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Look at impact of Cs on offshore consequences

Run Base Case:

atmos 76.inp' }  
 early 299.inp' }  
 chmcl-h.inp' }  
 SURSIT.INP' }  
 METSUR.INP' }  
 BEST B. OUT'

Remove only Cs-134      atmos 134.inp'      → 134.out'  
 Remove only Cs-136      atmos 136.inp'      → 136.out'  
 Remove only Cs-137      atmos 137.inp'      → 137.out'  
 Remove all 3 Cs's      atmos all.inp'      → all.out'

Remove all other isotopes  
 besides 3 Cs's      atmos off.inp'      → off.out'

Cs inventory in above runs corresponds to 3 cores.  
 How much consequences with 1.5 cores?  
 How much consequences with 6 cores?

<u>Isotope</u>	<u>Half-life</u>	<u>1.5 cores</u>	<u>6 cores</u>
Cs-134	2.1 yr	same as 3 cores	same as 3 cores
Cs-136	13 days	same as 3 cores	same as 3 cores
Cs-137	30 yrs	x .5	x 2

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1.5 cores

Take atmos. inp.

Reduce the Cs-137 inventory by factor of 2.

$$\text{Cs-137} \quad 8.58 \times 10^{17} \times .5 = 4.29 \times 10^{17} \text{ Bq}$$

atmos lp5. inp  $\rightarrow$  lp5. out

6.0 cores

Take atmos. inp.

Increase Cs-137 inventory by factor of 2.

$$\text{Cs-137} \quad 8.58 \times 10^{17} \times 2 = 1.72 \times 10^{18} \text{ Bq}$$

atmos 6pφ. inp  $\rightarrow$  6pφ. out

Why no Cs-135 in MACCS? Ans:  $t_{1/2} = 3 \times 10^6$  yrs

$$Xe-135 \quad 53,630 \text{ Ci} \quad \lambda = 2.09 \times 10^{-5} / \text{sec} \quad t_{1/2} = 9.2 \text{ h}$$

$$Cs-135 \quad .03 \text{ Ci} \quad \lambda = 7.29 \times 10^{-15} / \text{sec} \quad t_{1/2} = 3 \times 10^6 \text{ y}$$

$$\lambda N = \text{Ci}$$

$$\lambda_{Xe-135} N_{Xe-135} = \text{Activity} \\ (\text{Ci})$$

$$53,630 \text{ Ci} = 2.09 \times 10^{-5} \frac{\text{}}{\text{sec}} \cdot N$$

$$Xe-135 \quad N = \frac{53,630 \text{ Ci}}{2.09 \times 10^{-5} / \text{sec}}$$

$$Cs-135 \quad N = \frac{53,630 \text{ Ci}}{2.09 \times 10^{-5} / \text{sec}}$$

$$Cs-135 \text{ activity} = \lambda_{Cs-135} \cdot N = 7.29 \times 10^{-15} \frac{\text{}}{\text{sec}} \cdot \frac{53,630 \text{ Ci}}{2.09 \times 10^{-5} \frac{\text{}}{\text{sec}}}$$

$$Cs-135 \text{ activity} = .00002 \text{ Ci}$$