S2D	8.22
S4D	8.20
S5D	8.41
S12D	8.60
S14D	8.36
S15D	8.58

2/5/01 JF 0900hr.

Alkalinity of samples collected on previous page  
were measured by procedure on p 61.

Results -

===== METTLER DL21 Titrator date 05-02-01 =====

TITRATION 1 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 8.039

pH0 pH 7.023

SLOPE mV/pH -57.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .77568

M-VALUE mmol/L

1.9392

$$\text{Alkalinity} = \frac{.77568 \times .1 \times 500000}{40} = 96.96 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 05-02-01 =====

TITRATION 2 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 50.000

START pH 7.567

pH0 pH 7.023

SLOPE mV/pH -57.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .22100

M-VALUE mmol/L

1.1050

$$\text{Alkalinity} = \frac{.22100 \times .1 \times 500000}{20} = 55.25 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 05-02-01 =====

TITRATION 3 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 40.000

START pH 7.902

pH0 pH 7.023

SLOPE mV/pH -57.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .41396

M-VALUE mmol/L  
1.6559

$$\text{Alkalinity} = \frac{.41396 \times .1 \times 50000}{25} = 82.79 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 05-02-01 =====

TITRATION 4 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 40.000

START pH 8.082

pH0 pH 7.023

SLOPE mV/pH -57.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .27774

M-VALUE mmol/L  
1.1110

$$\text{Alkalinity} = \frac{.27774 \times .1 \times 50000}{20} = 69.435 \text{ mg CaCO}_3/\text{L}$$

2/5/01 JP 0930h.

After pH & alkalinity measurements samples were split into 2 aliquots & placed in 30 ml PP bottles. One aliquot was acidified with  $\text{HNO}_3$  (10 ml). These acidified samples were labeled as follows.

S2D-A

S4D-A

S5D-A

S12D-A

S14D-A

S15D-A

The unacidified samples kept their original labels. Samples were refrigerated for later analysis.

2/5/01 JP 1000h.

Variable temperature was changed from a setting of 40 to a setting of 44.

2/6/01 JP 0930h.

Variable temperature was changed from a setting of 44 to a setting of 48.

2/7/01 JP 0930h.

Variable temperature was changed from setting of 48 to 50.



Temperature and Relative Humidity Readings  
(cont. from p 57)

Date/Time	RH	Temp	RH	Temp
02/13 0730	17.83	10.92	86.44%	66.5°C
02/14 1350	17.86	10.94	86.62%	66.7°C
02/15 0940	18.31	10.83	89.44%	65.4°C
02/16 0945	18.16	10.81	88.5%	65.1°C
02/20 0855	16.09	11.11	75.6%	68.9°C
02/21 0940	14.75	11.28	67.2%	71.0°C
02/22 0715	14.14	11.45	63.4%	73.1°C
02/23 1010	14.03	11.45	62.7%	73.1°C
02/24 0740	13.04	11.79	56.5%	77.3°C
02/27 0730	14.22	11.65	63.9%	75.6°C
02/28 0940	14.94	11.58	68.4%	74.75°C
03/01 0820	15.99	11.21	74.44%	70.1°C
03/02 0920	16.06	11.27	75.4%	70.9°C
03/05 0725	13.77	11.78	61.06%	77.2°C
03/06 0940	17.98	11.00	81.4%	67.5°C
03/07 0950	17.45	11.27	84.1%	70.9°C
03/08 0750	18.22	11.24	88.9%	70.5°C
03/09 1000	18.56	11.11	91.0%	68.9°C
03/12 0740	18.41	11.12	90.1%	69.0°C
03/13 0740	18.32	11.10	89.5%	68.7°C
03/14 0740	17.15	11.24	82.2%	70.5°C
03/15 0745	18.63	11.03	91.4%	67.9°C
3/16 0745	17.05	11.29	81.6%	71.1°C
3/19 0750	15.80	11.37	73.7%	72.1°C
3/20 1020	16.80	11.22	80.0%	70.2°C
3/21 1010	17.40	11.20	83.7%	70.0°C
3/22 0940	14.25	11.73	76.6%	74.0°C
3/23 0935	16.08	11.88	75.5%	78.5°C
3/24 0715	17.72	11.70	85.7%	76.2°C
3/27 0745	17.90	11.78	86.9%	77.2°C
3/28 0940	18.24	11.82	89.0%	77.7°C
3/29 0735	18.27	11.86	89.2%	78.2°C

Date/Time	RH	Temp	RH	Temp
3/30	18.54	11.89	90.9%	78.6°C
4/2	15.19	12.38	69.9%	84.7°C
4/3 0955	14.16	12.57	63.5%	87.1°C
4/4 0945	13.82	12.60	61.4%	87.5°C
4/5 0720	13.05	12.81	56.6%	90.1°C
4/6 0945	13.46	12.42	59.1%	85.2°C
4/9 0705	14.02	12.38	62.6%	84.7°C
4/10 0940	13.92	12.49	62.0%	86.1°C
4/11 0945	13.75	12.46	60.9%	85.7°C
4/12 0725	13.54	12.56	59.6%	87.0°C
4/13 0745	12.29	13.06	51.8%	93.2°C
4/16 1740	7.84	14.03	24.0%	112.8°C
4/17 1415	7.87	14.57	24.2%	112.1°C
4/18 1815	10.36	13.56	39.7%	99.5°C
4/19 1340	10.16	13.72	38.5%	101.5°C
4/20 1435	7.62	14.64	22.6%	113.0°C
4/23 1700	6.87	15.13	17.9%	119.1°C
4/24 0730	8.95	13.00	30.9%	92.5°C
4/25 1400	10.47	11.57	40.4%	74.6°C
4/26 1140	11.06	11.10	44.1%	68.7°C
4/27 0745	10.95	10.87	43.4%	65.8°C
4/27 1545	15.76	10.25	73.5%	58.1°C
4/30 1645	18.33	10.15	89.6%	56.9°C
5/1 1640	18.97	10.12	93.4%	56.5°C
5/2 1620	18.80	10.68	92.5%	63.5°C
5/3 0740	17.68	11.09	85.5%	68.6°C
5/4 1515	17.28	11.26	83.0%	70.7°C
5/7 1005	15.70	12.00	76.0%	80.0°C
5/8 0655	14.61	12.28	66.3%	83.5°C
5/9 1400	11.61	12.83	47.5%	90.4°C
5/10 1500	13.40	12.56	61.9%	87.0°C
5/11 1500	14.33	12.51	64.6%	86.4°C
5/14 0715	10.74	13.58	42.1%	99.8°C
5/15 0725	9.40	13.88	34.33%	103.5°C
5/16 0715	7.96	14.79	24.7%	114.9°C
5/17 0750	8.68	13.79	29.2%	102.4°C

Readings cont on p 138

2/15/01 JP

Cont. from p 53

Temperature data is stored in Solary file.

test-3-103.csv (2/16-2/19)	test-3-136.csv (4/5-4/6)
test-3-104.csv (2/19-2/20)	test-3-137.csv (4/6-4/9)
test-3-105.csv (2/20-2/21)	test-3-138.csv (4/9-4/10)
test-3-106.csv (2/21-2/22)	test-3-139.csv (4/10-4/11)
test-3-107.csv (2/22-2/23)	test-3-140.csv (4/11-4/12)
test-3-108.csv (2/23-2/24)	test-3-141.csv (4/12-4/13)
test-3-109.csv (2/24-2/27)	test-3-142.csv (4/13-4/16)
test-3-110.csv (2/27-2/28)	test-3-143.csv (4/16-4/17)
test-3-111.csv (2/28-2/29)	test-3-144.csv (4/17-4/18)
test-3-112.csv (2/29-2/30)	test-3-145.csv (4/18-4/19)
test-3-113.csv (2/30-3/1)	test-3-146.csv (4/19-4/20)
test-3-114.csv (3/1-3/2)	test-3-147.csv (4/20-4/23)
test-3-115.csv (3/2-3/3)	test-3-148.csv (4/23-4/24)
test-3-116.csv (3/3-3/4)	test-3-149.csv (4/24-4/25)
test-3-117.csv (3/4-3/5)	test-3-150.csv (4/25-4/26)
test-3-118.csv (3/5-3/6)	test-3-151.csv (4/26-4/27)
test-3-119.csv (3/6-3/7)	test-3-152.csv (4/27-4/30)
test-3-120.csv (3/7-3/8)	test-3-153.csv (4/30-5/1)
test-3-121.csv (3/8-3/9)	test-3-154.csv (5/1-5/2)
test-3-122.csv (3/9-3/10)	test-3-155.csv (5/2-5/3)
test-3-123.csv (3/10-3/11)	test-3-156.csv (5/3-5/4)
test-3-124.csv (3/11-3/12)	test-3-157.csv (5/4-5/7)
test-3-125.csv (3/12-3/13)	test-3-158.csv (5/7-5/8)
test-3-126.csv (3/13-3/14)	test-3-159.csv (5/8-5/9)
test-3-127.csv (3/14-3/15)	test-3-160.csv (5/9-5/10)
test-3-128.csv (3/15-3/16)	test-3-161.csv (5/10-5/11)
test-3-129.csv (3/16-3/17)	test-3-162.csv (5/11-5/14)
test-3-130.csv (3/17-3/18)	test-3-163.csv (5/14-5/15)
test-3-131.csv (3/18-3/19)	test-3-164.csv (5/15-5/16)
test-3-132.csv (3/19-3/20)	test-3-165.csv (5/16-5/17)
test-3-133.csv (3/20-3/21)	test-3-166.csv (5/17-5/18)
test-3-134.csv (3/21-3/22)	test-3-167.csv (5/18-5/21)
test-3-135.csv (3/22-3/23)	Temp data cont on p

3/1/01 JP 0930 hr.

Collected another set of water samples for chemical analysis.

Sample Cup No	Approximate Amount	Temp °C
S1E	60	25.6
S2E	60	25.2
S4E	10	37.4
S5E	25	35.2
S14E	20	45.1

3/1/01 JP 0930 hr.

The pH of solutions collected above were measured w/ procedure on p59

Sample	pH
S1E	8.41
S2E	8.24
S4E	7.98
S5E	8.25
S14E	8.54

3/1/01 JF 1015hr.

Alkalinity of samples collected on previous page  
were recorded by procedure on p 61.

Results:

===== METTLER DL21 Titrator date 01-03-01 =====

TITRATION 1 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 8.481

pH0 pH 6.980

SLOPE mV/pH -57.0

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .05857

P-VALUE mmol/L

.14642

RESULT mL 1.1080

M-VALUE mmol/L

2.7701

$$\text{Carbonate Alkalinity} = 2 \left( \frac{.05857 \times 1.1 \times 50000}{40} \right) = 14.64 \text{ mg CaCO}_3/\text{L}$$

$$\text{Bicarbonate conc} = \left( \frac{1.14642 \times 1.1 \times 50000}{40} \right) - 14.64 = 131.18 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 01-03-01 =====

TITRATION 2 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 7.714

pH0 pH 6.980

SLOPE mV/pH -57.0

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .65936

M-VALUE mmol/L

1.6484

Alkalinity =  $\frac{.65936 \times 1.1 \times 50000}{2540} = 131.87 \text{ mg CaCO}_3/\text{L}$

JF 3/1/01

===== METTLER DL21 Titrator date 01-03-01 =====

TITRATION	3	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	
START pH	7.852	
pH0 pH	6.980	
SLOPE mV/pH	-57.0	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.22501	
M-VALUE mmol/L	1.1250	

55E  
20ml  
sample  
vol.

$$\text{Alkalinity} = \frac{.22501 \times .1 \times 50000}{20} = 56.25 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 01-03-01 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	66.670	
START pH	8.092	
pH0 pH	6.980	
SLOPE mV/pH	-57.0	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.25970	
M-VALUE mmol/L	1.7314	

514E  
15ml  
sample  
volume

$$\text{Alkalinity} = \frac{.25970 \times .1 \times 50000}{15} = 86.57 \text{ mg CaCO}_3/\text{L}$$

3/1/01 gp 1100 hr.

After pH + alkalinity nanomole sample were split into 2 aliquots and placed in 30 ml PP bottles. One aliquot was acidified with  $\text{HNO}_3$  (10%) These acidified samples were labeled as follows.

S1E-A

S2E-A

S4E-A

S5E-A

S14E-A

The unacidified samples kept their original labels. Samples were refrigerated for later analysis.

03/01/01 gp 1135 hr.

Variable Transducer was changed from 50 to 52.

03/02/01 gp 0430 hr.

Variable transducer was changed from 52 to 55.



### ICP analyses of major and minor cations

Date collected	Solution temp C
11/1/2011	15.0
11/2/2011	15.0
11/3/2011	15.0
11/4/2011	15.0
11/5/2011	15.0
11/6/2011	15.0
11/7/2011	15.0
11/8/2011	15.0
11/9/2011	15.0
11/10/2011	15.0
11/11/2011	15.0
11/12/2011	15.0
11/13/2011	15.0
11/14/2011	15.0
11/15/2011	15.0
11/16/2011	15.0
11/17/2011	15.0
11/18/2011	15.0
11/19/2011	15.0
11/20/2011	15.0
11/21/2011	15.0
11/22/2011	15.0
11/23/2011	15.0
11/24/2011	15.0
11/25/2011	15.0
11/26/2011	15.0
11/27/2011	15.0
11/28/2011	15.0
11/29/2011	15.0
11/30/2011	15.0
12/1/2011	15.0
12/2/2011	15.0
12/3/2011	15.0
12/4/2011	15.0
12/5/2011	15.0
12/6/2011	15.0
12/7/2011	15.0
12/8/2011	15.0
12/9/2011	15.0
12/10/2011	15.0
12/11/2011	15.0
12/12/2011	15.0
12/13/2011	15.0
12/14/2011	15.0
12/15/2011	15.0
12/16/2011	15.0
12/17/2011	15.0
12/18/2011	15.0
12/19/2011	15.0
12/20/2011	15.0
12/21/2011	15.0
12/22/2011	15.0
12/23/2011	15.0
12/24/2011	15.0
12/25/2011	15.0
12/26/2011	15.0
12/27/2011	15.0
12/28/2011	15.0
12/29/2011	15.0
12/30/2011	15.0
12/31/2011	15.0

### IC analyses of anions

Alkalinity  
in  
mg CaCO<sub>3</sub>/l

Cation analyses are cont on  
following page

Cation analyses are cont on  
following page

3/21/01 JJP 1305hr.

Collected new set of water sample for  
chemical analysis.

Sample Cup No Label	Appropriate Amount	Temp °C
S1F	60ml	25.37°C
S2F	60ml	25.14°C
S4F	50ml	36.56°C
S5F	50ml	35.85°C
S14F	20ml	46.80°C
S15F	15ml	60.53°C

3/21/01 JJP 1330hr.

pH of solutions collected above. See p  
59 for procedure.

Sample	pH
S1F	8.33
S2F	8.22
S4F	8.28
S5F	8.22
S14F	8.52
S15F	8.75

3/21/01 1400hr JJP

Alkalinity of samples on previous page were  
determined by procedure in p. 61.

Results

===== METTLER DL21 Titrator date 21-03-01 ==

TITRATION	1	operator
METHOD	4011	2 04-12 @
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.223	S1F
pH0 pH	7.043	
SLOPE mV/pH	-57.5	40ml sample volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.05226	
P-VALUE mmol/L	.13065	
RESULT mL	1.3223	
M-VALUE mmol/L	3.3057	

$$\text{Carbonate Alkalinity} = 2 \left( \frac{.05226 \times .1 \times 50000}{40} \right) = 13.065 \text{ mg CaCO}_3/\text{L}$$

$$\text{Bicarbonate Alkalinity} = \left( \frac{1.3223 \times .1 \times 50000}{40} \right) = 13.065 = 158.75 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 21-03-01 =====

TITRATION	2	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	7.873	
pH0 pH	7.043	
SLOPE mV/pH	-57.5	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.76014	
M-VALUE mmol/L	1.9004	

52F

40 ml  
Sample  
Volume

$$\text{Alkalig} = \frac{.76014 \times .1 \times 50000}{40} = 95.02 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 21-03-01 =====

TITRATION	3	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	
START pH	7.539	
pH0 pH	7.043	
SLOPE mV/pH	-57.5	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.24306	
M-VALUE mmol/L	1.2153	

54F

20 ml  
Sample  
Volume

$$\text{Alkalig} = \frac{.24306 \times .1 \times 50000}{20} = 60.765 \text{ mg CaCO}_3/\text{L}$$

===== NETTLER DL21 Titrator date 21-03-01 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	50.000	
START pH	7.423	55F
PH0 pH	7.043	
SLOPE mV/pH	-57.5	20ml
M-POINT pH	4.300	Sample
P-POINT pH	8.200	Volume
RESULT mL	.19745	
M-VALUE mmol/L	.98724	

$$\text{Alkalinity} = \frac{.19745 \times .1 \times 50000}{20} = 49.36 \text{ mg CaCO}_3/\text{L}$$

===== NETTLER DL21 Titrator date 21-03-01 =====

TITRATION	5	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	66.670	
START pH	8.068	
PH0 pH	7.043	514F
SLOPE mV/pH	-57.5	15ml
M-POINT pH	4.300	Sample
P-POINT pH	8.200	Volume
RESULT mL	.25390	
M-VALUE mmol/L	1.6928	

$$\text{Alkalinity} = \frac{.25390 \times .1 \times 50000}{15} = 84.63 \text{ mg CaCO}_3/\text{L}$$

3/21/01 gp 1500h.

After pH & alkalinity measurements samples were split into 2 aliquots & placed in 30 ml PP bottles. One aliquot was acidified by adding 10  $\mu$  HNO<sub>3</sub>. These samples were labeled as follows.

S1F-A  
S2F-A  
S4F-A  
S5F-A  
S14F-A  
S15F-A

The acidified samples returned their original labels. Samples were reanalyzed for later analysis.

3/21/01 gp 1530h.

Variable frequency was changed from 55 to 58.60, gp 3/21/01

4/5/01 gp 0830h.

Relative humidity readings over the past 4 or 5 days indicated problems with flow to the system (probably problems with ceramic tube delivery water to system). The top of the apparatus was dismantled, the ceramic tube was cleaned with acid to remove residues and the tubes were repositioned. The top of the apparatus was then reassembled.

4/12/01 gp 0910h.

Collected water samples for chemical analysis

Sample Cup No label	Approximate Amount	Temp °C
S1G	60 ml	27.3°C
S2G	60 ml	27.8°C
S3G	8 ml	27.2°C
S4G	50 ml	41.7°C
S5G	50 ml	39.0°C
S12G	20 ml	53.3°C
S14G	20 ml	50.1°C
S15G	15 ml	58.8°C

4/12/01 gp 0945h.

pH of solutions above were measured using procedure on p 59.

Sample	pH
S1G	8.25
S2G	8.34
S3G	7.81
S4G	8.35
S5G	8.38
S12G	8.68
S14G	8.56
S15G	8.70



4/12/01 1030hr.

Alkalinity of sample in previous page were determined  
using procedure on p 61

Results

===== METTLER DL21 Titrator date 12-04-01 =====

TITRATION	1	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.212	516
pH0 pH	6.950	40ml sample volume
SLOPE mV/pH	-55.6	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.02668	
P-VALUE mmol/L	.06669	
RESULT mL	.82190	
M-VALUE mmol/L	2.0540	

$$\text{Alkalinity} = \frac{(.02668 + .82190) * .1 * 50000}{40} = 106.07 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 12-04-01 =====

TITRATION	3	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.259	526
pH0 pH	6.950	40ml sample volume
SLOPE mV/pH	-55.6	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.01885	
P-VALUE mmol/L	.04712	
RESULT mL	.82314	
M-VALUE mmol/L	2.0579	

$$\text{Alkalinity} = \frac{(.01885 + .82314) * .1 * 50000}{40} =$$

$$105.25 \text{ mg CaCO}_3/\text{L}$$

===== METTLER DL21 Titrator date 12-04-01 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	40.000	
START pH	7.669	546
pH0 pH	6.950	25ml sample volume
SLOPE mV/pH	-55.6	
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.27928	
M-VALUE mmol/L	1.1171	

$$\text{Alkalinity} = \frac{.27928 * .1 * 50000}{25} =$$

$$55.86 \text{ mg CaCO}_3/\text{L}$$

----- METTLER DL21 Titrator date 12-04-12

TITRATION 6 operator

METHOD 4011 2 04

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 40.000

START pH 8.157

pH0 pH 6.950

SLOPE mV/pH -55.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .28423

M-VALUE mmol/L 1.1369

$$\text{Alkalinity} = \frac{.28423 \times .1 \times 50000}{25} =$$

56.85 mg CaCO<sub>3</sub>/L

$$\text{Alkalinity} = \frac{(.02204 + .30884) \times .1 \times 50000}{20} =$$

82.72 mg CaCO<sub>3</sub>/L

----- METTLER DL21 Titrator date 12-04-12

TITRATION 9 operator

METHOD 4011 2 04

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 50.000

START pH 8.441

pH0 pH 6.950

SLOPE mV/pH -55.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .02204

P-VALUE mmol/L .11020

RESULT mL .30884

M-VALUE mmol/L 1.5442

----- METTLER DL21 Titrator date 12-04-12

TITRATION 8 operator

METHOD 4011 2 04-12

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 66.700

START pH 8.176

pH0 pH 6.950

SLOPE mV/pH -55.6

M-POINT pH 4.300

P-POINT pH 8.200

RESULT mL .17601

M-VALUE mmol/L 1.1740

$$\text{Alkalinity} = \frac{.17601 \times .1 \times 50000}{15} = 58.67 \text{ mg CaCO}_3/\text{L}$$

4/12/01 Jp 1130hr

Variable frequency changed from setting of 60 to 65.

4/19/01 Jp 1000hr.

Problems with the cassette pump caused delivery of water to the apparatus to be severely decreased over a several day period. This caused temperature around the adit to increase and relative humidity to decrease. Increase in temperature caused drying out of the rock surrounding the adit.

4/20/01 The cassette pump was repaired and delivery of water at an increased rate was resumed. The variable frequency was set to 40 to allow rewetting of the rock surrounding the adit.

5/1/01 1645hr.

Jp The variable frequency was set to 50.

5/2/01 1625

Jp The variable frequency was set to 55

5/4/01 1540

Jp The variable frequency was set to 60

5/10/01 1500 Jp

Variable frequency set to 63

5/11/01 1500hr Jp

Variable frequency set to 65

5/15 Jp 1020hr

Variable frequency set to 60.

5/17 Jp 1245hr

Variable frequency set to 55.

5/18 Jp 0730hr.

Variable frequency set to 48

6/20 Jp 0900hr.

A new ceramic tube was placed in the apparatus to deliver water. The old ceramic tube was clogged.

Temperature and RH Readings  
(cont from p 113)

mAt

Date/Time	RH	Temp	RH	Temp
5/18 0745	16.46	11.75	77.9%	76.8°C
5/21 1000	14.94	11.85	68.7%	78.1°C
5/26 0840	15.77	11.76	73.6%	77.0°C
5/29 0735	15.68	11.83	73.0%	77.9°C
5/30 0735	15.41	11.87	71.3%	78.4°C
5/31 0745	16.07	11.92	75.4%	79.0°C
6/1 0755	15.32	11.87	70.7%	78.4°C
6/4 0750	15.38	12.14	71.1%	81.7°C
6/5 0745	15.69	11.83	73.1%	77.9°C
6/6 0715	15.30	11.81	70.6%	77.6°C
6/7 0735	14.89	11.85	68.1%	78.1°C
6/8 0730	14.69	11.87	66.8%	78.4°C
6/11 0730	14.59	11.60	66.2%	75.0°C
6/12 0725	14.55	11.65	65.9%	75.6°C
6/13 0715	14.39	11.65	64.9%	75.6°C
6/14 0720	14.70	11.46	66.9%	73.2°C
6/15 0740	14.35	11.36	64.7%	72.0°C
6/18 0740	14.10	11.28	63.1%	71.0°C
6/19 0735	14.38	11.26	64.9%	70.7°C
6/20 1040	12.22	11.65	51.4%	75.0°C
6/21 0715	16.81	11.03	80.1%	67.9°C
6/22 0800	17.10	11.02	81.9%	67.7°C
6/25 0800	16.97	11.21	81.1%	70.1°C
6/26 0820	17.93	10.92	87.1%	66.5°C
6/27 1200	17.46	10.97	84.1%	67.1°C
6/28 0720	17.13	11.02	82.1%	67.7°C
6/29 0710	17.45	10.99	84.3%	67.4°C
7/2 0745	17.11	10.82	81.9%	65.2°C
7/3 0735	16.88	10.72	80.5%	64.0°C
7/5 0745	16.72	10.75	79.5%	64.4°C
7/6 0730	16.61	10.77	78.8%	64.6°C
7/9 0715	16.16	10.87	76.0%	65.8°C

mAt

Date/Time	RH	Temp	RH	Temp
7/10 0710	17.06	10.69	81.6%	63.6°C
7/11 1015	16.48	10.72	78.0%	64.0°C
7/12 0705	17.44	10.63	84.0%	62.9°C
7/13 0730	17.68	10.63	85.5%	62.9°C
7/16 0740	17.38	<del>0740</del> 10.72	83.6%	64.0°C
7/17 0730	18.79	10.63	92.4%	62.9°C
7/18 0725	18.80	10.61	92.5%	62.6°C
7/19 0740	19.16	10.56	94.7%	62.0°C
7/20 0720	18.87	10.59	92.9%	62.4°C
7/23 0710	16.41	10.86	77.6%	65.7°C
7/24 0710	19.22	10.70	95.1%	63.7°C
7/25 0745	17.86	10.69	86.6%	63.6°C
7/26 0705	17.70	10.68	85.6%	63.5°C
7/27 0720	17.75	10.71	85.9%	63.8°C
7/30 0730	17.37	10.77	83.6%	64.6°C
7/31 0710	17.33	10.75	83.3%	64.4°C
8/1 0710	17.48	10.74	84.2%	64.2°C
8/2 0735	17.78	10.68	86.1%	63.5°C
8/3 0750	17.63	10.67	85.2%	63.4°C
8/6 0720	17.08	10.81	81.7%	65.1°C
8/7 0720	17.30	10.80	83.1%	65.0°C
8/8 0740	16.99	10.77	81.2%	64.6°C
8/9 0725	16.72	10.84	79.5%	65.5°C
8/10 0725	16.68	10.86	79.2%	65.7°C
8/13 0715	19.72	10.56	98.2%	62.0°C
8/14 0655	20.02	10.59	100.1%	62.4°C
8/15 0730	4.01	10.75		64.4°C
8/16 0745	4.01	10.79		64.9°C
8/17 0745	18.37	10.77	89.8%	64.6°C
8/20 0910	17.23	10.70	82.7%	63.7°C
8/21 0905	17.38	9.02	83.6%	42.7°C
8/22 0935	17.36	9.01	83.5%	42.6°C
8/23 0925	17.49	9.02	84.3%	42.7°C
8/24 0920	17.52	8.99	84.5%	42.4°C

(cont on p 142)

5/29/01 JF

Cont. from p 114

Temperature data is stored in following files

test-3-168.csv (5/21-5/29)	test-3-200.csv (7/16-7/19)
test-3-169.csv (5/29-5/30)	test-3-201.csv (7/19-7/20)
test-3-170.csv (5/30-5/31)	test-3-202.csv (7/20-7/23)
test-3-171.csv (5/31-6/1)	test-3-203.csv (7/23-7/27)
test-3-172.csv (6/1-6/4)	test-3-204.csv (7/27-7/30)
test-3-173.csv (6/4-6/5)	test-3-205.csv (7/30-8/3)
test-3-174.csv (6/5-6/6)	test-3-206.csv (8/3-8/6)
test-3-175.csv (6/6-6/7)	test-3-207.csv (8/6-8/10)
test-3-176.csv (6/7-6/8)	test-3-208.csv (8/10-8/13)
test-3-177.csv (6/8-6/11)	test-3-209.csv (8/13-8/17)
test-3-178.csv (6/11-6/14)	test-3-210.csv (8/17-8/20)
test-3-179.csv (6/12-6/13)	test-3-211.csv (8/20-8/24)
test-3-180.csv (6/13-6/14)	test-3-212.csv (8/24-8/27)
test-3-181.csv (6/14-6/15)	test-3-213.csv (8/27-8/31)
test-3-182.csv (6/15-6/18)	test-3-214.csv (8/31-9/4)
test-3-183.csv (6/18-6/19)	test-3-215.csv (9/4-9/7)
test-3-184.csv (6/19-6/20)	test-3-216.csv (9/7-9/10)
test-3-185.csv (6/20-6/21)	test-3-217.csv (9/10-9/14)
test-3-186.csv (6/21-6/22)	test-3-218.csv (9/14-9/17)
test-3-187.csv (6/22-6/25)	test-3-219.csv (9/17-9/21)
test-3-188.csv (6/25-6/26)	test-3-220.csv (9/21-9/24)
test-3-189.csv (6/26-6/27)	test-3-221.csv (9/24-9/28)
test-3-190.csv (6/27-6/28)	test-3-222.csv (9/28-10/1)
test-3-191.csv (6/28-6/29)	test-3-223.csv (10/1-10/5)
test-3-192.csv (6/29-7/2)	test-3-224.csv (10/5-10/8)
test-3-193.csv (7/2-7/3)	test-3-225.csv (10/8-10/12)
test-3-194.csv (7/3-7/5)	test-3-226.csv (10/12-10/15)
test-3-195.csv (7/5-7/6)	test-3-227.csv (10/15-10/19)
test-3-196.csv (7/6-7/9)	test-3-228.csv (10/19-10/23)
test-3-197.csv (7/9-7/12)	test-3-229.csv (10/23-10/26)
test-3-198.csv (7/12-7/13)	test-3-230.csv (10/26-10/29)
test-3-199.csv (7/13-7/15)	test-3-231.csv (10/29-11/2)

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test-3-232.csv (11/2-11/5)	test-3-268.csv (3/13-3/18)
test-3-233.csv (11/5-11/9)	test-3-269.csv (3/18-3/20) *
test-3-234.csv (11/9-11/12)	test-3-270.csv (3/20-3/22)
test-3-235.csv (11/12-11/15)	test-3-271.csv (3/22-3/25)
test-3-236.csv (11/15-11/21)	test-3-272.csv (3/25-3/26)
test-3-237.csv (11/21-11/26)	test-3-273.csv (4/1-4/1)
test-3-238.csv (11/26-11/30)	test-3-274.csv (4/5-4/5)
test-3-239.csv (11/30-12/3)	test-3-275.csv (4/10-4/10)
test-3-240.csv (12/3-12/7)	test-3-276.csv (4/15-4/17)
test-3-241.csv (12/7-12/10)	test-3-277.csv (4/19-4/22)
test-3-242.csv (12/10-12/14)	test-3-278.csv (4/22-4/26)
test-3-243.csv (12/14-12/17)	test-3-279.csv (4/26-4/29)
test-3-244.csv (12/17-12/21)	test-3-280.csv (5/3-5/3)
test-3-245.csv (12/21-12/27)	test-3-281.csv (5/10-5/10)
test-3-246.csv (12/27-12/31)	test-3-282.csv (5/15-5/15)
test-3-247.csv (12/31-1/4)	test-3-283.csv (5/29-5/29)
test-3-248.csv (1/4-1/7)	test-3-284.csv (6/3-6/3)
test-3-249.csv (1/7-1/11)	test-3-285.csv (6/3-6/4)
test-3-250.csv (1/11-1/14)	test-3-286.csv (6/6-6/10)
test-3-251.csv (1/14-1/18)	test-3-287.csv (6/10-6/17)
test-3-252.csv (1/18-1/21)	test-3-288.csv (7/29-7/29)
test-3-253.csv (1/21-1/24)	test-3-289.csv (7/30-8/2)
test-3-254.csv (1/24-1/28)	test-3-290.csv (8/2-8/5)
test-3-255.csv (1/28-2/1)	test-3-291.csv (8/8-8/12)
test-3-256.csv (2/1-2/4)	test-3-292.csv (10/1-10/4)
test-3-257.csv (2/4-2/8)	test-3-293.csv (10/4-10/7)
test-3-258.csv (2/8-2/11)	test-3-294.csv (10/7-10/8)
test-3-259.csv (2/11-2/15)	test-3-295.csv (10/10-10/14)
test-3-260.csv (2/15-2/18)	test-3-296.csv (10/14-10/18)
test-3-261.csv (2/18-2/22)	test-3-297.csv (10/18-10/21)
test-3-262.csv (2/22-2/25)	test-3-298.csv (10/21-10/25)
test-3-263.csv (2/25-3/1)	test-3-299.csv (10/25-10/28)
test-3-264.csv (3/1-3/4)	test-3-300.csv (10/28-10/29)
test-3-265.csv (3/4-3/8)	test-3-301.csv (11/1-11/7)
test-3-266.csv (3/8-3/11)	test-3-302.csv (11/7-11/12)
test-3-267.csv (3/11-3/13) *	test-3-303.csv (11/12-11/18)

Cont on p. 159



Temperature & RH Readings  
(cont from p 139)

MA

Date/Time	RH	Temp	RH	Temp
8/27 0700	17.23	8.99	82.7%	42.4°C
8/28 0920	17.37	8.96	83.6%	42.0°C
8/29 0955	17.45	8.97	84.1%	42.1°C
8/30 1530	17.57	9.04	84.8%	43.0°C
8/31 0910	17.64	9.08	85.2%	43.5°C
9/1 0920	18.05	9.07	87.8%	43.4°C
9/5 0740	18.48	9.16	90.5%	44.5°C
9/6 0915	18.04	9.17	87.7%	44.6°C
9/7 0850	18.22	9.17	88.9%	44.6°C
9/10 0850	16.58	9.26	78.6%	45.7°C
9/11 0920	17.67	9.23	85.4%	45.4°C
9/12 0850	19.05	9.17	93.7%	44.6°C
9/13 0910	19.08	9.14	94.2%	44.2°C
9/14 0810	18.97	9.16	93.6%	44.5°C
9/17 0805	19.04	9.10	94.0%	43.7°C
9/18 0905	20.00	9.11	100.0%	43.9°C
9/19 0700	20.00	9.10	100.0%	43.7°C
9/20 0925	20.00	9.12	100.0%	44.0°C
9/21 0820	20.00	9.13	100.0%	44.1°C
9/24 0755	17.85	9.08	86.6%	43.5°C
9/25 0905	17.55	9.09	84.7%	43.6°C
9/26 0800	17.43	9.11	83.9%	43.9°C
9/27 0820	17.03	9.11	81.4%	43.9°C
9/28 0840	17.56	9.12	84.7%	44.0°C
10/1 0735	17.47	9.10	84.2%	43.7°C
10/2 0945	17.54	9.09	84.6%	43.6°C
10/3 0835	17.66	9.10	85.4%	43.7°C
10/4 0930	17.42	9.09	83.9%	43.6°C
10/5 0810	17.63	9.10	85.2%	43.7°C
10/8 0750	17.20	9.09	82.5%	43.6°C
10/9 0915	17.57	9.03	84.8%	42.9°C
10/10 0745	18.32	9.00	89.5%	42.5°C

MA

Date/Time	RH	Temp	RH	Temp
10/11 0925	18.62	8.93	91.4%	41.6°C
10/12 0920	19.05	8.99	94.1%	42.4°C
10/15 0800	18.36	8.95	89.7%	41.9°C
10/16 0730	18.65	10.66	91.6%	63.2°C
10/17 0815	18.86	10.62	92.9%	62.7°C
10/18 0845	18.90	10.63	93.1%	62.9°C
10/19 0800	18.94	10.66	93.4%	63.2°C
10/22 0940	18.75	10.91	92.2%	66.4°C
10/23 0800	18.63	10.93	91.4%	66.6°C
10/24 0805	19.50	10.67	96.9%	63.4°C
10/25 1030	18.87	10.69	92.4%	63.6°C
10/26 0840	20.02	10.70	100.0%	63.7°C
10/29 0730	18.93	10.59	93.3%	62.4°C
10/30 0930	19.05	10.61	94.1%	62.6°C
10/31 0800	19.00	10.61	93.7%	62.6°C
11/1 0800	19.00	10.65	93.7%	63.1°C
11/2 0825	19.02	10.69	93.9%	63.6°C
11/5 0815	18.36	10.80	89.7%	65.0°C
11/6 0740	18.19	10.71	94.9%	63.9°C
11/7 0750	19.53	10.73	97.1%	64.1°C
11/8 0755	19.43	10.74	96.4%	64.2°C
11/9 0800	19.41	10.75	96.3%	64.4°C
11/12 0800	18.95	10.83	93.4%	65.4°C
11/13 0920	19.20	10.64	95.0%	63.0°C
11/14 0845	19.15	10.65	94.7%	63.1°C
11/15 0810	19.12	10.66	94.5%	63.2°C
11/16 0720	19.07	10.67	94.2%	63.4°C
11/19 0855	18.90	10.64	93.1%	63.0°C
11/20 1605	18.73	10.60	92.1%	62.5°C
11/21 1005	18.71	10.58	91.9%	62.2°C
11/26 0935	17.09	10.69	81.8%	63.6°C
11/27 0740	19.46	10.61	96.6%	63.1°C
11/28 0955	19.44	10.71	96.6%	63.9°C
11/29 0840	19.62	10.77	97.6%	64.6°C
11/30 0930	19.98	10.72	99.9%	64.0°C

cont on next page

		mA			
Date/Time	RH	Temp	RH	Temp	
12/3 0920	20.02	10.63	100%	62.9°C	
12/4 0730	20.00	10.71	100%	63.9°C	
12/5 0930	20.00	10.73	100%	64.1°C	
12/4 0725	20.00	10.60	100%	62.5°C	
12/7 1000	20.00	10.68	100%	63.5°C	
12/10 0920	20.00	10.86	100%	65.7°C	
12/11 0920	20.00	10.84	100%	65.5°C	
12/12 0750	20.00	10.88	100%	66.0°C	
12/13 0930	20.00	10.87	100%	65.9°C	
12/14 0930	20.00	10.90	100%	66.2°C	
12/17 0800	20.00	10.86	100%	65.7°C	
12/18 0745	20.00	10.90	100%	66.2°C	
12/19 0800	20.00	10.93	100%	66.6°C	
12/20 0810	20.00	10.91	100%	66.4°C	
12/21 0815	20.00	10.97	100%	67.1°C	
12/27 0805	17.86	11.10	86.6%	68.7°C	
12/28 0800	19.19	11.09	94.9%	68.6°C	
12/31 0820	18.86	11.08	92.8%	68.5°C	
1/2 0820	19.11	11.06	94.4%	68.3°C	
1/3 0825	19.73	11.10	98.3%	68.7°C	
1/4 0825	19.49	11.12	96.8%	68.9°C	
1/7 0930	19.22	11.11	95.1%	68.8°C	
1/8 0805	18.99	11.16	93.7%	69.5°C	
1/9 0925	18.04	11.31	87.7%	71.4°C	
1/10 0755	18.26	11.25	89.1%	70.0°C	
1/11 0935	18.39	11.26	89.9%	70.7°C	
1/14 0745	17.74	11.30	85.9%	71.2°C	
1/15 0805	17.91	11.28	86.9%	71.0°C	
1/16 0920	19.55	11.05	97.2%	68.1°C	
1/17 0930	19.71	11.09	98.2%	68.6°C	
1/18 0935	19.71	11.06	98.2%	68.2°C	
1/21 0755	18.27	11.24	89.2%	70.5°C	
1/23 0935	19.24	11.11	95.2%	68.9°C	
1/24 0930	19.36	11.11	96.0%	68.9°C	
1/28 0740	17.74	11.27	86.0%	70.9°C	

		mA			
Date/Time	RH	Temp	RH	Temp	
1/29 0745	19.41	11.09	96.3%	68.6°C	
1/30 0935	18.72	11.10	92.0%	68.7°C	
1/31 0930	19.53	10.91	98.1%	66.4°C	
2/1 0930	18.33	11.06	89.6%	68.3°C	
2/4 0720	17.62	11.06	85.1%	68.3°C	
2/5 0740	19.08	10.86	94.2%	65.7°C	
2/6 0930	19.20	10.92	95.0%	66.4°C	
2/7 0925	18.26	11.04	89.1%	68.1°C	
2/8 0925	18.10	11.05	88.1%	68.2°C	
2/11 0730	17.69	11.02	85.5%	67.7°C	
2/12 0755	18.06	11.05	87.9%	68.2°C	
2/13 0935	17.80	11.05	84.2%	68.2°C	
2/14 0925	18.49	10.84	90.5%	65.5°C	
2/15 0935	17.78	10.89	86.1%	66.1°C	
2/18 0750	17.26	10.91	81.6%	66.4°C	
2/19 0755	17.06	10.97	81.6%	67.1°C	
2/20 0930	17.24	10.98	82.7%	67.2°C	
2/21 0800	17.24	11.04	82.7%	68.1°C	
2/22 0940	18.23	11.05	88.9%	68.2°C	
2/25 0715	18.02	11.08	87.6%	68.5°C	
2/26 1315	18.28	11.00	89.2%	67.5°C	
2/27 0930	18.11	11.08	88.2%	68.5°C	
2/28 0925	18.20	11.06	88.7%	68.3°C	
3/1 0925	18.28	11.06	89.2%	68.3°C	
3/4 0735	17.71	11.06	85.7%	68.3°C	
3/5 0810	18.24	10.95	89.0%	66.9°C	
3/6 0930	18.19	10.97	88.7%	67.1°C	
3/8 0915	18.21	10.99	88.8%	67.3°C	
3/11 1345	17.80	11.01	86.2%	67.6°C	
3/12 0750	18.20	10.99	88.7%	67.3°C	
3/13 0755	17.98	11.02	87.4%	67.7°C	
3/14 0740	18.01	11.00	87.4%	67.5°C	
3/15 0815	18.06	11.01	87.9%	67.6°C	
3/18 0745	17.96	10.89	87.2%	66.1°C	
3/19 0725	18.00	11.03	87.5%	67.8°C	
3/20 0915	18.08	10.96	88.0%	66.2°C	

Cont on p. 149

1/8/02 JP 0925hr.

Collected water samples for chemical analysis; pH was determined using procedure on p 59

Sample Cup No.	Volume	Temp °C	pH
51H	60 ml	28.0°C	7.29
52H	60 ml	27.7°C	6.97
53H	60 ml	26.9°C	7.96
54H	50 ml	38.5°C	7.73-7.58 JP 1/8/02
55H	50 ml	31.4°C	7.58
512H	—	—	—
513H	20 ml	45.0°C	7.96
514H	20 ml	44.4°C	8.01
515H	10 ml	52.5°C	8.12

1/8/02  
1030hr  
JPAlkalinity of samples were determined using procedure on p 61.  
Results

===== METTLER DL21 Titrator date 08-01-02 =====

TITRATION	2	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.540	51H
pH0 pH	7.067	
SLOPE mV/pH	-57.3	40 ml Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.04920	
P-VALUE mmol/L	.12300	
RESULT mL	.66645	
M-VALUE mmol/L	1.6661	

Alkalinity =

$$\frac{(.04920 + .66645) * .1 * 50000}{40} =$$

89.46 mg CaCO<sub>3</sub>/L

===== METTLER DL21 Titrator date 08-01-02 =====

TITRATION	3	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	7.341	52H
pH0 pH	7.067	
SLOPE mV/pH	-57.3	40 ml Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.75827	
M-VALUE mmol/L	1.8957	

$$\rightarrow \text{Alkalinity} = \frac{.75827 * .1 * 50000}{40}$$

94.78 mg CaCO<sub>3</sub>/L

===== METTLER DL21 Titrator date 08-01-02 =====

TITRATION	6	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	8.414	
pH0 pH	7.067	54H
SLOPE mV/pH	-57.3	40 ml Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.04554	
P-VALUE mmol/L	.11386	
RESULT mL	.67585	
M-VALUE mmol/L	1.6896	

$$\text{Alkalinity} = \frac{(.04554 + .67585) * .1 * 50000}{40}$$

90.17 mg CaCO<sub>3</sub>/L

$$\rightarrow \text{Alkalinity} = \frac{.66187 * .1 * 50000}{40}$$

82.73 mg CaCO<sub>3</sub>/L

===== METTLER DL21 Titrator date 08-01-02 =====

TITRATION	4	operator
METHOD	4011	2 04-12 0
WEIGHT g	1	
IDENT	0	remarks
conc mol/L	.10000	
const	25.000	
START pH	7.369	53H
pH0 pH	7.067	
SLOPE mV/pH	-57.3	40 ml Sample Volume
M-POINT pH	4.300	
P-POINT pH	8.200	
RESULT mL	.66187	
M-VALUE mmol/L	1.6547	

===== METTLER DL21 Titrator date 08-01-02

TITRATION 7 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 7.315 55H

pH0 pH 7.067 40ml

SLOPE mV/pH -57.3 Sample

M-POINT pH 4.300 Volume

P-POINT pH 8.200

RESULT mL .49726

M-VALUE mmol/L

1.2431

$$\text{Alkality} = \frac{.49726 \times .1 \times 50000}{40} =$$

62.16 mg CaCO<sub>3</sub>/L

11/8/02

1200h gpp

After pH & alkalinity measurement samples were split into 2 aliquots & placed in 30 ml PP bottles. One aliquot was acidified by adding 10 µl HNO<sub>3</sub> and labeled as follows.

S1H-A, S2H-A, S3H-A, S4H-A, S5H-A, S13H-A, S14H-A, & S15H-A

The acidified samples retained their original labels.  
Samples were refrigerated for later analysis.

Temperature & RH readings (cont from p145)

Date/Time mT

		RH	Temp	RH	Temp
3/21	0935	18.16	10.84	88.5%	65.5°C
3/22	0940	18.22	10.79	88.9%	64.9°C
3/25	0725	17.87	11.06	86.7%	68.2°C
3/26	0745	17.97	11.09	87.3%	68.6°C
3/27	0920	18.05	11.10	87.8%	68.7°C
3/28	0925	17.94	11.12	87.1%	68.9°C
3/29	0845	18.30	11.04	89.4%	68.0°C
4/1	0725	17.59	11.02	84.9%	67.7°C
4/2	0735	17.97	11.05	87.3%	68.1°C
4/3	0925	18.14	11.03	88.4%	67.9°C
4/4	0915	18.31	11.00	89.5%	67.5°C
4/5	0935	18.30	10.98	89.4%	67.2°C
4/8	0755	17.80	11.00	86.2%	67.5°C
4/9	0740	18.26	11.00	89.1%	67.5°C
4/10	0905	18.17	11.01	88.6%	67.6°C
4/11	0925	18.24	11.00	89.0%	67.5°C
4/12	0925	18.18	11.01	88.6%	67.6°C
4/15	0740	18.06	11.00	87.9%	67.5°C
4/16	0750	13.67	11.65	60.4%	75.6°C
4/17	1000	19.22	10.88	95.1%	66.0°C
4/18	0925	20.00	11.00	100.0%	67.5°C
4/19	0920	20.00	11.03	100.0%	67.9°C
4/22	0735	19.32	11.12	95.7%	69.0°C
4/23	0750	20.02	11.09	100%	68.6°C
4/24	0930	20.00	10.97	100%	67.1°C
4/26	0730	20.00	10.91	100%	66.4°C
4/29	0720	20.00	10.90	100%	66.2°C
4/30	0800	19.58	10.91	97.4%	66.4°C
5/1	0925	18.95	10.94	93.4%	66.7°C
5/2	0940	18.97	10.95	93.6%	66.8°C
5/3	0925	19.40	10.90	96.2%	66.2°C
5/6	0930	18.89	10.90	95.1% <sup>2</sup> 93.1%	66.2°C
5/7	0740	19.03	10.92	93.9%	66.5°C
5/8	0800	18.73	11.00	92.1%	67.5°C

Cont on page 150

Date/Time	mA		RH	Temp
	RH	Temp		
5/9 0925	18.08	11.05	58.0%	68.1°C
5/10 0925	18.66	11.03	91.6%	67.9°C
5/13 0735	17.68	11.12	89.5%	69.0°C
5/14 0725	17.15	11.14	82.2%	69.2°C
5/15 0740	19.98	10.98	99.9%	67.2°C
5/16 0940	19.43	10.98	96.4%	67.2°C
5/17 1800	19.11	11.06	94.4%	68.2°C
5/20 0945	19.35	11.11	95.9%	68.9°C
5/21 1325	19.58	11.18	97.4%	69.7°C
5/23 1310	19.47	11.08	96.7%	68.5°C
5/28 0805	18.41	11.09	90.1%	68.6°C
5/29 0755	19.47	11.13	96.7%	69.1°C
5/30 0745	19.78	11.11	98.6%	68.9°C
6/3 0745	14.97	11.54	66.1%	74.2°C
6/4 0750	16.90	11.33	80.6%	71.6°C
6/5 0805	16.92	11.33	80.7%	71.6°C
6/6 0750	17.54	11.22	84.6%	70.2°C
6/7 0815	16.99	11.27	81.2%	70.9°C
6/10 0725	17.02	11.30	81.4%	71.2°C
6/11 0920	17.52	11.19	84.5%	69.9°C
6/12 0730	17.44	11.21	84.0%	70.1°C
6/13 0735	17.55	11.23	84.7%	70.3°C
6/14 0745	17.24	11.21	85.2%	70.1°C
6/17 0725	15.80	11.36	73.7%	72.0°C
6/18 0755	16.01	11.32	75.0%	71.5°C
6/19 0705	16.08	11.45	75.5%	73.1°C
6/20 0725	16.04	11.35	75.1%	71.9°C
6/21 0735	16.70	11.23	79.4%	70.4°C
6/24 0725	16.44	11.38	77.7%	72.2°C
6/25 0720	17.14	11.32	82.1%	71.5°C
6/26 0720	17.40	11.29	83.7%	71.1°C
6/27 0715	17.42	11.30	83.9%	71.2°C
6/28 0730	17.48	11.31	84.2%	71.3°C
7/1 0725	16.52	11.38	78.2%	72.2°C

Date/Time	mA		RH	Temp
	RH	Temp		
7/2 0730	16.69	11.40	79.3%	72.5°C
7/3 0740	17.53	11.32	84.5%	71.5°C
7/5 0750	17.42	11.27	83.9%	70.9°C
7/8 0735	17.00	11.24	81.2%	70.5°C
7/9 0730	17.40	11.25	83.7%	70.6°C
7/10 0730	17.35	11.26	83.4%	70.8°C
7/11 0730	17.26	11.25	82.9%	70.6°C
7/12 0735	17.30	11.28	83.1%	80.0°C

\* Switched Vaisala Temp/Humidity probe S/N 81240028  
with S/N 61840037.

7/12 0800	19.64	8.78		
7/15 0815	17.19	9.27	82.4%	45.8°C
7/16 0825	17.38	9.27	83.6%	45.8°C
7/17 0755	17.38	9.20	83.6%	45.0°C
7/18 0735	16.73	9.26	79.6%	45.7°C
7/19 0755	15.32	9.35	70.7%	46.7°C
7/22 0810	16.38	9.29	77.4%	46.0°C
7/23 0725	16.67	9.26	79.2%	45.7°C
7/24 0730	16.59	9.31	80.6%	46.4°C
7/25 0745	16.59	9.34	80.6%	46.7°C 46.7°C
7/26 0740	16.62	9.33	78.9%	46.6°C
7/29 0740	16.26	9.34	76.6%	44.7°C 46.7°C
7/30 0725	16.48	9.32	78.0%	46.5°C
7/31 0700	16.54	9.31	78.4%	46.4°C
8/1 0830	16.80	9.24	80.0%	45.5°C
8/2 0745	16.68	9.24	79.2%	45.5°C
8/5 0725	16.39	9.23	77.4%	45.4°C
8/6 0725	16.82	9.21	80.1%	45.1°C
8/7 0725	11.65	9.71	47.8%	51.4°C
8/8 0745	16.54	9.33	78.4%	46.6°C
8/12 0755	15.88	9.34	74.2%	46.7°C
8/13 0750	16.75	9.32	79.7%	46.5°C
8/14 0800	16.60	9.31	78.7%	46.4°C

Cont on next page



Date/Time	MA	RH	Temp	RH	Temp
8/15 0745		16.46	9.31	77.9%	46.4°C
8/16 0750		16.69	9.30	79.3%	46.3°C
8/19 0755		16.55	9.32	78.4%	46.5°C
8/20 0750		17.18	9.30	82.4%	46.3°C
8/21 0800		17.29	9.29	83.1%	46.2°C
8/22 0805		17.17	9.28	82.3%	46.1°C
8/23 0805		17.48	9.26	84.2%	45.9°C
8/26 0720		16.93	9.30	80.8%	46.3°C
8/27 0820		16.49	9.32	78.1%	46.5°C
8/28 0815		16.12	9.28	75.7%	46.1°C
8/29 0810		16.36	9.23	77.2%	45.4°C
9/5 0940		16.72	9.24	79.5%	45.5°C
9/6 0910		16.49	9.26	78.1%	45.9°C
9/9 0822		17.35	9.22	83.4%	45.2°C
9/10 0905		17.16	9.22	82.2%	45.2°C
9/11 0907		17.14	9.23	82.1%	45.4°C
9/12 0905		16.92	9.25	80.7%	45.0°C
9/13 0925		17.01	9.23	81.3%	45.4°C
9/14 0936		16.76	9.23	79.7%	45.4°C
9/17 1030		17.11	9.24	81.9%	45.5°C
9/18 0905		16.75	9.26	79.7%	45.7°C
9/19 0925		16.84	9.27	80.2%	45.8°C
9/20 0905		16.70	9.27	79.4%	45.8°C
9/23 0725		16.82	9.20	80.1%	45.0°C
9/24 0905		16.77	9.22	79.8%	45.2°C
9/25 0910		16.72	9.21	79.5%	45.1°C
9/26 0905		16.70	9.22	79.4%	45.2°C
9/27 0920		16.78	9.21	79.9%	45.1°C
9/30 0920		16.82	9.22	80.1%	45.2°C
10/1 0920		16.83	9.25	80.2%	45.6°C
10/4 0920		16.61	9.29	78.8%	46.1°C
10/7 0925		16.71	9.23	79.4%	45.4°C
10/8 0940		16.76	9.25	79.7%	45.6°C
10/9 0925		16.81	9.28	80.1%	46.0°C
10/10 0905		16.86	9.27	80.4%	45.9°C
10/11 0915		16.91	9.25	80.7%	45.6°C

Date/Time	MA	RH	Temp	RH	Temp
10/14 0740		16.70	9.19	79.4%	44.9°C
10/15 0740		16.66	9.20	79.1%	45.0°C
10/16 0915		16.65	9.22	79.1%	45.2°C
10/17 0900		16.54	9.24	78.4%	45.5°C
10/18 0915		16.46	9.27	77.9%	45.8°C
10/21 0910		16.80	9.24	80.0%	45.5°C
10/22 0830		16.68	9.26	79.2%	45.7°C
10/23 0915		16.69	9.28	79.3%	46.0°C
10/24 1620		16.58	9.29	78.6%	46.1°C
10/25 0915		16.73	9.29	79.6%	46.1°C
10/28 0915		16.48	9.31	78.0%	46.4°C
10/29 0745		16.50	9.31	78.1%	46.4°C
10/30 0915		16.59	9.28	78.7%	46.0°C
10/31 1210		16.47	9.28	77.9%	46.0°C
11/1 0935		17.09	9.17	81.8%	44.6°C
11/4 0930		16.95	9.12	80.9%	44.0°C
11/5 0955		16.57	9.19	79.4%	44.9°C
11/6 0915		16.62	9.20	78.9%	45.0°C
11/7 1315		16.34	9.24	77.1%	45.5°C
11/8 1020		16.24	9.26	76.5%	45.7°C
11/12 0745		15.61	9.32	72.6%	44.5°C
11/13 1005		15.40	9.30	71.2%	46.3°C
11/14 0905		12.84	9.49	55.2%	48.6°C
11/15 1000		12.90	9.50	55.6%	48.7°C
11/18 0905		14.19	9.36	76.2%	46.3°C
11/19 0740		16.25	9.31	76.6%	46.4°C
11/20 0916		16.53	9.30	78.3%	46.3°C
11/21 0900		16.54	9.28	78.4%	46.0°C
11/22 0910		14.27	9.32	76.7%	46.5°C
12/2 0930		15.00	9.47	68.7%	48.4°C
12/3 0805		15.39	9.46	71.2%	48.3°C
12/4 0910		15.94	9.34	74.6%	46.7°C
12/5 0915		15.99	9.31	74.9%	46.4°C
12/6 0915		15.54	9.35	72.1%	46.8°C
12/9 0915		13.91	9.47	61.9%	48.4°C

(cont on p2 of scientific notebook 559).

11/22/02 Jp 1335hr

Collected water samples for chemical analysis, pH  
was determined using procedure on p 59

Sample cup label	Amount	Temp °C	pH
S1I	80ml	29.2°C	8.41
S2I	80ml	30.38°C	8.34
S3I	70ml	26.7°C	8.38
S4I	10ml	40.52°C	7.93
S5I	30ml	39.5°C	8.40
S12I	—	—	—
S13I	—	—	—
S14I	20ml	43.3°C	8.51
S15I	—	—	—

11/22/02 Jp 1505hr

Alkalinity of samples S1I, S2I, & S3I  
was determined using procedure on p 61.  
Results are shown on Solley page.

\*\*\*\*\* METTLER DL21 Titrator date 22-11-02 \*\*\*\*\*

TITRATION 1 operator

METHOD 4011 2 04-12-0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 9.317

PHO pH 7.060

SLOPE mV/pH -59.1

M-POINT pH 4.200

P-POINT pH 3.200

RESULT mL .03381

P-VALUE mmol/L

.08453

RESULT mL .04710

M-VALUE mmol/L

1.6178

S1I

40ml  
sample

$$\text{Alkalinity} = \frac{(0.03381 + 0.04710) \times 1 \times 50000}{40} = 85.11 \text{ mg CaCO}_3/\text{L}$$

\*\*\*\*\* METTLER DL21 Titrator date 22-11-02 \*\*

TITRATION 4 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 7.961

pH0 pH 7.060

SLOPE mV/pH -57.1

M-POINT pH 4.300

P-POINT pH 9.200

RESULT mL .66217

M-VALUE mmol/L

1.5054

$$\text{Alkalinity} = \frac{.60217 * .1 * 50000}{40} = 75.27 \text{ mg CaCO}_3/\text{L}$$

\*\*\*\*\* METTLER DL21 Titrator date 22-11-02 \*\*

TITRATION 5 operator

METHOD 4011 2 04-12 0

WEIGHT g 1

IDENT 0 remarks

conc mol/L .10000

const 25.000

START pH 7.967

pH0 pH 7.060

SLOPE mV/pH -57.1

M-POINT pH 4.300

P-POINT pH 9.200

RESULT mL .73146

M-VALUE mmol/L

1.8286

$$\text{Alkalinity} = \frac{.73146 * .1 * 50000}{40} = 91.43 \text{ mg CaCO}_3/\text{L}$$

11/22/02 JJP 1620hr

After pH & alkalinity measurements samples were split into two aliquots & placed in 30 ml PP bottles. One aliquot was acidified by adding 10  $\mu$ l conc  $HNO_3$  & labeled as follows.

S1I-A

S2I-A

S3I-A

S4I-A

S5I-A

S14I-A

The unacidified aliquots retained their original labels.

Samples were refrigerated for later analysis.

12/2/02 JJP

Cont from p 141.

Temperature data is stored in following files.

test-3-304.csv (11/22-12/02)

test-3-305.csv (12/2-12/6)

test-3-306.csv (12/6-12/9)

test-3-307.csv (12/20-12/30)

test-3-308.csv (12/30-1/2)

test-3-309.csv (1/2-1/6)

test-3-310.csv (1/6-1/10)

test-3-311.csv (1/10-1/14)

test-3-312.csv (1/14-1/20)

test-3-313.csv (1/20-1/24)

test-3-314.csv (1/24-1/27)

12/10/02 JP

Entries for the lab scale heater  
test are contained in scientific  
notebook 559.

I have reviewed this scientific notebook and find it in agreement with QAP-001.  
There is sufficient information regarding methods used for conducting tests,  
acquiring and analyzing data so that another qualified individual could repeat  
the activity.

R.C. Pearson  
3/13/2003